

1 **TITLE**

2 **Ni-phylosilicates (garnierites) from the Falcondo Ni-laterite deposit (Dominican Republic):**  
3 **mineralogy, nanotextures and formation mechanisms by HRTEM and AEM**

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5 **AUTHORS AND AFFILIATIONS**

6 Cristina Villanova-de-Benavent – [cvillanovadb@ub.edu](mailto:cvillanovadb@ub.edu)

7 *Departament de Cristal·lografia, Mineralogia i Dipòsits Minerals, Facultat de Geologia,*  
8 *Universitat de Barcelona (UB), Martí i Franquès s/n, 08028 Barcelona, Spain. +34 934021341*

9 Fernando Nieto – [nieto@gr.es](mailto:nieto@gr.es)

10 *Departamento de Mineralogía y Petrología and IACT, Universidad de Granada, CSIC, Av.*  
11 *Fuentenueva 18071 Granada, Spain. +34 609132940*

12 Cecilia Viti – [cecilia.viti@unisi.it](mailto:cecilia.viti@unisi.it)

13 *Dipartimento di Scienze Fisiche, della Terra e dell'Ambiente, Università degli Studi di Siena, Via*  
14 *Laterina 8, 53100 Siena, Italy. +39 0577233988*

15 Joaquín A. Proenza – [japroenza@ub.edu](mailto:japroenza@ub.edu)

16 *Departament de Cristal·lografia, Mineralogia i Dipòsits Minerals, Facultat de Geologia,*  
17 *Universitat de Barcelona (UB), Martí i Franquès s/n, 08028 Barcelona, Spain. +34 934021351*

18 Salvador Galí – [gali@ub.edu](mailto:gali@ub.edu)

19 *Departament de Cristal·lografia, Mineralogia i Dipòsits Minerals, Facultat de Geologia,*  
20 *Universitat de Barcelona (UB), Martí i Franquès s/n, 08028 Barcelona, Spain. +34 934021341*

21 Josep Roqué-Rosell – [jrrosell@lbl.gov](mailto:jrrosell@lbl.gov)

22 *Advanced Light Source, Lawrence Berkeley National Laboratory, One Cyclotron Road, MS*  
23 *15R0317 Berkeley, California 94720, USA. +1 510 486 7035*

24

25 **ABSTRACT**

26 Ni-bearing magnesium phyllosilicates (garnierites) are significant Ni ores in Ni-laterites worldwide.  
27 The present paper reports a detailed TEM investigation of garnierites from the Falcondo Ni-laterite  
28 deposit (Dominican Republic). Different types of garnierites have been recognized, usually  
29 consisting of mixtures between serpentine and talc-like phases which display a wide range of  
30 textures at the nanometer scale. In particular, chrysotile tubes, polygonal serpentine and lizardite  
31 lamellae are intergrown with less crystalline, talc-like lamellae. Samples consisting uniquely of talc-  
32 like and of sepiolite-falcondoite were also observed, occurring as distinctive thin lamellae and long  
33 ribbon-shaped fibers, respectively. HRTEM imaging indicates that serpentine is replaced by the talc-  
34 like phase, whereas TEM-AEM data show preferential concentration of Ni in the talc-like phase. We  
35 suggest, therefore, that the crystallization of Ni-bearing phyllosilicates is associated with an increase  
36 in the silica activity of the system, promoting the replacement of the Ni-poor serpentine by the Ni-  
37 enriched talc-like phase. These results have interesting implications in material science, as  
38 garnierites are natural analogues of Ni-bearing phyllosilicate-supported synthetic catalysts. Finally,  
39 SAED and HRTEM suggest that the Ni-bearing talc-like phase corresponds to a variety of talc with  
40 extra water, showing larger  $d_{001}$  than talc (i.e. 9.2–9.7 Å), described as “kerolite”-“pimelite” in clay  
41 mineral literature.

