

1 **TITLE**

2 **Ni-phylosilicates (garnierites) from the Falcondo Ni-laterite deposit (Dominican Republic):**
3 **mineralogy, nanotextures and formation mechanisms by HRTEM and AEM**

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5 **AUTHORS AND AFFILIATIONS**

6 Cristina Villanova-de-Benavent – cvillanovadb@ub.edu

7 *Departament de Cristal·lografia, Mineralogia i Dipòsits Minerals, Facultat de Geologia,*
8 *Universitat de Barcelona (UB), Martí i Franquès s/n, 08028 Barcelona, Spain. +34 934021341*

9 Fernando Nieto – nieto@gr.es

10 *Departamento de Mineralogía y Petrología and IACT, Universidad de Granada, CSIC, Av.*
11 *Fuentenueva 18071 Granada, Spain. +34 609132940*

12 Cecilia Viti – cecilia.viti@unisi.it

13 *Dipartimento di Scienze Fisiche, della Terra e dell'Ambiente, Università degli Studi di Siena, Via*
14 *Laterina 8, 53100 Siena, Italy. +39 0577233988*

15 Joaquín A. Proenza – japroenza@ub.edu

16 *Departament de Cristal·lografia, Mineralogia i Dipòsits Minerals, Facultat de Geologia,*
17 *Universitat de Barcelona (UB), Martí i Franquès s/n, 08028 Barcelona, Spain. +34 934021351*

18 Salvador Galí – gali@ub.edu

19 *Departament de Cristal·lografia, Mineralogia i Dipòsits Minerals, Facultat de Geologia,*
20 *Universitat de Barcelona (UB), Martí i Franquès s/n, 08028 Barcelona, Spain. +34 934021341*

21 Josep Roqué-Rosell – jrrosell@lbl.gov

22 *Advanced Light Source, Lawrence Berkeley National Laboratory, One Cyclotron Road, MS*
23 *15R0317 Berkeley, California 94720, USA. +1 510 486 7035*

24

25 **ABSTRACT**

26 Ni-bearing magnesium phyllosilicates (garnierites) are significant Ni ores in Ni-laterites worldwide.
27 The present paper reports a detailed TEM investigation of garnierites from the Falcondo Ni-laterite
28 deposit (Dominican Republic). Different types of garnierites have been recognized, usually
29 consisting of mixtures between serpentine and talc-like phases which display a wide range of
30 textures at the nanometer scale. In particular, chrysotile tubes, polygonal serpentine and lizardite
31 lamellae are intergrown with less crystalline, talc-like lamellae. Samples consisting uniquely of talc-
32 like and of sepiolite-falcondoite were also observed, occurring as distinctive thin lamellae and long
33 ribbon-shaped fibers, respectively. HRTEM imaging indicates that serpentine is replaced by the talc-
34 like phase, whereas TEM-AEM data show preferential concentration of Ni in the talc-like phase. We
35 suggest, therefore, that the crystallization of Ni-bearing phyllosilicates is associated with an increase
36 in the silica activity of the system, promoting the replacement of the Ni-poor serpentine by the Ni-
37 enriched talc-like phase. These results have interesting implications in material science, as
38 garnierites are natural analogues of Ni-bearing phyllosilicate-supported synthetic catalysts. Finally,
39 SAED and HRTEM suggest that the Ni-bearing talc-like phase corresponds to a variety of talc with
40 extra water, showing larger d_{001} than talc (i.e. 9.2–9.7 Å), described as “kerolite”-“pimelite” in clay
41 mineral literature.

42

43 **KEYWORDS**

44 Ni-laterites; garnierites; chrysotile; polygonal serpentine; lizardite; “kerolite”-“pimelite”; sepiolite-
45 falcondoite; HRTEM

46 **MANUSCRIPT**

47

INTRODUCTION

48 *Garnierite* was originally the name of a mineral species discovered in 1863 in New Caledonia
49 (Garnier 1867), although it was later proved to be a mixture of different Ni-hydrous silicates (e.g.
50 Pecora et al. 1949, Faust 1966). Since then, the term garnierite has been widely used in the literature
51 to refer to the group of green, fine-grained, poorly crystallized, Ni-bearing magnesium
52 phyllosilicates, including serpentine, talc, sepiolite, smectite and chlorite, often occurring as
53 mixtures (e.g. Faust 1966, Brindley and Hang 1973, Springer 1974, Brindley 1978, Wells et al.
54 2009, Villanova-de-Benavent et al. 2014a). Therefore, despite not being a valid mineral name,
55 *garnierite* became a convenient field term used by mine geologists to designate all green Ni-
56 phyllosilicates when a more specific characterization was not possible (Brindley 1978), and many
57 authors have used this term with this general meaning (e.g. Moraes, 1935, Pecora and Hobbs 1942,
58 Pecora et al. 1949, Varela 1984, Gleeson et al. 2003, 2004, Freyssinet et al. 2005, Wells et al. 2009).
59 In addition, the classification and naming of the garnierite minerals represents a complex, long-
60 lasting controversy, because of their fine-grained nature, poor crystallinity and frequent occurrence
61 as intimate mixtures of different mineral species (Brindley and Hang 1973). Brindley and co-
62 authors, after various studies, distinguished the following Mg-Ni series: the serpentine group
63 minerals i) lizardite-népouite and ii) chrysotile-pecoraite; iii) berthierine-brindleyite; the talc-like
64 structures iv) talc-willemseite and v) “kerolite”-“pimelite”; vi) the chlorite series clinochlore-nimite;
65 and finally vii) sepiolite-falcondoite (Brindley and Hang 1973, Brindley and Maksimović 1974,
66 Brindley 1978, 1980). The most common garnierites found in nature are formed by lizardite-
67 népouite and “kerolite”-“pimelite” (Brindley 1978), and many authors have been referred to them as
68 “serpentine-like” (or “7 Å-type”) and “talc-like” (or “10 Å-type”) garnierites, respectively (e.g.
69 Brindley and Hang 1973, Brindley and Maksimović 1974, Wells et al. 2009, Galí et al. 2012).

70 “Kerolite” and “pimelite” can be described as phases with talc affinity and extra water in their
71 structure, within the ideal structural formula $(\text{Mg,Ni})_3\text{Si}_4\text{O}_{10}(\text{OH})_2 \cdot n(\text{H}_2\text{O})$. Faust (1966) classified
72 “pimelite” into the smectite group, although other authors proved that neither “kerolite” nor
73 “pimelite” exhibit intracrystalline swelling (e.g. Kato 1961, Brindley and Hang 1973, Brindley
74 1978). Regardless being both discredited species by the Commission on New Minerals,
75 Nomenclature and Classification of the International Mineralogical Association (CNMNC-IMA),
76 these names have been used during the following decades (e.g. Gleeson et al. 2003, Freyssinet et al.
77 2005, Tauler et al. 2009, Wells et al. 2009, Galí et al. 2012, Villanova-de-Benavent et al. 2014a,
78 Cathelineau et al. 2015) and are considered as valid species by the International Association for the
79 Study of Clays (AIPEA). According to Dosbaba and Novák (2012), “kerolite” represents a fine
80 crystalline, poorly ordered, hydrated variant of talc, from which it can be distinguished by its very
81 broad peak at approximately 10 Å; from smectite it can be distinguished by the absolute lack of
82 swelling at an ethylene glycol atmosphere (Brindley and Brown 1980).

83 The remarkable Ni content of garnierites makes them significant Ni ores in some Ni-laterite
84 deposits, such as in the Falcondo deposit, in the Dominican Republic (e.g. Golightly 1981, Elias
85 2002, Gleeson et al. 2003, Freyssinet et al. 2005). In the Falcondo Ni-laterite, characterized by an
86 Al-poor ultramafic protolith, five garnierite types were distinguished according to their color,
87 mineralogy, textures and chemical composition (namely types I to V, Villanova-de-Benavent et al.
88 2014a, b): I) Ni-Fe-bearing serpentine-like, II) Ni-bearing mixture of serpentine-like and “kerolite”-
89 “pimelite”, III) Ni-dominant mixture of serpentine-like and “kerolite”-“pimelite”, IV) “kerolite”-
90 “pimelite”, and V) sepiolite-falcondoite.

91 These previous results, obtained by means of X-ray powder diffraction (XRD), optical and scanning
92 electron microscopy (SEM, with energy dispersive spectrometry, EDS) and electron microprobe
93 (EMP), indicated that, in general, the garnierites from Falcondo are complex mixtures at the

94 nanometer scale (Villanova-de-Benavent et al. 2014a, b). Although the methods used are sufficient
95 for a description of the phases from an economic point of view, a detailed study including textures at
96 the nanometer scale is necessary, to unravel the characteristics of the various mixtures and to gain
97 further insight into the distribution of Ni.

98 Transmission Electron Microscopy (TEM) is a powerful technique to study the mineralogy,
99 chemistry, and textural features of phases which commonly occur as poorly crystalline, fine grained
100 mixtures at the nanometer scale. It has been extensively used to study phyllosilicates and, in
101 particular, this method allows distinction of the different minerals of the serpentine group (e.g.
102 Baronnet and Devouard 2005, Andréani et al. 2008, Suárez et al. 2011, Mellini 2014).

103 Serpentine group minerals are 1:1 trioctahedral phyllosilicates that develop different structural forms
104 because of a geometrical misfit between tetrahedral and octahedral sheets. The three most common
105 varieties are i) lizardite planar lamellae, ii) chrysotile tubes and iii) antigorite modulated structures.

106 In addition, other microstructures may be found, such as polygonal serpentine, which develop
107 sectorized rolled fibers, larger than chrysotile tubes, with 15 or 30 sectors separated by 24° and 12°,
108 respectively (e.g. Baronnet and Devouard 2005, Andréani et al. 2008, Mellini 2014, and references
109 therein). Up-to-date studies report that sectors are actually composed of lizardite (e.g. Mellini 2014).

110 However, a limited number of exhaustive TEM works on Ni-bearing Mg-phyllosilicates exists up to
111 the present (Uyeda et al. 1973, Esson and Carlos 1978, Poncelet et al. 1979, Pelletier 1983, Soler et
112 al. 2008, Tauler et al. 2009, Suárez et al. 2011), and high resolution imaging and electron diffraction
113 studies are in general scarce. Furthermore, most of these publications are based on crushed and
114 dispersed material onto TEM grids, omitting the textural information of the assemblages and the
115 relationships between different phases.

116 This paper presents a textural and chemical characterization by TEM of the different garnierites
117 described in the Falcondo Ni-laterite deposit. The aim of this work was to describe the textural

118 relationships between the different garnierite-forming minerals and to determine the distribution of
119 Ni at the nanometer scale based on High Resolution TEM (HRTEM) and Analytical Electron
120 Microscopy (AEM).

121 **GEOLOGICAL SETTING AND GARNIERITE OCCURRENCE**

122 Hydrous silicate Ni-laterites of the Falcondo deposit are developed on the Loma Caribe, an
123 ophiolitic peridotite belt in the Cordillera Central in the Dominican Republic. This belt consists of an
124 elongated body, 4–5 km wide and 95 km long, oriented NW-SE and bounded by major faults (Fig.
125 1a, b). The Loma Caribe ultramafic rocks, formed in the upper mantle, are composed of
126 harzburgites, dunites and lherzolites, partially altered to serpentinites during their emplacement to
127 the present day tectonic position (Lewis et al. 2006, Proenza et al. 2007). Partially serpentinised
128 peridotites have been exposed to weathering and erosion since the early Miocene, when the
129 laterisation process began (Lewis et al. 2006).