42

43 **KEYWORDS**

44 Ni-laterites; garnierites; chrysotile; polygonal serpentine; lizardite; “kerolite”-“pimelite”; sepiolite-  
45 falcondoite; HRTEM

46 **MANUSCRIPT**

47

## INTRODUCTION

48 *Garnierite* was originally the name of a mineral species discovered in 1863 in New Caledonia  
49 (Garnier 1867), although it was later proved to be a mixture of different Ni-hydrous silicates (e.g.  
50 Pecora et al. 1949, Faust 1966). Since then, the term garnierite has been widely used in the literature  
51 to refer to the group of green, fine-grained, poorly crystallized, Ni-bearing magnesium  
52 phyllosilicates, including serpentine, talc, sepiolite, smectite and chlorite, often occurring as  
53 mixtures (e.g. Faust 1966, Brindley and Hang 1973, Springer 1974, Brindley 1978, Wells et al.  
54 2009, Villanova-de-Benavent et al. 2014a). Therefore, despite not being a valid mineral name,  
55 *garnierite* became a convenient field term used by mine geologists to designate all green Ni-  
56 phyllosilicates when a more specific characterization was not possible (Brindley 1978), and many  
57 authors have used this term with this general meaning (e.g. Moraes, 1935, Pecora and Hobbs 1942,  
58 Pecora et al. 1949, Varela 1984, Gleeson et al. 2003, 2004, Freyssinet et al. 2005, Wells et al. 2009).  
59 In addition, the classification and naming of the garnierite minerals represents a complex, long-  
60 lasting controversy, because of their fine-grained nature, poor crystallinity and frequent occurrence  
61 as intimate mixtures of different mineral species (Brindley and Hang 1973). Brindley and co-  
62 authors, after various studies, distinguished the following Mg-Ni series: the serpentine group  
63 minerals i) lizardite-népouite and ii) chrysotile-pecoraite; iii) berthierine-brindleyite; the talc-like  
64 structures iv) talc-willemseite and v) “kerolite”-“pimelite”; vi) the chlorite series clinochlore-nimite;  
65 and finally vii) sepiolite-falcondoite (Brindley and Hang 1973, Brindley and Maksimović 1974,  
66 Brindley 1978, 1980). The most common garnierites found in nature are formed by lizardite-  
67 népouite and “kerolite”-“pimelite” (Brindley 1978), and many authors have been referred to them as  
68 “serpentine-like” (or “7 Å-type”) and “talc-like” (or “10 Å-type”) garnierites, respectively (e.g.  
69 Brindley and Hang 1973, Brindley and Maksimović 1974, Wells et al. 2009, Galí et al. 2012).

70 “Kerolite” and “pimelite” can be described as phases with talc affinity and extra water in their  
71 structure, within the ideal structural formula  $(\text{Mg,Ni})_3\text{Si}_4\text{O}_{10}(\text{OH})_2 \cdot n(\text{H}_2\text{O})$ . Faust (1966) classified  
72 “pimelite” into the smectite group, although other authors proved that neither “kerolite” nor  
73 “pimelite” exhibit intracrystalline swelling (e.g. Kato 1961, Brindley and Hang 1973, Brindley  
74 1978). Regardless being both discredited species by the Commission on New Minerals,  
75 Nomenclature and Classification of the International Mineralogical Association (CNMNC-IMA),  
76 these names have been used during the following decades (e.g. Gleeson et al. 2003, Freyssinet et al.  
77 2005, Tauler et al. 2009, Wells et al. 2009, Galí et al. 2012, Villanova-de-Benavent et al. 2014a,  
78 Cathelineau et al. 2015) and are considered as valid species by the International Association for the  
79 Study of Clays (AIPEA). According to Dosbaba and Novák (2012), “kerolite” represents a fine  
80 crystalline, poorly ordered, hydrated variant of talc, from which it can be distinguished by its very  
81 broad peak at approximately 10 Å; from smectite it can be distinguished by the absolute lack of  
82 swelling at an ethylene glycol atmosphere (Brindley and Brown 1980).