130 The Falcondo Ni-laterite profile is up to 60 m thick and is divided into two main zones or horizons:
131 the limonite on the top and the saprolite at the bottom (Fig. 1c; Haldemann et al. 1979, Lithgow
132 1993, Lewis et al. 2006). However, the contacts between subzones and their thicknesses vary
133 vertically and laterally in all outcrops of the Falcondo deposit (Lewis et al. 2006, Aiglsperger et al.
134 2014, Villanova-de-Benavent et al. 2014a).

135 Garnierites in the Falcondo Ni-laterite occur mainly as mm-cm vein infillings in fractures, as thin
136 coatings on joints and along fault planes, and as clasts and/or cements in different kinds of breccias.
137 They are usually found within the lowermost part of the saprolite horizon, but may also be located
138 near the unweathered serpentinized peridotite rocks, at the base of the lateritic profile, and in the
139 upper saprolite horizon (Tauler et al., 2009, Villanova-de-Benavent et al. 2014a).

140 **MATERIALS AND METHODS**

141 Forty samples from the lower saprolite horizon containing different garnierite types were previously
142 identified by X-ray powder diffraction (XRD), observed by optical and scanning electron
143 microscopy (SEM-EDS), and analyzed by electron microprobe (EMP) at the Centres Científics i
144 Tecnològics of the Universitat de Barcelona (CCiT-UB) (Villanova-de-Benavent et al. 2014a). From
145 these forty samples, nine were selected as the most representative to be studied by transmission
146 electron microscopy for the present work (Table 1). Three different methods of specimen preparation
147 were combined, due to the difficulties posed by the friability of the material, and twelve grids were
148 prepared with similar outcomes.

149 First, some samples were prepared as polished thin sections with Canada balsam. Representative
150 areas containing garnierites to be studied by TEM were selected under the optical microscope.
151 Copper grids were attached on the thin section including the areas of interest, and they were later
152 detached, ion-thinned by a Gatan 600 ion mill and a Gatan PIPS 691 at the Centro de
153 Instrumentación Científica in the Universidad de Granada (CIC-UGR) (Suárez et al., 2011), and by a
154 Gatan 600 Duo Mill at the Dipartimento di Scienze Fisiche, della Terra e dell'Ambiente of the
155 Università degli Studi di Siena (UniSi). Second, small fragments of other selected samples were
156 carefully separated by hand-picking, polished manually up to ~30 micrometers thick and glued to a
157 copper grid. The grids were ion-thinned by a Fischione 1010 Low Angle Ion Milling & Polishing
158 System and a Gatan PIPS 691 at the CCiT-UB. It is worth noting that the aforementioned specimen
159 preparation methods preserve the original texture of the mineral phases. Third, other samples were
160 separated by hand-picking, ground in an agate mortar, suspended with ethanol and put on a copper
161 grid in order to obtain additional, higher quality EDX analyses (e.g. López Munguira and Nieto
162 2000, Abad et al. 2001). Both ion-milled samples and powders were carbon-coated prior to the TEM
163 study.

164 The TEM study was performed by Philips CM20 (CIC-UGR) and a Jeol JEM2010 (UniSi), equipped
165 with Energy Dispersive X-ray spectrometer detectors (EDS-Oxford Isis, respectively) and operating
166 at 200 kV. Additional images and electron diffraction patterns were obtained using a Jeol JEM2100
167 at 200 kV (CCiT-UB) as well.

168 Atoms per formula unit (apfu) were calculated from the atomic concentrations obtained by AEM and
169 based on the theoretical number of positive charges of each mineral (14, 22 and 32 for serpentine,
170 talc and sepiolite, respectively). Fe was calculated as Fe^{3+} , following Villanova-de-Benavent et al.
171 (2014a) and according to the $\text{Fe}^{2+}/\text{Fe}^{3+}$ XANES maps obtained in equivalent samples (Roqué-Rosell
172 et al. submitted).

173 RESULTS

174 TEM results are shown in Figures 2-7, including a photograph of the selected sample (Figs. 2a-6a), a
175 photomicrograph of the ion thinned specimen (Figs. 2b-6b), and low magnification, high resolution
176 images and electron diffraction patterns. A comparison between the chemical compositions of the
177 minerals obtained with TEM-AEM and EMP is shown in Figure 8, and AEM analyses are presented
178 in Table 2.

179 Mineralogy and textures at the nano scale

180 **Ni-bearing serpentine-dominant (type I).** Type I garnierite consists of serpentine particles and
181 minor talc-like bent thin lamellae with very little porosity (Fig. 2c). Serpentine occurs as tubes of
182 various sizes which are randomly oriented, as basal and longitudinal sections are observed in low
183 magnification images. There are two groups of serpentine tubes: i) thick, short tubes up to 5000 Å
184 long and about 3000 Å diameter, with 85 Å in diameter hollow cores; ii) less abundant, narrow, long
185 tubes up to 3000 Å in length and around 750 Å in diameter, with ~50 Å hollow cores (Fig. 2d).
186 These nanostructures together with the particle size indicate that this sample is formed mostly by
187 polygonal serpentine in relatively short fibers, finely intermixed with talc-like lamellae. Uyeda et al.

188 (1973) also reported short, stubby tubes with central holes in serpentine-like garnierites from Brazil,
189 with an average width of 740 Å (4.4 wt.% NiO; serpentine-talc mixture with a talc fraction of 0.13),
190 with less abundant ill-defined platy fragments, which may resemble the talc-like particles in type I
191 garnierite (Fig. 2c, d).

192 As seen in Fig. 2d, the short tubes display bending at the fiber tip. These features have also been
193 observed in polygonal serpentines coexisting with polyhedral serpentines by Andréani et al. (2008).
194 In the high resolution images of the short tubes longitudinal sections 7.3 Å spacings are observed, in
195 some cases presenting dislocations (Fig. 2e). The 7.3 Å spacings were confirmed in both the electron
196 diffraction patterns and the Fast Fourier Transform obtained for the same short tubes, which
197 displayed few reflections of the $h0l$ plane that can be indexed with the clinochrysotile- $2M_{c1}$ 14 Å
198 structure (Brindley and Brown 1980) (Fig. 2f). However, no high resolution images of the long tubes
199 could be obtained due to electron beam damage. According to the total diameters and the measured
200 spacings, the short, thick ones have approximately 200 T-O layers and the long, thin ones have 50 T-
201 O layers.

202 **Ni-bearing mixture of serpentine- and talc-like particles (type II).** In the specimen of type II
203 garnierite, serpentine particles, with characteristic 7.2 Å spacings, display various shapes and sizes
204 and are found scattered in a matrix of talc-like bundles (Fig. 3c, d). Some serpentine basal sections
205 have diameters around 5000 Å with tiny hollow cores, and are divided in 15 equidimensional sectors
206 with an angle of 24° between adjacent (001) lattice planes, so-called polygonal serpentine (Fig. 3e).
207 In some cases, polygonal sections are kidney-shaped, resembling the polyhedral serpentine of
208 Andréani et al. (2008) (Fig. 3f). The inner diameter of some polygonal serpentines is similar to the
209 outer diameter of the chrysotile basal sections in type I (Fig. 3g). Other serpentine particles display
210 less rounded cylindrical fibers, probably corresponding to oblique sections of serpentine tubes,
211 smaller than the polygonal fibers (up to 3000 Å in diameter), which are hollow-cored or contain

212 disordered layers in the centre, probably of a talc-like phase (Fig. 3g). Uyeda et al. (1973) also
213 observed those features and mentioned that this material in the cores could be amorphous. Talc-like
214 lamellae were also observed to concentrate in curved aggregates (Fig. 3h).

215 **Ni-dominant mixture of serpentine- and talc-like particles (type III).** Type III garnierites display
216 long and bent lamellae with basal fringe spacings of 9.2–9.4 Å and 7.2–7.4 Å mixed in single
217 particles (Fig. 4c-e), which are frequently parallel to each other (Fig. 4d-e). Lamellae of 7.2–7.4 Å
218 spacings are curved, more regular and thicker (from six to sixteen 7.2–7.4 Å spacings) when
219 compared to the 9.2–9.4 Å lamellae (four fringes mostly, and up to six) (Fig. 4c-e). This is supported
220 by electron diffraction patterns, showing that the sample is composed of crystalline lizardite with a
221 7.2 Å spacing, coexisting with a lower crystallinity talc-like phase (Fig. 4f). No polygonal serpentine
222 was found and only one basal and one longitudinal section of chrysotile tubes were observed in the
223 specimen. Similar features involving ~7 Å and ~10 Å structures were observed in a garnierite from
224 Oregon (18.6 wt.% NiO, $X_{\text{talc}} = 27\%$) by Uyeda et al. (1973). These values differ from those of the
225 type III garnierite (29–50 wt.% NiO and average X_{talc} of 47 %, Villanova-de-Benavent et al.
226 2014a). Garnierites from New Caledonia shown by Pelletier (1983) and references therein also
227 displayed comparable features.

228 **Talc-like particles (type IV).** The specimen containing Ni-“kerolite”-“pimelite” garnierites is very
229 homogeneous, as it consists uniquely of sets of superimposed and/or plaited, thin, apparent lamellae
230 with two to six basal spacings of 9.5–9.7 Å (Fig. 5c, d, e), confirmed by the absence of ~7 Å fringes
231 in this specimen. Identical textures were observed in 10 Å-type garnierites from Brazil by Esson and
232 Carlos (1978), in which no other crystalline phases were detected either. These Brazilian garnierites
233 under the optical microscope display botryoidal features and coexist with silica, like type IV
234 garnierites (Villanova-de-Benavent et al. 2014a). The small particle size, together with the wide and

235 diffuse rings observed in the electron diffraction patterns (typical of a random polycrystalline
236 pattern), suggest that the sample may have low crystallinity (Fig. 5f).

237 **Sepiolite-falcondoite (type V).** The obtained TEM bright field images demonstrate the presence of
238 Ni-rich sepiolite roughly oriented in fine ribbons with lengths above several hundreds of nanometers
239 (Fig. 6c, d) and frequently enclosed within an amorphous silica matrix (Fig. 6b, c, e). The high
240 magnification image in Fig. 6e shows a normal section of the fiber with the well defined (110)
241 crystalline planes. This image allows measuring the 12 Å spacing typical of sepiolite-falcondoite
242 (Post et al. 2007, Tauler et al. 2009).

243 **Mineral chemistry at the nanoscale**

244 The composition of garnierite-forming minerals obtained by AEM is represented in Fig. 7, compared
245 to previous analyses obtained by EMP (Villanova-de-Benavent et al. 2014a), and in Table 2. The
246 identification of the mineral species based on the lattice fringe spacing and that from the chemical
247 composition were coherent. In addition, to increase the number of analysis of each mineral species,
248 additional analyses were obtained on particles deposited on copper grids (e.g. López Munguira and
249 Nieto 2000, Abad et al. 2001).

250 In general, Al is virtually absent in the Falcondo garnierites. Serpentine particles have low Ni, up to
251 0.2 apfu in type I and 0.6 apfu in type II; and yield remarkable amounts of Fe, 0.1–0.4 apfu in type I
252 and up to 0.3 in type II garnierites. Type III compositions are actually mixed analyses of lizardite
253 lamellae and the talc-like phase, with Ni contents ranging from 2.7 to 3.3 apfu and Al and Fe below
254 detection limits. Talc-like analyses from all the studied garnierite types yield highly variable Ni
255 contents, between 1.2 to 2.3 apfu (1.7 on average), and Al and Fe are generally low (up to 0.13 apfu
256 Al, 0.03 on average; and up to 0.04 apfu Fe, <0.01 on average). In addition, Ni content in talc-like
257 particles increases gradually from type I to type IV garnierites (Fig. 7).

258 AEM results plot within or near the compositional fields previously obtained through EMP in the
259 same samples (Villanova-de-Benavent et al. 2014a), although type III mixtures and some analyses of
260 type II serpentines still deviate towards the “kerolite”-“pimelite” series. Even if in most cases it was
261 impossible to obtain pure analysis of the constituent phases of the mixture due to the very small size
262 of the individual packets (e.g. Fig. 4c, d, e), the AEM spot size of tens of nanometers versus the
263 EMP spot size of one micrometer gives unique results of the preferential partitioning of Ni in talc-
264 like particles rather than in the associated serpentines. In summary, the higher spatial resolution of
265 AEM analyses allowed better discrimination than EMP analyses, with slight deviations from the
266 average composition obtained by EMP toward the respective constituent end-members (Fig. 7).
267 Finally, the EDS spectra in type V garnierite enabled distinction of the Ni-sepiolite ribbons from the
268 surrounding matrix. The Ni-sepiolite fibers show a distinctive Ni and Mg content in addition to Si
269 and O, and the presence of Si seems to be related not only with sepiolite but also with amorphous
270 SiO₂ matrix. Ni contents varied from 1.2 to 2.2 apfu and Al and Fe were below detection.

271 DISCUSSION

272 **Nanotextural variability of garnierite from the Falcondo Ni-laterite**

273 As stated by Brindley (1978), most garnierites are a mixture of serpentine-like and talc-like phases,
274 and previous TEM imaging studies reported the occurrence of some garnierites as intimate mixtures
275 of 7.2–7.4 and 9.2–9.7 Å phases at the nanometer scale (e.g. Uyeda et al. 1973, Poncelet et al. 1979,
276 Soler et al. 2008). Most garnierites from the Falcondo Ni-laterite deposit (Figs. 2-5) actually consist
277 of mixtures of different relative proportions of serpentine and talc-like phases (“kerolite”-
278 “pimelite”). TEM analysis distinguished a wide variety of textures and mineral species of the
279 serpentine minerals in the different garnierite mixtures (chrysotile tubes in type I, polygonal
280 serpentine in type II, lizardite lamellae in type III). In addition, TEM revealed the presence talc-like

281 particles in all the serpentine-bearing garnierites even when these had not been detected by XRD, as
282 in type I, probably due to their small quantity and low crystallinity.

283 Another characteristic feature is that 7.2–7.4 Å fringes occurred in large numbers, whereas 9.2–9.7
284 Å fringes occurred in smaller sets (as stated by Uyeda et al. 1973), suggesting a lower crystallinity of
285 talc-like phases, which is coherent with their broad peaks, or absence of them, in powder X-ray
286 diffraction (Villanova-de-Benavent et al. 2014a) and with the diffuse and weak ring-shaped electron
287 diffraction patterns in this study. Kato (1961) also reported that talc-like garnierites from New
288 Caledonia produced ring-shaped selected area electron diffraction (SAED) patterns whereas
289 serpentine-like phases gave single crystal diffraction patterns with well-defined spots. Furthermore,
290 Pelletier (1983) (and references therein) distinguished the electron diffraction pattern of the talc-
291 willemseite, with a regular structure, from the “kerolite”-“pimelite”, showing concentric circles.
292 Other studies reported unspecified disordered regions in the TEM photomicrographs as well (e.g.
293 Uyeda et al. 1973, Brindley 1978, Esson and Carlos 1978).

294 Such low crystallinity of the talc-like phase could be related to specific conditions, such as very low
295 temperature, high water availability and/or high growth rate during crystallization. The fact that the
296 10 Å fringes are less regularly defined than the 7 Å ones, could be also explained by a variable
297 degree of hydration of their interlayers (Brindley and Hang 1973, Uyeda et al. 1973) and/or possible
298 volatilization phenomena under the TEM vacuum.

299 Finally, sepiolite-falcondoite occurs as an independent phase, never mixed with serpentine nor talc-
300 like garnierites. It displays the characteristic elongated ribbon shape of sepiolite, commonly related
301 to amorphous silica and/or quartz, but has a remarkably higher Ni content (26.8 wt.% NiO) when
302 compared to sepiolites examined under TEM in other localities (maximum 3.3 wt.% NiO in
303 Indonesia; Kuhnel et al. 1978).

304 **Preferential Ni concentration in the talc-like structure**

305 According to AEM, talc-like phases yield higher Ni concentrations than serpentine, which is
306 coherent with the good correlation between Ni content and the talc fraction in the garnierites from
307 Falcondo, as suggested by Galí et al. (2012) and Villanova-de-Benavent et al. (2014a), and by Soler
308 et al. (2008), in the Loma de Hierro Ni-laterite (Venezuela). The higher resolution of the AEM with
309 respect to EMP enabled distinction of the serpentine particles from the “kerolite”-“pimelite”
310 lamellae (Fig. 7), despite it being difficult to obtain pure, single-phase analyses of talc-like or
311 serpentine in type III. Therefore, the distribution of Ni between serpentine and talc-like phases in
312 type III could not be well established because of its finer particle size.

313 Despite Uyeda et al. (1973) stating that there was no correlation between Ni content and morphology
314 of the particles under TEM, and between proportions of platy and elongated particles and the number
315 of serpentine-like and talc-like layers (equivalent to serpentine or talc fraction, respectively), other
316 studies reported that Ni is mostly concentrated in talc-like phases (e.g. Esson and Carlos 1978,
317 Poncelet et al. 1979). Besides, Vitovskaya and Berkhin (1968) also showed some low-magnification
318 and electron diffraction patterns of what they described as Mg- and Ni-bearing “kerolites” (actually
319 garnierite mixtures). They identified typical, tiny clinochrysotile tubes in Mg-“kerolite” (consisting
320 predominantly of serpentine with Ni below detection limit), and scales with tiny tubes of
321 clinochrysotile in the Ni-“kerolite” (mostly a 10 Å mineral with 11.3 wt.% NiO). Consequently in
322 this case Ni was also related to a 10 Å phase. Furthermore, Poncelet et al. (1979) showed by EMP
323 analyses and by heating experiments that most of the nickel was concentrated in the octahedral layer
324 of the 10 Å phase, and not homogeneously distributed in the octahedral layer of both the 7 Å and 10
325 Å. During these experiments, metallic Ni particles were deposited onto the 10 Å flakes and rarely
326 onto the serpentine fibers after heating the sample. In addition to these observations and
327 interpretations, our AEM data confirm that the Ni was preferably contained in the talc-like phase.
328 Possibly, the TOT talc structure is more likely to host Ni than the TO serpentine structure, as

329 demonstrated by the equilibrium constants of the simultaneous precipitation of Ni-serpentine and
330 “kerolite”-“pimelite” (Galí et al. 2012).

331 AEM results of garnierites from the Falcondo Ni-laterite showed that Ni concentration in serpentine
332 is lower than in the talc-like particles, and always below 50% of the total elements in octahedral
333 coordination. Therefore, neither népouite nor pecoraite, the Ni analogues of lizardite and chrysotile
334 respectively, are found in the garnierites from the Falcondo Ni-laterite deposit. However, the
335 presence of népouite and pecoraite is widely reported in other localities. In particular, népouite has
336 been extensively studied in New Caledonia (e.g. Brindley and Wan, 1975; Wells et al. 2009 and
337 references therein). One possible explanation for the different mineralogy between New Caledonian
338 and Dominican Ni-laterites is the lithology of the primary ultramafic rocks. In New Caledonia, the
339 protolith is mainly harzburgite and dunite (e.g. Pelletier, 1983, 1996), whereas in the Dominican
340 Republic the protolith is mostly clinopyroxene-rich harzburgite and lherzolite, commonly intruded
341 by microgabbro and dolerite dykes (e.g. Marchesi et al., 2012). The greater pyroxene content may
342 imply a greater availability of silica in the Falcondo Ni-laterite deposit, leading to the preferential
343 formation of talc rather than serpentine during weathering.

344 When examined under TEM, népouite usually occurs as highly crystalline plates (34.8 wt.% NiO,
345 Montoya and Baur 1963, 47.6 wt.% NiO, Manceau and Calas 1985), whereas pecoraite develops
346 coils and spiral shapes with three to five revolutions about the spiral axis (Faust et al. 1969, 1973,
347 Milton et al. 1983). In contrast, the serpentine particles in Falcondo present larger diameters and
348 lengths than the Ni-dominant counterparts described in the literature. Roy and Roy (1954) and
349 Milton et al. (1983) suggested that the substitution of Mg by Ni in chrysotile gives rise to non-
350 tubular but also not well formed, platy crystals. The Ni substitution probably prevents the tubes to
351 grow up to several hundreds of Å and to develop well rounded spirals or concentric tubes.
352 Accordingly, despite that the serpentine particles in type III were not analysed by AEM due to their

353 small particle size, the possibility of these serpentine lamellae yielding higher Ni contents than the
354 chrysotile tubes and polygonal serpentine in types I and II cannot be discarded.

355 **The formation mechanism of garnierites in the Falcondo Ni-laterite**

356 Low magnification and high resolution images are fundamental to understand the genetic
357 relationships between serpentine (in most cases, polygonal fibers) and Ni-rich, talc-like phases. The
358 talc-like lamellae are observed inside the serpentine central hole (Fig. 8a, b, e), surrounding the
359 serpentines, forming at the very edges of the particles (Fig. 8a, c, d, f) and at the boundaries between
360 adjacent sectors of the largest polygonal serpentines (Fig. 8e, f). In addition, HRTEM details indicate
361 that, at the reaction front, the (001) talc-like planes are parallel to the 001 serpentine ones. This
362 textural evidence suggests that talc-like phases form after serpentine, taking advantage of high stress
363 sites, such as the outer rims (where the basal planes of serpentine are bent), the fiber cores (highest
364 layer curvature), and the intersector boundaries. When replacement of serpentine by talc is more
365 advanced, sectors become fully pseudomorphed by talc-like lamellae (Fig. 8g, h) and talc-like
366 lamellae may develop rounded aggregates, mimicking the rounded shapes of former serpentine
367 particles (Fig. 8i, j). It is worth noting that these poorly crystalline, talc-like layers are bent and
368 characterized by extremely wide interlayer partings, giving rise to a highly porous nanotexture (Fig.
369 5d, e; 8h).

370 The overall data indicates that the formation of the successive phases in the garnierite mixtures may
371 be explained by an early formation of serpentine tubes. The diameter of the serpentine cores in type
372 II is comparable to the outer diameter of narrow chrysotile tubes in type I. This may indicate that the
373 diameter of the chrysotile increases in size until its curled structure is unstable. Then it is
374 transformed into polygonal serpentine while preserving a cylindrical core, which may more likely be
375 altered by later dissolution-precipitation than polygonal sectors. The serpentine particles are
376 subsequently replaced by “kerolite”-“pimelite”, starting from high stress sites and structural

377 discontinuities, such as cores, rims and contact between sectors. The final product is possibly a
378 garnierite formed exclusively by talc-like lamellae (Fig. 9).

379 The later formation of “kerolite”-“pimelite” from serpentine particles is in accordance to the
380 garnierite precipitation model proposed by Galí et al. (2012). This model is based on the assumption
381 that in an Al-free system, such as the Falcondo laterite profile, the stability of serpentine, kerolite–
382 pimelite or sepiolite–falcondoite is mainly controlled by the silica activity. As a result, the ideal
383 formation of the Ni ore occurs as a successive precipitation of mineral phases progressively enriched
384 in Ni and Si, because silica activity increases with time and through the profile. Thus, the first
385 garnierite-forming phase to precipitate is serpentine, then followed by “kerolite”-“pimelite”, and
386 eventually by sepiolite–falcondoite and sepiolite–falcondoite with amorphous silica and/or quartz
387 (Galí et al. 2012).