83 The remarkable Ni content of garnierites makes them significant Ni ores in some Ni-laterite  
84 deposits, such as in the Falcondo deposit, in the Dominican Republic (e.g. Golightly 1981, Elias  
85 2002, Gleeson et al. 2003, Freyssinet et al. 2005). In the Falcondo Ni-laterite, characterized by an  
86 Al-poor ultramafic protolith, five garnierite types were distinguished according to their color,  
87 mineralogy, textures and chemical composition (namely types I to V, Villanova-de-Benavent et al.  
88 2014a, b): I) Ni-Fe-bearing serpentine-like, II) Ni-bearing mixture of serpentine-like and “kerolite”-  
89 “pimelite”, III) Ni-dominant mixture of serpentine-like and “kerolite”-“pimelite”, IV) “kerolite”-  
90 “pimelite”, and V) sepiolite-falcondoite.

91 These previous results, obtained by means of X-ray powder diffraction (XRD), optical and scanning  
92 electron microscopy (SEM, with energy dispersive spectrometry, EDS) and electron microprobe  
93 (EMP), indicated that, in general, the garnierites from Falcondo are complex mixtures at the

94 nanometer scale (Villanova-de-Benavent et al. 2014a, b). Although the methods used are sufficient  
95 for a description of the phases from an economic point of view, a detailed study including textures at  
96 the nanometer scale is necessary, to unravel the characteristics of the various mixtures and to gain  
97 further insight into the distribution of Ni.

98 Transmission Electron Microscopy (TEM) is a powerful technique to study the mineralogy,  
99 chemistry, and textural features of phases which commonly occur as poorly crystalline, fine grained  
100 mixtures at the nanometer scale. It has been extensively used to study phyllosilicates and, in  
101 particular, this method allows distinction of the different minerals of the serpentine group (e.g.  
102 Baronnet and Devouard 2005, Andréani et al. 2008, Suárez et al. 2011, Mellini 2014).

103 Serpentine group minerals are 1:1 trioctahedral phyllosilicates that develop different structural forms  
104 because of a geometrical misfit between tetrahedral and octahedral sheets. The three most common  
105 varieties are i) lizardite planar lamellae, ii) chrysotile tubes and iii) antigorite modulated structures.

106 In addition, other microstructures may be found, such as polygonal serpentine, which develop  
107 sectorized rolled fibers, larger than chrysotile tubes, with 15 or 30 sectors separated by 24° and 12°,  
108 respectively (e.g. Baronnet and Devouard 2005, Andréani et al. 2008, Mellini 2014, and references  
109 therein). Up-to-date studies report that sectors are actually composed of lizardite (e.g. Mellini 2014).

110 However, a limited number of exhaustive TEM works on Ni-bearing Mg-phyllosilicates exists up to  
111 the present (Uyeda et al. 1973, Esson and Carlos 1978, Poncelet et al. 1979, Pelletier 1983, Soler et  
112 al. 2008, Tauler et al. 2009, Suárez et al. 2011), and high resolution imaging and electron diffraction  
113 studies are in general scarce. Furthermore, most of these publications are based on crushed and  
114 dispersed material onto TEM grids, omitting the textural information of the assemblages and the  
115 relationships between different phases.

116 This paper presents a textural and chemical characterization by TEM of the different garnierites  
117 described in the Falcondo Ni-laterite deposit. The aim of this work was to describe the textural

118 relationships between the different garnierite-forming minerals and to determine the distribution of  
119 Ni at the nanometer scale based on High Resolution TEM (HRTEM) and Analytical Electron  
120 Microscopy (AEM).

## 121 **GEOLOGICAL SETTING AND GARNIERITE OCCURRENCE**

122 Hydrous silicate Ni-laterites of the Falcondo deposit are developed on the Loma Caribe, an  
123 ophiolitic peridotite belt in the Cordillera Central in the Dominican Republic. This belt consists of an  
124 elongated body, 4–5 km wide and 95 km long, oriented NW-SE and bounded by major faults (Fig.  
125 1a, b). The Loma Caribe ultramafic rocks, formed in the upper mantle, are composed of  
126 harzburgites, dunites and lherzolites, partially altered to serpentinites during their emplacement to  
127 the present day tectonic position (Lewis et al. 2006, Proenza et al. 2007). Partially serpentinitised  
128 peridotites have been exposed to weathering and erosion since the early Miocene, when the  
129 laterisation process began (Lewis et al. 2006).