388

CONCLUSIONS

389 ~~This work provides low and high resolution TEM images, showing that the five types of garnierites~~
390 ~~present characteristic features at the nanometer scale. Most consist of an ultrafine mixture of~~
391 ~~serpentine and a talc-like phases, but nanotextures and the relative amounts of serpentine and talc-~~
392 ~~like particles are variable from sample to sample.~~

393 ~~Serpentine occurred as chrysotile tubes, polygonal serpentine and lizardite lamellae, and in general~~
394 ~~was more crystalline than the associated talc-like particles. We note that all garnierite samples~~
395 ~~contained at least a small portion of talc-like lamellae, even when this phase was not detected by~~
396 ~~XRD, due to its low crystallinity and/or low abundance. These particles, which could be the only~~
397 ~~component of garnierites, are in fact minerals of the “kerolite”-“pimelite” series, a talc-like phase~~
398 ~~with extra water in their structure and showing an expansion of d_{001} -spacing from 9.2 (talc) to 9.7 Å.~~
399 ~~AEM data confirmed that Ni was mostly concentrated in “kerolite”-“pimelite” instead of in~~
400 ~~serpentine particles. Low magnification and HRTEM images established an insight of the~~

401 ~~mechanisms by which Falcondo garnierites form, showing the preferential replacement of serpentine~~
402 ~~by “kerolite”-“pimelite” at high stress sites in the serpentine structure, such as the inner walls of~~
403 ~~cores, outer rims and intersector boundaries.~~

404 **IMPLICATIONS**

405 The results obtained in this comprehensive TEM study on the garnierites from the Falcondo Ni-
406 laterite deposit (Dominican Republic) reveal that the Ni-bearing serpentine, “kerolite”-“pimelite”
407 and sepiolite-falcondoite components display a wide variety of textures and sizes at the nanometer
408 scale. It is worth noting that most of the observed textures in Mg-serpentine (e.g. lamellae,
409 chrysotile tubes, polygonal) are also found in their Ni-bearing analogues. However, the studied
410 samples are characterized by low Ni contents in serpentines, whereas Ni is concentrated in the talc-
411 like phases, reinforcing previous works (Villanova-de-Benavent et al., 2014). The results altogether
412 provide further insight on the mineralogy of Ni-phyllsilicates as the highest grade Ni ores in a
413 world-class hydrous silicate type Ni-laterite deposit.

414 The most noteworthy result of this work is the first evidence of replacement at the nanometer scale
415 among garnierite-forming minerals. We have demonstrated that the Ni-enriched talc-like lamellae
416 replace the polygonal serpentine and chrysotile tubes in the higher stress sites. In this model the Ni-
417 enriched talc finally replaces the serpentine completely leaving some rolled and curved talc lamellae
418 as remnants of the former serpentine (Fig. 9). This provides the direct proof that the crystallization of
419 the Ni-bearing phyllosilicates at low temperature is associated with an increase in the silica activity
420 of the system, which is coherent with previous observations at the micrometer scale (Villanova-de-
421 Benavent et al., 2014) and the thermodynamic model for garnierites that predicted that Ni is
422 preferably contained within the talc-like phase (Galí et al., 2012). As a consequence, Ni-enriched
423 talc phases precipitate from already formed secondary serpentine particles. In the long term, in a
424 more advanced stage of weathering, this will lead to the complete dissolution of serpentine particles

425 that will be entirely substituted by Ni-enriched talc-like and even sepiolite-like phases as the silica
426 and Ni activity increase. In addition it is worth noting that this Ni-enrichment process leads to the
427 formation of remarkable porosity in the material.

428 Therefore, the results of this work contribute to the knowledge of the formation of Ni-phylosilicates
429 under tropical conditions demonstrating that the Ni-poor serpentine particles are replaced by Ni-rich
430 talc-like phases. This in turn has interesting implications in the field of material sciences. Recently
431 (Sivaiah et al., 2011), synthesized Ni-containing serpentine-like and talc-like phyllosilicates were
432 used as catalyst precursors for processing greenhouse gases (i.e. CO₂, CH₄). Thus the TEM
433 characterization presented in this work also suggests that natural garnierites could be a good
434 candidate to accomplish the requirements of a suitable catalyst precursor.

435

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448

449 **REFERENCES**

- 450 Abad, I., Mata, M.P., Nieto, F., and Velilla, N. (2001) The phyllosilicates in diagenetic-metamorphic
451 rocks of the South Portuguese Zone, southwestern Portugal. *Canadian Mineralogist*, 39, 1571-1589.
- 452 Aiglsperger, T., Proenza, J.A., Zaccarini, F., Lewis, J.F., Garuti, G., Labrador, M., and Longo, F.
453 (2015) Platinum group minerals (PGM) in the Falcondo Ni-laterite deposit, Loma Caribe peridotite
454 (Dominican Republic). *Mineralium Deposita*, 50, 105-123.
- 455 Andréani, M., Grauby, O., Baronnet, A., and Muñoz, M. (2008) Occurrence, composition and
456 growth of polyhedral serpentine. *European Journal of Mineralogy*, 20, 159-171.
- 457 Baronnet, A., and Devouard, A. (2005) Microstructures of common polygonal serpentines from axial
458 HRTEM imaging, electron diffraction and lattice-simulation data. *Canadian Mineralogist*, 43, 513-
459 542.
- 460 Brindley, G.W. (1978) The structure and chemistry of hydrous nickel containing silicate and
461 aluminate minerals. *Bulletin du Bureau de Recherches Géologiques et Minières*, section II, 3, 233-
462 245.
- 463 Brindley, G.W. (1980) The structure and chemistry of hydrous nickel-containing silicate and nickel-
464 aluminium hydroxy minerals. *Bulletin de Minéralogie*, 103, 161-169.
- 465 Brindley, G.W., and Brown, G. (1980) Crystal structures of clay minerals and their X-ray
466 identification, 495 p. *Mineralogical Society Monograph 5*, London.
- 467 Brindley, G.W., and Hang, P.T. (1973) The nature of garnierite: I. Structure, chemical compositions
468 and color characteristics. *Clays and Clay Minerals*, 21, 27-40.
- 469 Brindley, G.W., and Maksimović, Z. (1974) The nature and nomenclature of hydrous nickel-
470 containing silicates. *Clay Minerals*, 10, 271-277.
- 471 Brindley, G.W., and Wan, H.-M. (1975) Compositions, structures and thermal behaviour of
472 nickel-containing minerals in the lizardite–népouite series. *American Mineralogist*, 60, 863–871.

- 473 Cathelineau, M., Caumon, M.C., Massei, F., Brie, D., Harlaux, M. (2015) Raman spectra of Ni-Mg
474 kerolite: effect of Ni-Mg substitution on O-H stretching vibrations. *Journal of Raman Spectroscopy*.
475 DOI: 10.1002/jrs.4746.
- 476 Dosbaba, M., and Novák, M. (2012) Quartz replacement by “kerolite” in graphic quartz-feldspar
477 intergrowths from the Věžná I pegmatite, Czech Republic: a complex desilication process related to
478 episyenitization.. *Canadian Mineralogist*, 50, 1609-1622.
- 479 Elias, M. (2002) Nickel laterite deposits—Geological overview, resources and exploration. In D.
480 Cooke, and J. Pongratz, Eds., *Giant Ore Deposits—Characteristics, Genesis, and Exploration*,
481 CODES Special Publication 4, p. 205–220. Hobart, University of Tasmania, Australia.
- 482 Esson, J., and Carlos, L. (1978) The occurrence, mineralogy and chemistry of some garnierites from
483 Brazil. *Bulletin du Bureau de Recherches Géologiques et Minières*, section II, 3, 263-274.
- 484 Faust, G.T. (1966) The hydrous nickel-magnesium silicates – the garnierite group. *American*
485 *Mineralogist*, 51, 33-36.
- 486 Faust, G.T., Fahey, J.J., Mason, B., and Dwornik, E.J. (1969) Pecoraite, $\text{Ni}_6\text{Si}_4\text{O}_{10}(\text{OH})_8$, nickel
487 analog of clinochrysotile, formed in the Wolf Creek meteorite. *Science*, 165, 59-60.
- 488 Faust, G.T., Fahey, J.J., Mason, B., and Dwornik, E.J. (1973) The disintegration of the Wolf Creek
489 meteorite and the formation of pecoraite, the nickel analog of clinochrysotile. *United States*
490 *Geological Survey Professional Paper*, 3480C, 107-135.
- 491 Freyssinet, Ph., Butt, C.R.M., and Morris, R.C. (2005) Ore-forming processes related to lateritic
492 weathering. *Economic Geology*, 100th Anniversary Volume, 681-722.
- 493 Galí, S., Soler, J.M., Proenza, J.A., Lewis, J.F., Cama, J., and Tauler, E. (2012) Ni-enrichment and
494 stability of Al-free garnierite solid-solutions: a thermodynamic approach: *Clays and Clay Minerals*,
495 60, 121-135.

- 496 Garnier, J. (1867) Essai sur la géologie et les ressources minérales de la Nouvelle-Calédonie. In
497 Annales des mines, 6ème série, tome XII., 92 p. Dunod, Paris.
- 498 Gleeson, S.A., Butt, C.R., Elias, M. (2003) Nickel laterites: a review. Society of Economic
499 Geologists Newsletter 54, 11–18.
- 500 Gleeson, S.A., Herrington, R.J., Durango, J., Velásquez, C.A., Koll, G. (2004) The mineralogy and
501 geochemistry of the Cerro Matoso S.A. Ni laterite deposit, Montelibano, Colombia. Society of
502 Economic Geology 99, 1197-1213.
- 503 Golightly, J.P. (1981) Nickeliferous laterite deposits. Economic Geology, 75th Anniversary Volume,
504 710-735.
- 505 Haldemann, E.G., Buchan, R., Blowes, J.H., and Chandler, T. (1979) Geology of lateritic nickel
506 deposits, Dominican Republic. In D.J.I. Evans, R.S. Shoemaker, H. Veltman, Eds., International
507 Laterite Symposium, 4, p. 57-84. Society of Mining Engineers of the American Institute of Mining,
508 Metallurgical, and Petroleum Engineers, New York.
- 509 Kato, T. (1961) A study on the so-called garnierite from New-Caledonia. Mineralogical Journal, 3,
510 107-121.
- 511 Kuhnel, R.A., Roorda, H.J., and Steensma, J.J.S. (1978) Distribution and partitioning of elements in
512 nickeliferous laterites. Bulletin du Bureau de Recherches Géologiques et Minières section II, 3, 191-
513 206.
- 514 Lewis, J.F., Draper, G., Proenza, J.A., Espaillet, J., and Jiménez, J. (2006) Ophiolite-Related
515 Ultramafic Rocks (Serpentinites) in the Caribbean Region: A Review of their Occurrence,
516 Composition, Origin, Emplacement and Ni-Laterite Soils Formation. Geologica Acta, 4, 237-263.
- 517 Lithgow, E.W. (1993) Nickel laterites of central Dominican Republic Part I. Mineralogy and ore
518 dressing. In R.G. Reddy, R.N. Weizenbach, Eds., The Paul E. Queneau International Symposium,

- 519 Extractive Metallurgy of Copper, Nickel and Cobalt, Volume I: Fundamental Aspects, p. 403-425.
520 The Minerals, Metals and Materials Society, Portland.
- 521 Lopez Munguira, A, and Nieto, F. (2000) Transmission Electron Microscopy study of very low-
522 grade metamorphic rocks in Cambrian sandstones and shales. Ossa-Morena Zone. South-West
523 Spain. Clays and Clay Minerals, 48, 213-223.
- 524 Manceau, A., and Calas, G. (1985) Heterogeneous distribution of nickel in hydrous silicates from
525 New Caledonia ore deposits. American Mineralogist, 70, 549-558.
- 526 Marchesi, C., Garrido, C.J., Proenza, J.A., Konc, Z., Hidas, K., Lewis, J.F., and Lidiak, E., (2012)
527 Mineral and whole rock compositions of peridotites from Loma Caribe (Dominican
528 Republic): insights into the evolution of the oceanic mantle in the Caribbean region. European
529 Geosciences Union General Assembly, Geophysical Research Abstracts 14, EGU2012–EGU12161.
530 Vienna, Austria.
- 531 Mellini, M. (2013) Structure and microstructure of serpentine minerals. In F. Nieto, K.J.T. Livi,
532 Eds., Minerals at the nanoscale, p. 153-179. European Mineralogical Union Notes in Mineralogy
533 (Mineralogical Society of Great Britain and Ireland), London.
- 534 Milton, C., Dwornik, E.J., and Finkelman, R.B. (1983) Pecoraite, the nickel analogue of chrysotile,
535 $\text{Ni}_3\text{Si}_2\text{O}_5(\text{OH})_4$ from Missouri. Neues Jahrbuch für Mineralogie - Monatshefte, 11, 513-523.
- 536 Montoya, J.W., and Baur, G.S. (1963) Nickeliferous serpentines, chlorites and related minerals
537 found in two lateritic ores. American Mineralogist, 48, 1227-1238.
- 538 Moraes, L.J. (1935) Niquel no Brasil. Boletim Republica dos Estados Unidos do Brasil, 9, 168 p.
- 539 Nickel, E.H., and Nichols, M.C. (2009) Materials Data Minerals Database (Online). Available:
540 <http://www.materialsdata.com>. (accessed June 19, 2015). Livermore, California.
- 541 Post, J.E., Bish, D.L., and Heaney, P.J. (2007) Synchrotron powder X-ray diffraction study of the
542 structure and dehydration behavior of sepiolite. American Mineralogist, 92, 91-97.

- 543 Pecora, W.T., and Hobbs S.W. (1942) Nickel deposit near Riddle Douglas County, Oregon. United
544 States Geological Survey Bulletin, 931-I, 205-225.
- 545 Pecora, W.T., Hobbs S.W., and Murata, K.J. (1949) Variations in garnierite from the nickel deposit
546 near Riddle, Oregon. Economic Geology, 44, 13-23.
- 547 Pelletier, B. (1983) Localisation du nickel dans les minerais “garniéritiques” de Nouvelle-Calédonie.
548 In D. Nahon, Ed., International Congress on Alteration Petrology, Sciences Géologiques Mémoires,
549 73, p. 173-183. Centre National de la Recherche Scientifique, Paris.
- 550 Pelletier, B. (1996) Serpentes in nickel silicate ore from New Caledonia. Australasian Institute of
551 Mining and Metallurgy Publication Series - Nickel conference, 6/96, p. 197-205. Kalgoorlie,
552 Western Australia.
- 553 Poncelet, G., Jacobs, P., Delannay, F., Genet, M., Gerard, P., and Herbillon, A. (1979) Étude
554 préliminaire sur la localisation du nickel dans une garnierite naturelle. Bulletin de Minéralogie, 102,
555 379-385.
- 556 Proenza, J.A., Zaccarini, F., Lewis, J., Longo, F., and Garuti, G. (2007) Chromite composition and
557 platinum-group mineral assemblage of PGE-rich Loma Peguera chromitites, Loma Caribe peridotite,
558 Dominican Republic. Canadian Mineralogist, 45, 211-228.
- 559 Roqué-Rosell, J., Villanova-de-Benavent, C., Proenza, J.A. (submitted) The accumulation of Ni in
560 serpentines and garnierites from Falcondo Ni-laterite deposit (Dominican Republic) elucidated by
561 means of μ XAS. Geochimica et Cosmochimica Acta.
- 562 Roy, D.M. and Roy, R. (1954) An experimental study of the formation and properties of synthetic
563 serpentines and related layer silicate minerals. American Mineralogist, 39, 957-975.
- 564 Sivaiah, M.V., Petit, S, Beaufort, M.F., Eyidi, D., Barrault, J., Batiot-Dupeyrat, C., Valange, S.
565 (2011) Nickel based catalysts derived from hydrothermally synthesized 1:1 and 2:1 phyllosilicates as

- 566 precursors for carbon dioxide reforming of methane. *Microporous and Mesoporous Materials*, 140,
567 69-80.
- 568 Soler, J.M., Cama, J., Galí, S., Meléndez, W., Ramírez, A., and Estanga, J. (2008) Composition and
569 dissolution kinetics of garnierite from the Loma de Hierro Ni-laterite deposit, Venezuela. *Chemical*
570 *Geology*, 249, 191-202.
- 571 Springer, G. (1974) Compositional and structural variations in garnierites. *Canadian Mineralogist*,
572 12, 381-388.
- 573 Suárez, S., Nieto, F., Velasco, F., and Martín, F.J. (2011) Serpentine and chlorite as effective Ni-Cu
574 sinks during weathering of the Aguablanca sulphide deposit (SW Spain). TEM evidence for metal-
575 retention mechanisms in sheet silicates. *European Journal of Mineralogy*, 23, 179-196.
- 576 Tauler, E., Proenza, J.A., Galí, S., Lewis, J.F., Labrador, M., García-Romero, E., Suárez, M., Longo,
577 F., and Bloise, G. (2009) Ni-sepiolite-falcondoite in garnierite mineralization from the Falcondo Ni-
578 laterite deposit, Dominican Republic. *Clay Minerals*, 44, 435-454.
- 579 Uyeda, N., Hang, P.T., and Brindley, G.W. (1973) The nature of garnierites: II. Electron-optical
580 study. *Clays and Clay Minerals*, 21, 41-50.
- 581 Varela, J.dD. (1984) La estructura de las garnieritas. *Geología Colombiana*, 13, 29-40.
- 582 Villanova-de-Benavent, C., Proenza, J.A., Galí, S., García-Casco, A., Tauler, E., Lewis, J.F., and
583 Longo, F. (2014a) Garnierites and garnierites: Textures, mineralogy and geochemistry of garnierites
584 in the Falcondo Ni-laterite deposit, Dominican Republic. *Ore Geology Reviews*, 58, 91–109.
- 585 Villanova-de-Benavent, C., Proenza, J.A., Galí, S., Nieto, F., García-Casco, A., Roqué-Rosell, J.,
586 Tauler, E., and Lewis, J.F. (2014b) Mineralogy of Ni-phyllsilicates in the Falcondo Ni-laterite
587 deposit (Dominican Republic): A multiscale approach. Proceedings of the twenty first meeting of the
588 International Mineralogical Association IMA2014, 298.

589 Vitovskaya, I.V., and Berkhin, S.I. (1968) К вопросу о природе керолита (On the question of the
590 nature of kerolite). *Kora Vyvetrivaniya*, 10, 134-159.

591 Wells, M.A., Ramanaidou, E.R., Verrall, M., and Tessarolo, C. (2009) Mineralogy and crystal
592 chemistry of “garnierites” in the Goro lateritic nickel deposit, New Caledonia. *European Journal of*
593 *Mineralogy*, 21, 467-483.

594 Whitney, D.L., and Evans, B.W. (2010) Abbreviations for names of rock-forming minerals.
595 *American Mineralogist*, 95, 185-187.

596 TABLE AND FIGURE CAPTIONS

597

598 **Table 1.** Summary of the studied garnierite samples from the Falcondo Ni-laterite (Dominican
599 Republic). The talc fraction (X_{talc}) in the serpentine-talc mixtures is calculated according to
600 Brindley & Hang (1973).

601 **Table 2.** Representative normalized AEM analyses performed on Falcondo garnierites in atoms per
602 formula unit (apfu).

603

604 **Figure 1.** Geological map of the Falcondo Ni-laterite (a) after Haldemann et al. (1979), location of
605 the deposit in the Dominican Republic (b) and schematic profile of the Falcondo Ni-laterite (c)
606 modified from Lithgow et al. (1979), Lewis et al. (2006), Villanova-de-Benavent et al. (2014a).

607 **Figure 2.** Type I garnierite as seen under the TEM (specimen LC-101): a) image of the sample, the
608 specimen was detached from the area in the red rectangle; b) detail of the hole in the grid under the
609 optical microscope (plane polarised light); c) low magnification images of the specimen showing
610 short serpentine tubes (Srp) and minor talc-like lamellae (Tlc); d) detail of a short serpentine tube
611 with a hollow core (red arrow), next to a long serpentine tube (yellow arrow); e) high resolution
612 image of the area in the rectangle in d) showing the typical basal spacings of serpentine minerals of

613 7.3 Å, and a dislocation marked with a white arrow; f) electron diffraction pattern of the serpentine
614 tubes. All images were obtained in a CM20 (CIC-UGR).

615 **Figure. 3.** Type II as seen under the TEM (specimen LC-100B): a) image of the sample, the
616 specimen was detached from the area in the red rectangle; b) detail of the hole in the grid under the
617 optical microscope (plane polarised light); c) low magnification image of the specimen showing
618 circular sections of serpentine (Srp) within a matrix of talc-like platelets (Tlc), including an electron
619 diffraction pattern with distinctive 7.2 Å spots and a less crystalline phase; d) high resolution image
620 of polygonal serpentine, showing the angle between sectors and the basal spacings; e) kidney-shaped
621 sectored serpentine; f) serpentine tubes with large holes filled with talc-like lamellae; g) oblique
622 section of serpentine tube, the central hole being altered to talc as in f), and curled up talc-like
623 lamellae aggregate (lower left). All images were obtained in a CM20 (CIC-UGR).

624 **Figure. 4.** Type III as seen under the TEM (specimen GAR-2.2): a) image of the sample, the
625 specimen was detached from the area in the red rectangle; b) detail of the hole in the grid under the
626 optical microscope (plane polarised light); c-e) high resolution images showing the presence of 7.2–
627 7.4 Å and 9.2–9.4 Å basal spacings; f) electron diffraction image showing a higher crystallinity 7 Å
628 spacing coexisting with a less crystalline material. All images were obtained in a Philips CM20
629 (CIC-UGR).

630 **Figure. 5.** Type IV (kerolite-pimelite) as seen under the TEM (specimen LC-100B): a) image of the
631 sample, the specimen was detached from the area in the red rectangle; b) detail of the hole in the grid
632 under the optical microscope (plane polarised light); c) low magnification images of the specimen
633 showing bundles of talc-like lamellae; d) high resolution images of talc-like lamellae; e) close-up of
634 image d) displaying the typical basal spacings of talc-like minerals; f) electron diffraction image. All
635 TEM images were obtained in a Philips CM20 (CIC-UGR).

636 **Figure 6.** Sepiolite-falcondoite (type V) as seen under the TEM (specimen GAR-7v): a) image of
637 the sample, the specimen was detached from area in the red rectangle, b) detail of the hole in the grid
638 under the optical microscope (plane polarised light), c) low magnification image of the specimen
639 showing sepiolite ribbons in a matrix of amorphous silica (obtained in a Jeol 2010, CCiT-UB), d)
640 detailed view of criss-cross sepiolite ribbons (in a Philips CM20, CIC-UGR), e) detail of 12 Å
641 fringes of a Ni-sepiolite ribbon oriented normal to the beam, in the matrix of amorphous silica
642 (Philips CM20, CIC-UGR).

643 **Figure 7.** Triangular diagram comparing AEM (circles, squares and triangles) with previous EMP
644 (fields). Compositional fields from Villanova-de-Benavent et al. (2014a). Abbreviations from
645 Whitney & Evans (2010) except falcondoite (Fal), kerolite (Ker), pimelite (Pim), pecoraite (Pec).