130 The Falcondo Ni-laterite profile is up to 60 m thick and is divided into two main zones or horizons:  
131 the limonite on the top and the saprolite at the bottom (Fig. 1c; Haldemann et al. 1979, Lithgow  
132 1993, Lewis et al. 2006). However, the contacts between subzones and their thicknesses vary  
133 vertically and laterally in all outcrops of the Falcondo deposit (Lewis et al. 2006, Aiglsperger et al.  
134 2014, Villanova-de-Benavent et al. 2014a).

135 Garnierites in the Falcondo Ni-laterite occur mainly as mm-cm vein infillings in fractures, as thin  
136 coatings on joints and along fault planes, and as clasts and/or cements in different kinds of breccias.  
137 They are usually found within the lowermost part of the saprolite horizon, but may also be located  
138 near the unweathered serpentinitized peridotite rocks, at the base of the lateritic profile, and in the  
139 upper saprolite horizon (Tauler et al., 2009, Villanova-de-Benavent et al. 2014a).

## 140 **MATERIALS AND METHODS**

141 Forty samples from the lower saprolite horizon containing different garnierite types were previously  
142 identified by X-ray powder diffraction (XRD), observed by optical and scanning electron  
143 microscopy (SEM-EDS), and analyzed by electron microprobe (EMP) at the Centres Científics i  
144 Tecnològics of the Universitat de Barcelona (CCiT-UB) (Villanova-de-Benavent et al. 2014a). From  
145 these forty samples, nine were selected as the most representative to be studied by transmission  
146 electron microscopy for the present work (Table 1). Three different methods of specimen preparation  
147 were combined, due to the difficulties posed by the friability of the material, and twelve grids were  
148 prepared with similar outcomes.

149 First, some samples were prepared as polished thin sections with Canada balsam. Representative  
150 areas containing garnierites to be studied by TEM were selected under the optical microscope.  
151 Copper grids were attached on the thin section including the areas of interest, and they were later  
152 detached, ion-thinned by a Gatan 600 ion mill and a Gatan PIPS 691 at the Centro de  
153 Instrumentación Científica in the Universidad de Granada (CIC-UGR) (Suárez et al., 2011), and by a  
154 Gatan 600 Duo Mill at the Dipartimento di Scienze Fisiche, della Terra e dell'Ambiente of the  
155 Università degli Studi di Siena (UniSi). Second, small fragments of other selected samples were  
156 carefully separated by hand-picking, polished manually up to ~30 micrometers thick and glued to a  
157 copper grid. The grids were ion-thinned by a Fischione 1010 Low Angle Ion Milling & Polishing  
158 System and a Gatan PIPS 691 at the CCiT-UB. It is worth noting that the aforementioned specimen  
159 preparation methods preserve the original texture of the mineral phases. Third, other samples were  
160 separated by hand-picking, ground in an agate mortar, suspended with ethanol and put on a copper  
161 grid in order to obtain additional, higher quality EDX analyses (e.g. López Munguira and Nieto  
162 2000, Abad et al. 2001). Both ion-milled samples and powders were carbon-coated prior to the TEM  
163 study.

164 The TEM study was performed by Philips CM20 (CIC-UGR) and a Jeol JEM2010 (UniSi), equipped  
165 with Energy Dispersive X-ray spectrometer detectors (EDS-Oxford Isis, respectively) and operating  
166 at 200 kV. Additional images and electron diffraction patterns were obtained using a Jeol JEM2100  
167 at 200 kV (CCiT-UB) as well.

168 Atoms per formula unit (apfu) were calculated from the atomic concentrations obtained by AEM and  
169 based on the theoretical number of positive charges of each mineral (14, 22 and 32 for serpentine,  
170 talc and sepiolite, respectively). Fe was calculated as Fe<sup>3+</sup>, following Villanova-de-Benavent et al.  
171 (2014a) and according to the Fe<sup>2+</sup>/Fe<sup>3+</sup> XANES maps obtained in equivalent samples (Roqué-Rosell  
172 et al. submitted).