646 **Figure 8.** Replacement textures between Ni-phyllsilicates: a) sectored serpentine being replaced by
647 talc-like lamellae, with talc-like lamellae inside the serpentine core (b) and growing from the
648 serpentine edges (c) (sample LC-100AB-5); d) large 15-sectored polygonal serpentine, showing talc-
649 like lamellae inside the core and at the edges of sectors (e), and at the border (f) (sample LC-100AB-
650 5); g) polygonal serpentine being altered to talc-like lamellae; h) detail of (g) (sample 09GAR-2.1);
651 i-j) rolled talc-like lamellae recording textures of former serpentine particles (i: sample 09GAR-2.1;
652 j: LC-100AB-5). All these images were obtained in a Jeol 2010 (UniSi).

653 **Figure 9.** Replacement model which may explain the formation of Ni-enriched talc-like (“kerolite”-
654 “pimelite”) lamellae from the alteration of Ni-poor serpentine particles (1a-3a) and HRTEM images
655 representative of each precipitation stage (1b-3b): early formation of polygonal serpentine and
656 chrysotile tubes (1a, 1b), which are subsequently replaced by talc-like lamellae mainly in the higher
657 stress sites (inner cores, outer rims, borders between sectors) (2a, 2b), leading to a complete
658 replacement of serpentine particles by talc-like, leaving some rolled and curved talc-like lamellae as

659 remnants of the former serpentine features (3a, 3b). The HRTEM images were obtained in a Jeol
660 2010 (UniSi): sample LC-100AB-5 (1b), sample 09GAR-2.1 (2b), sample LC-100AB-4 (2c).
661

5

Type	Mineralogy (XRD)	Structural formula (EMP) [number of analyses averaged]	Xtal
TYPE I	serpentine + traces kerolite-pimelite (type I)	(Mg _{2.29} Ni _{0.25} Fe _{0.19})Si _{2.17} O _{5.53} (OH) _{3.27} ·0.19H ₂ O [23]	0.19
TYPE I	serpentine + traces kerolite-pimelite (type I)	(Mg _{2.34} Ni _{0.48} Fe _{0.15})Si _{2.48} O _{6.27} (OH) _{3.49} ·0.25H ₂ O [16]	0.25
TYPE II	serpentine + minor kerolite-pimelite (type II)	(Mg _{1.94} Ni _{1.02} Fe _{0.03})(Si _{2.56} Al _{0.04})O _{6.49} (OH) _{3.40} ·0.30H ₂ O [23]	0.30
TYPE III	serpentine + kerolite-pimelite (type III)	(Ni _{2.56} Mg _{0.43})Si _{2.97} O _{7.44} (OH) _{3.02} ·0.49H ₂ O [27]	0.49
TYPE III	serpentine + kerolite-pimelite (type III)	(Ni _{2.42} Mg _{0.56} Fe _{0.01})Si _{2.71} O _{6.81} (OH) _{3.28} ·0.36H ₂ O [4]	0.36
TYPE IV	kerolite-pimelite (type IV)	(Mg _{1.90} Ni _{1.09})Si _{3.89} O _{9.76} (OH) _{2.09} ·0.91H ₂ O [4]	0.91
TYPE IV	kerolite-pimelite (type IV)	(Mg _{1.72} Ni _{1.27} Fe _{0.01})Si _{3.84} O _{9.61} (OH) _{2.16} ·0.92H ₂ O [17]	0.92
TYPE IV	kerolite-pimelite (type IV)	(Ni _{1.95} Mg _{1.04})Si _{3.69} O _{9.24} (OH) _{2.30} ·0.85H ₂ O [109]	0.85
TYPE V	sepiolite-falcondoite (type V)	(Mg _{4.60-7.05} Ni _{1.01-2.39} Fe _{0-0.05})(Si _{11.75-12.75} Al _{0-0.02})O ₁₅ (OH) ₂ ·6H ₂ O [30]	-
TYPE V	sepiolite-falcondoite (type V)	(Mg _{3.42-5.15} Ni _{2.29-5.24} Fe _{0-0.02})(Si _{11.57-12.55} Al _{0-0.15})O ₁₅ (OH) ₂ ·6H ₂ O [4]	-
TYPE V	sepiolite-falcondoite (type V)	(Mg _{6.51-7.94} Ni _{0.34-0.58} Fe _{0.01-0.27})(Si _{11.46-12.32} Al _{0.01-0.03})O ₁₅ (OH) ₂ ·6H ₂ O [16]	-

	Type I (Srp)	Type I (Srp)	Type I (Tlc)	Type I (Tlc)	Type II (Srp)	Type II (Srp)	Type II (Srp)	Type II (Tlc)	Type II (Tlc)
	LC-101 serp 1-2	LC-101 serp 1-8	LC-101 serp 1-10	LC-101 serp 1-11	LC-100B- Serp-12	LC-100B- Serp-15	LC-100B serp 3b-5	LC-100B- 2 (SERP)- 4	LC-100B- 2 (SERP)- 14
S	14	14	22	22	14	14	14	22	22
	2.737	2.509	2.374	2.111	2.756	2.556	2.590	1.482	1.008
	0.000	0.000	0.000	0.000	0.266	0.094	0.000	0.000	0.054
	1.825	2.085	3.869	3.846	1.888	1.899	2.132	3.905	3.907
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.386	0.160	0.158	0.072	0.023	0.305	0.067	0.018	0.018
	0.035	0.080	0.651	1.091	0.035	0.047	0.045	1.681	2.070
	3.158	2.750	3.183	3.273	2.814	2.908	2.702	3.182	3.097
	1.825	2.085	3.869	3.846	2.154	1.993	2.132	3.905	3.961
	type III	type III	type III	type IV	type IV	type IV	type V	type V	type V
	GAR- 2.3B-3	GAR- 2.3B-7	GAR- 2.3B-8	LC-100B- 4 (TLC)-2	LC-100B- 4 (TLC)-5	LC-100B- 4 (TLC)- 12	GAR-7v- 3	GAR-7v- 6	GAR-7v- 7
S	22	22	22	22	22	22	32	32	32
	0.364	0.417	0.413	1.440	1.494	1.725	1.961	2.138	2.013
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	3.770	3.757	3.966	3.997	4.015	4.026	6.065	6.106	6.090
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	3.096	3.068	2.656	1.566	1.476	1.222	1.910	1.649	1.806
	3.460	3.485	3.069	3.007	2.971	2.948	3.871	3.787	3.819
	3.770	3.757	3.966	3.997	4.015	4.026	6.065	6.106	6.090

FIGURES

Figure 1.

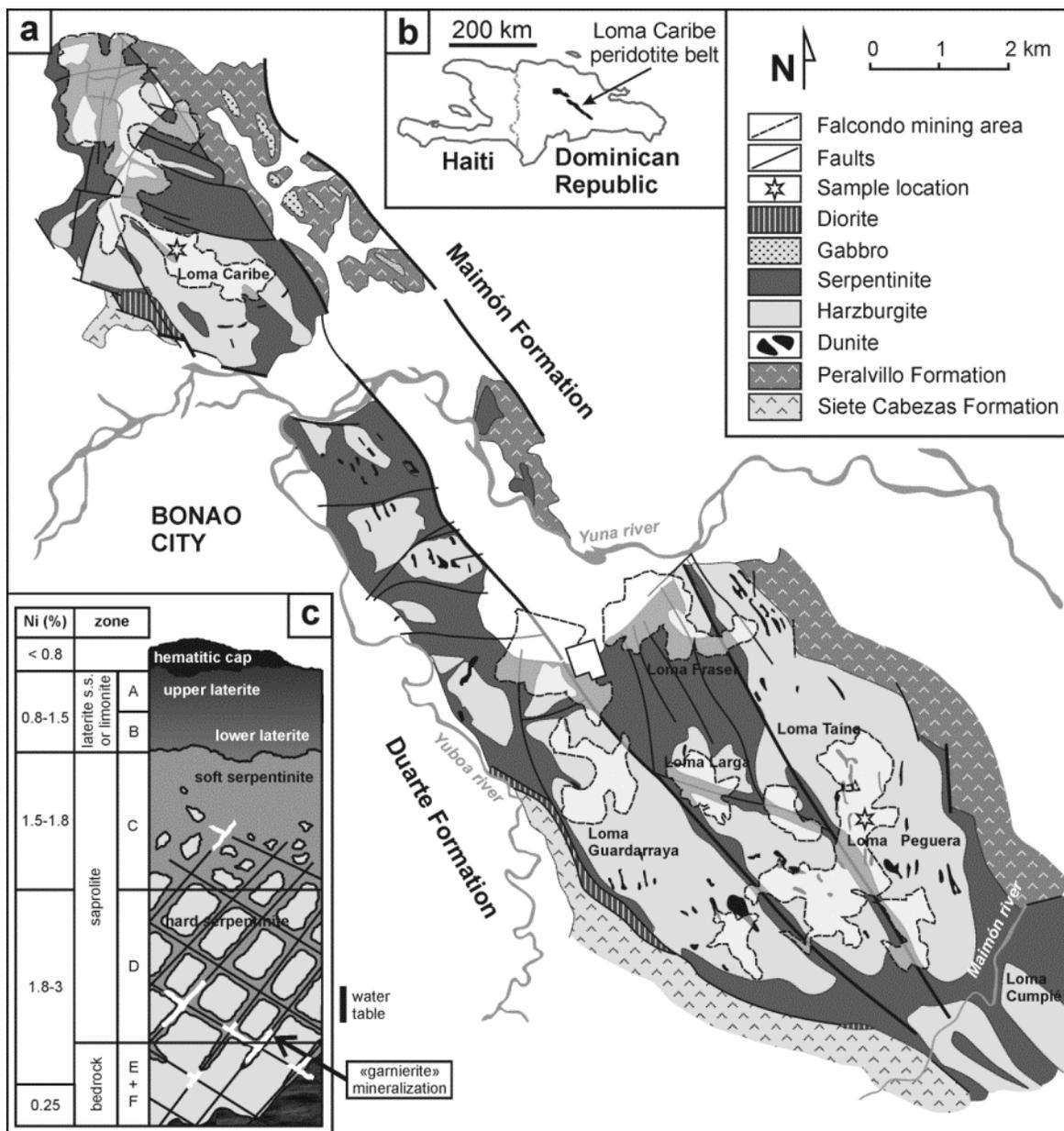


Figure 2.

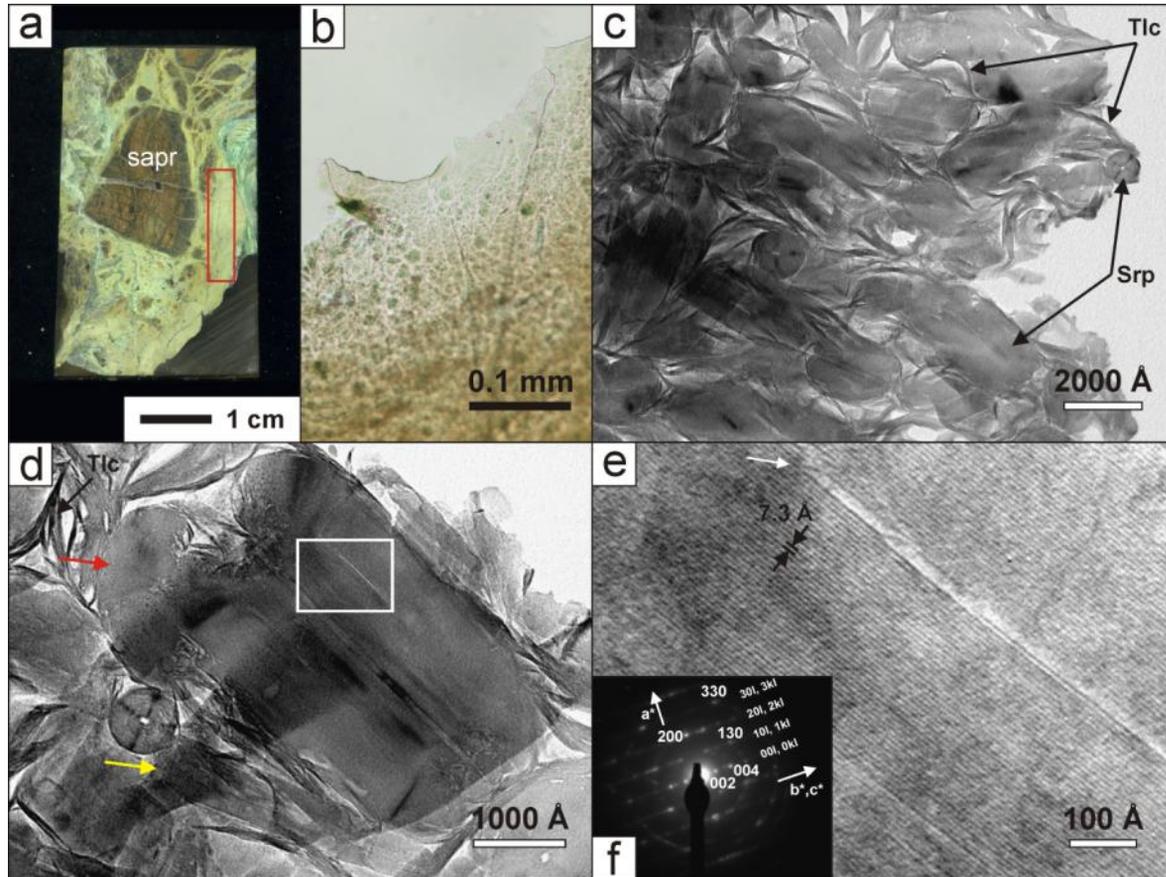


Figure 3.

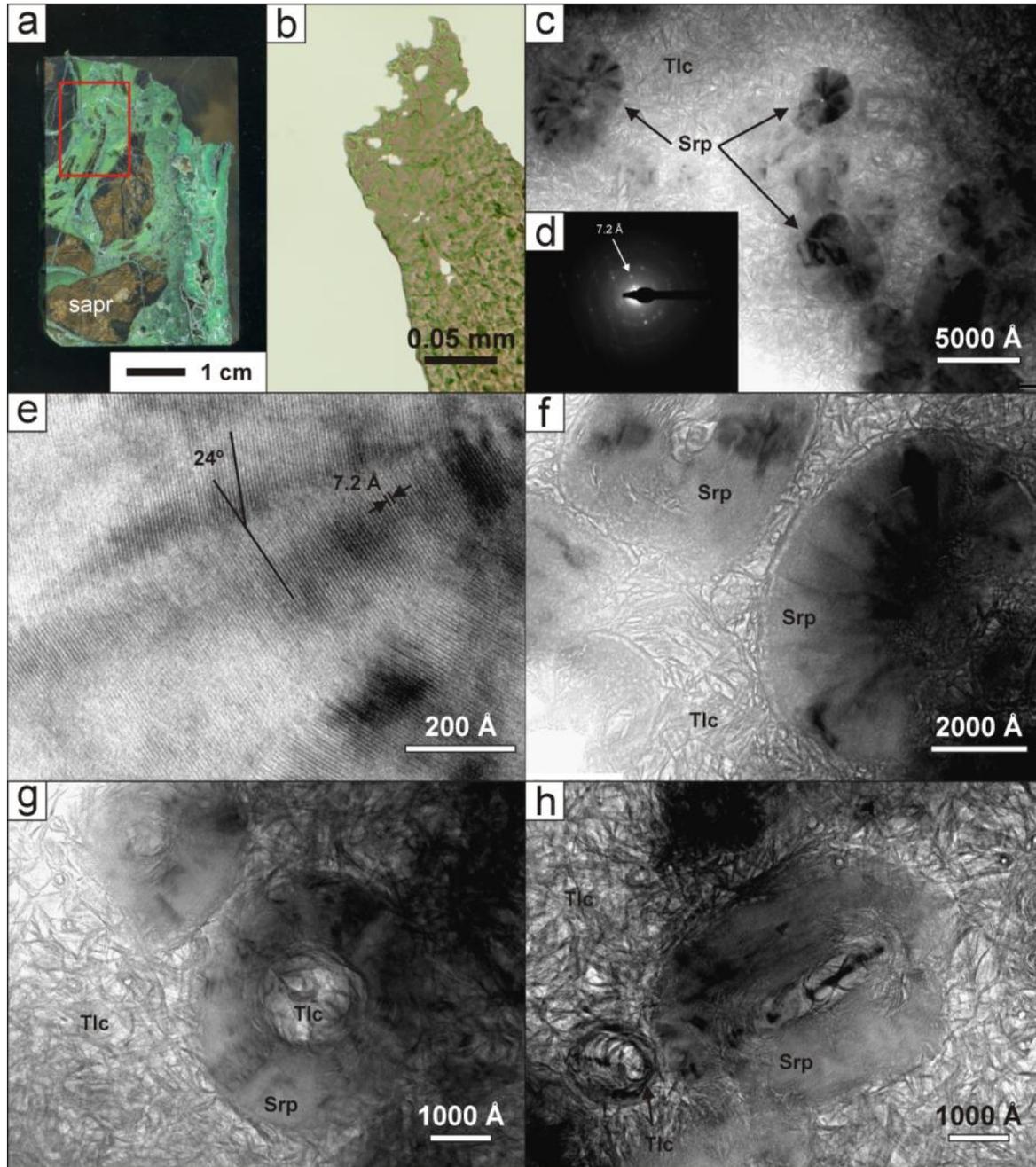


Figure 4.

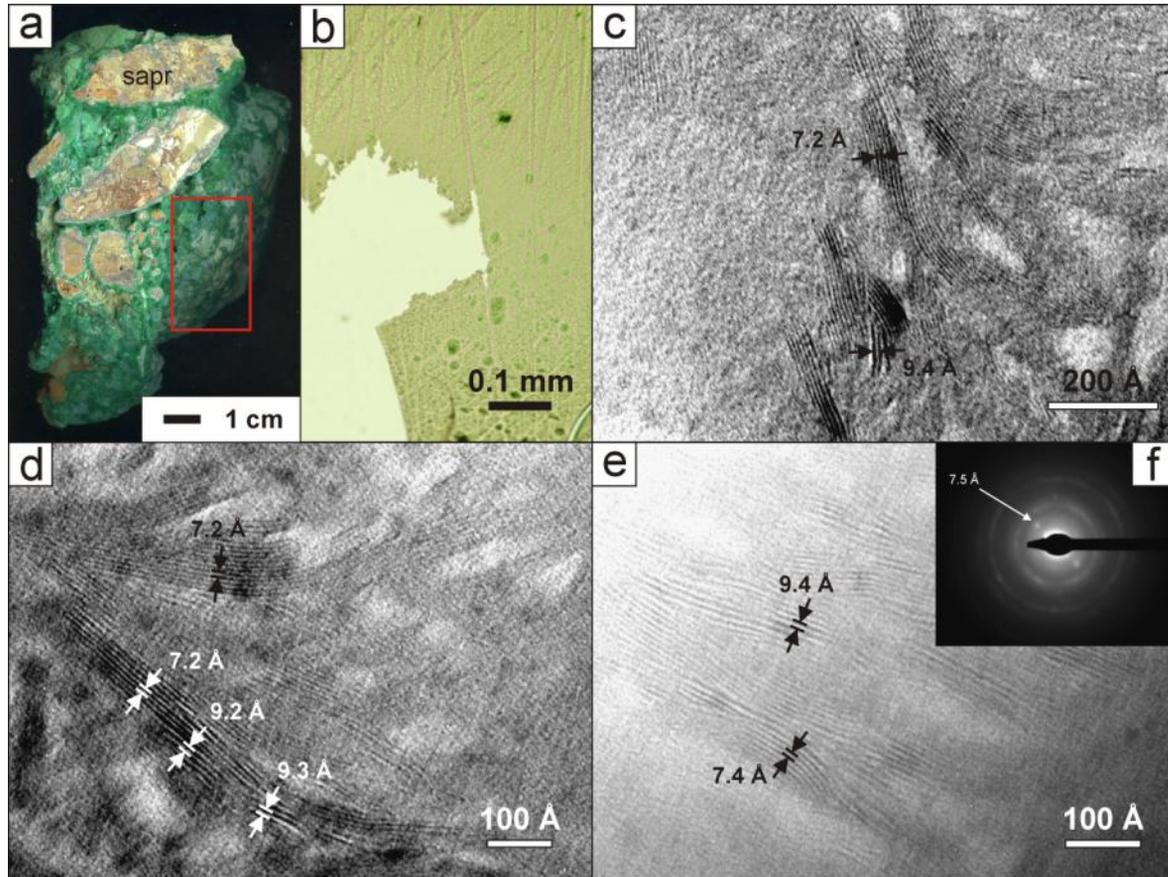


Figure 5.

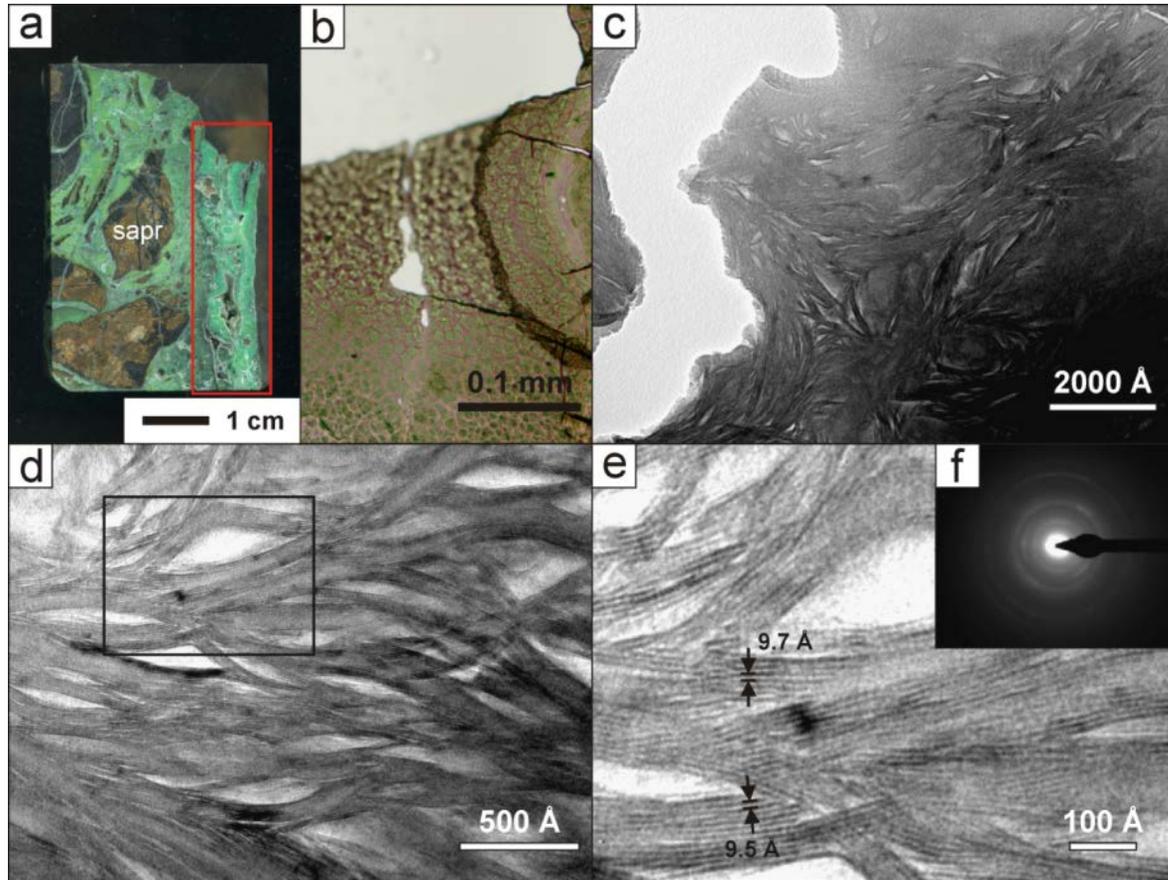


Figure 6.

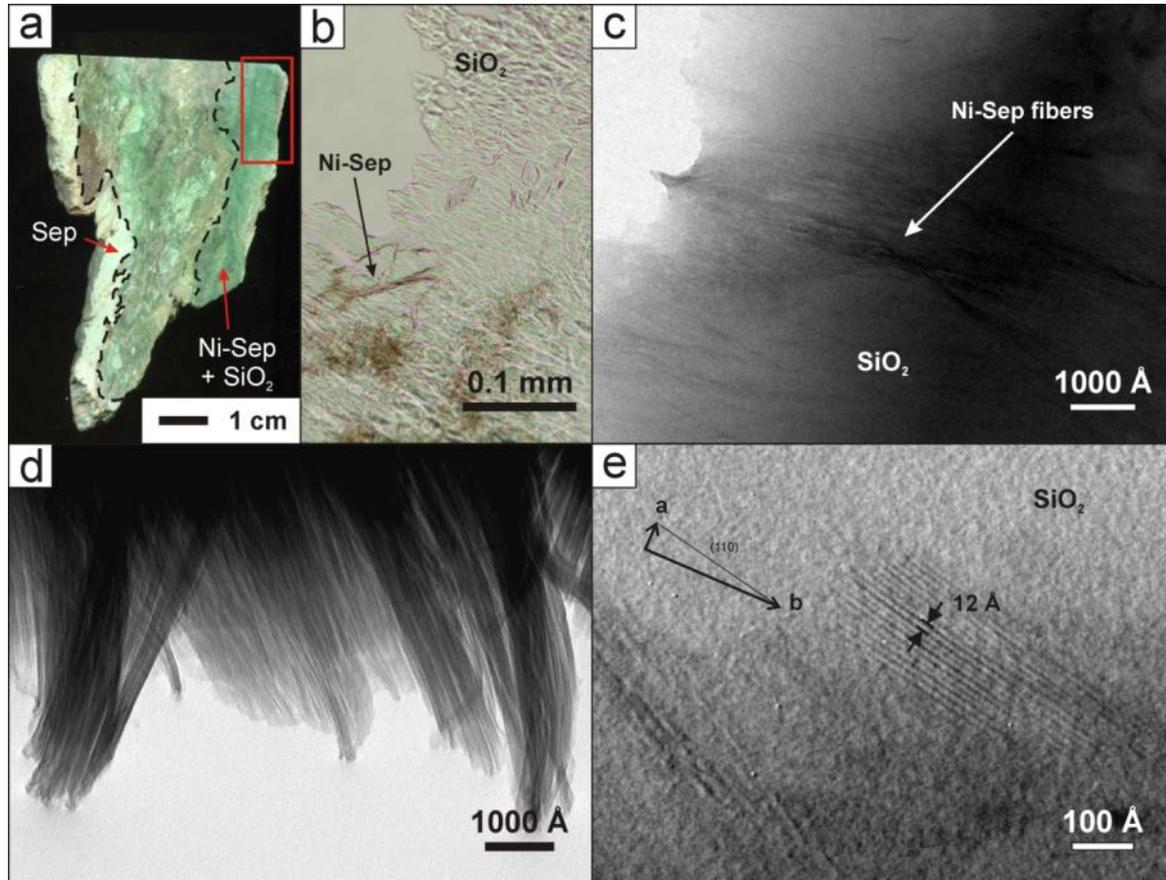


Figure 7.

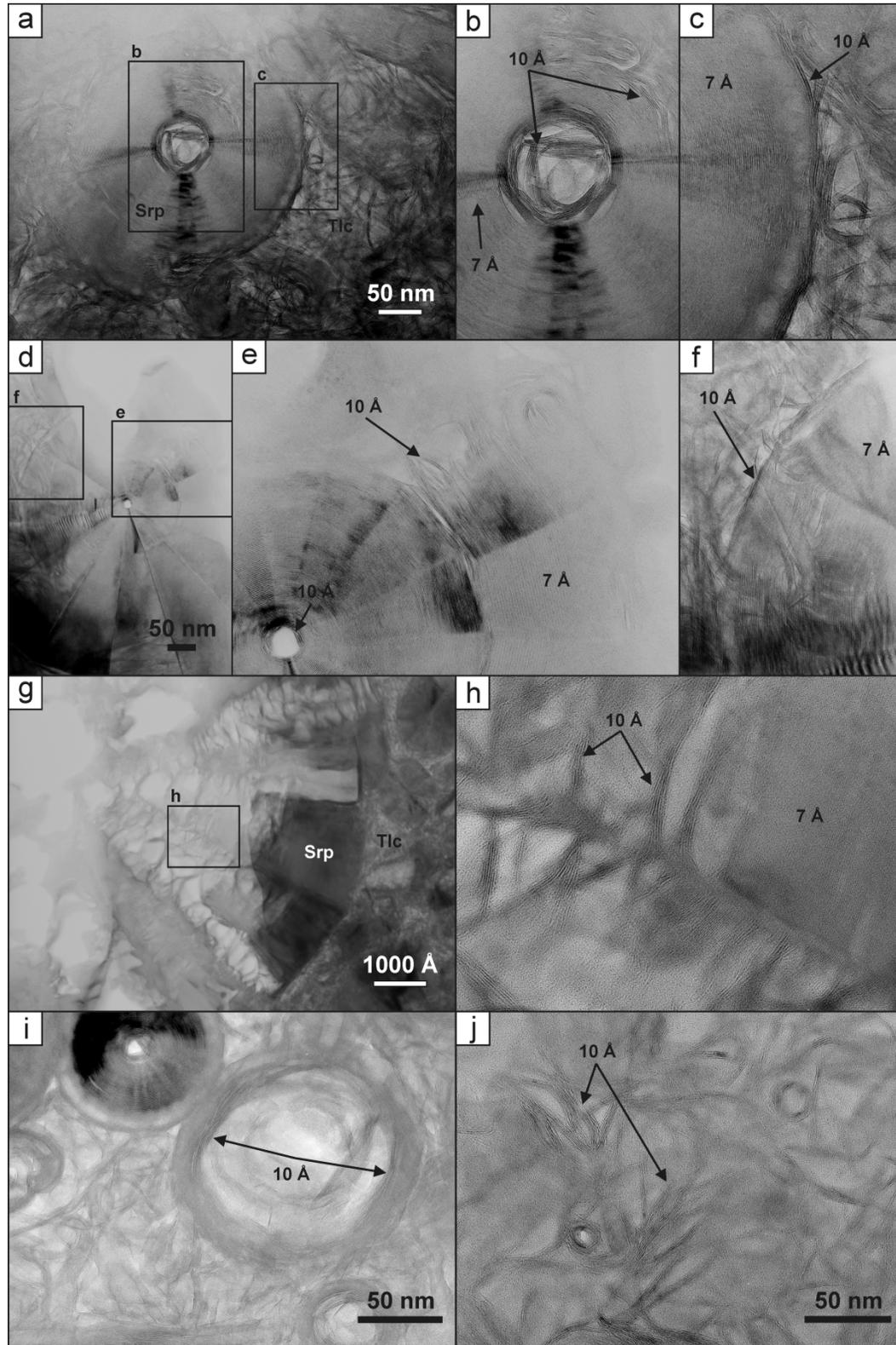


Figure 8.

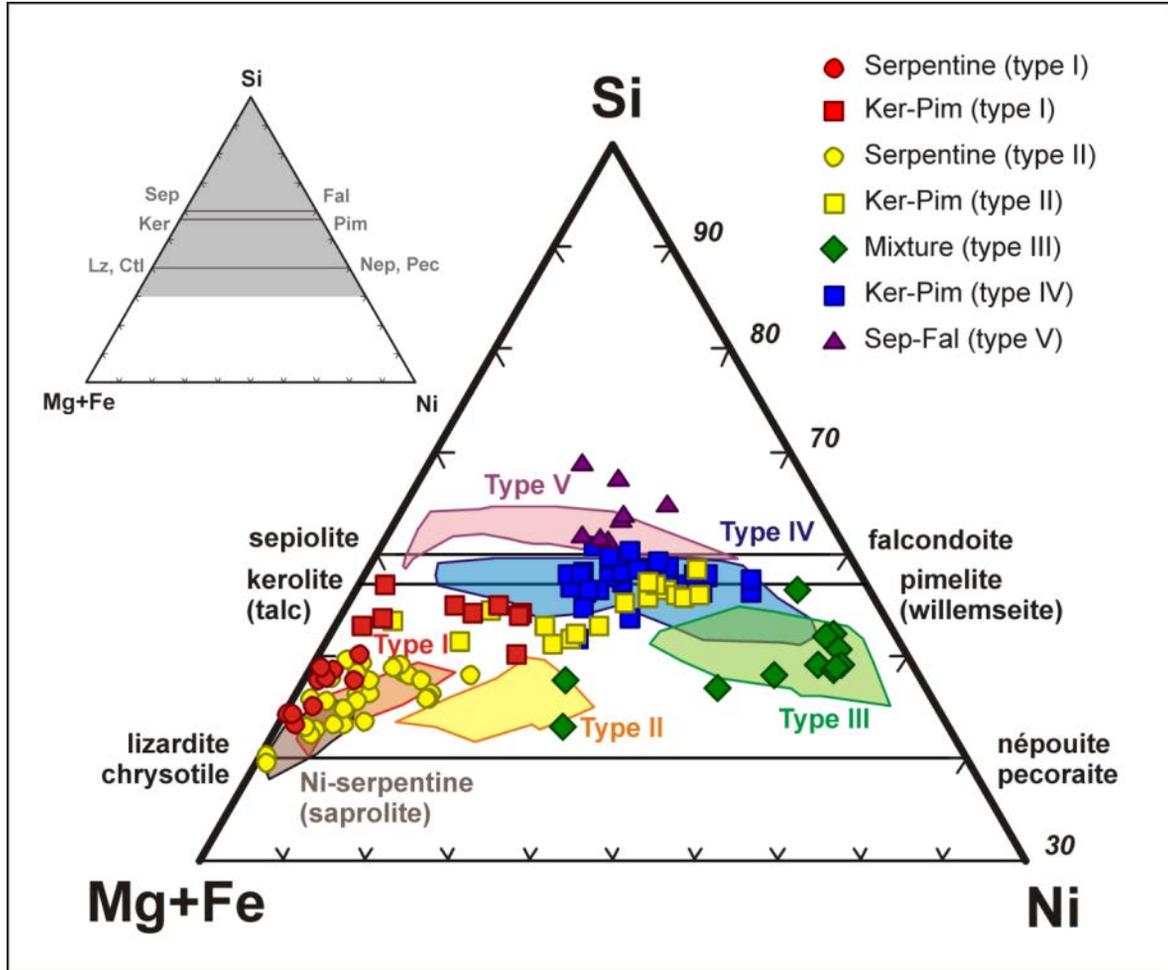


Figure 9.