## 173 RESULTS

174 TEM results are shown in Figures 2-7, including a photograph of the selected sample (Figs. 2a-6a), a  
175 photomicrograph of the ion thinned specimen (Figs. 2b-6b), and low magnification, high resolution  
176 images and electron diffraction patterns. A comparison between the chemical compositions of the  
177 minerals obtained with TEM-AEM and EMP is shown in Figure 8, and AEM analyses are presented  
178 in Table 2.

### 179 Mineralogy and textures at the nano scale

180 **Ni-bearing serpentine-dominant (type I).** Type I garnierite consists of serpentine particles and  
181 minor talc-like bent thin lamellae with very little porosity (Fig. 2c). Serpentine occurs as tubes of  
182 various sizes which are randomly oriented, as basal and longitudinal sections are observed in low  
183 magnification images. There are two groups of serpentine tubes: i) thick, short tubes up to 5000 Å  
184 long and about 3000 Å diameter, with 85 Å in diameter hollow cores; ii) less abundant, narrow, long  
185 tubes up to 3000 Å in length and around 750 Å in diameter, with ~50 Å hollow cores (Fig. 2d).  
186 These nanostructures together with the particle size indicate that this sample is formed mostly by  
187 polygonal serpentine in relatively short fibers, finely intermixed with talc-like lamellae. Uyeda et al.



188 (1973) also reported short, stubby tubes with central holes in serpentine-like garnierites from Brazil,  
189 with an average width of 740 Å (4.4 wt.% NiO; serpentine-talc mixture with a talc fraction of 0.13),  
190 with less abundant ill-defined platy fragments, which may resemble the talc-like particles in type I  
191 garnierite (Fig. 2c, d).

192 As seen in Fig. 2d, the short tubes display bending at the fiber tip. These features have also been  
193 observed in polygonal serpentines coexisting with polyhedral serpentines by Andréani et al. (2008).  
194 In the high resolution images of the short tubes longitudinal sections 7.3 Å spacings are observed, in  
195 some cases presenting dislocations (Fig. 2e). The 7.3 Å spacings were confirmed in both the electron  
196 diffraction patterns and the Fast Fourier Transform obtained for the same short tubes, which  
197 displayed few reflections of the  $h0l$  plane that can be indexed with the clinochrysotile- $2M_{c1}$  14 Å  
198 structure (Brindley and Brown 1980) (Fig. 2f). However, no high resolution images of the long tubes  
199 could be obtained due to electron beam damage. According to the total diameters and the measured  
200 spacings, the short, thick ones have approximately 200 T-O layers and the long, thin ones have 50 T-  
201 O layers.

202 **Ni-bearing mixture of serpentine- and talc-like particles (type II).** In the specimen of type II  
203 garnierite, serpentine particles, with characteristic 7.2 Å spacings, display various shapes and sizes  
204 and are found scattered in a matrix of talc-like bundles (Fig. 3c, d). Some serpentine basal sections  
205 have diameters around 5000 Å with tiny hollow cores, and are divided in 15 equidimensional sectors  
206 with an angle of 24° between adjacent (001) lattice planes, so-called polygonal serpentine (Fig. 3e).  
207 In some cases, polygonal sections are kidney-shaped, resembling the polyhedral serpentine of  
208 Andréani et al. (2008) (Fig. 3f). The inner diameter of some polygonal serpentines is similar to the  
209 outer diameter of the chrysotile basal sections in type I (Fig. 3g). Other serpentine particles display  
210 less rounded cylindrical fibers, probably corresponding to oblique sections of serpentine tubes,  
211 smaller than the polygonal fibers (up to 3000 Å in diameter), which are hollow-cored or contain

212 disordered layers in the centre, probably of a talc-like phase (Fig. 3g). Uyeda et al. (1973) also  
213 observed those features and mentioned that this material in the cores could be amorphous. Talc-like  
214 lamellae were also observed to concentrate in curved aggregates (Fig. 3h).

215 **Ni-dominant mixture of serpentine- and talc-like particles (type III).** Type III garnierites display  
216 long and bent lamellae with basal fringe spacings of 9.2–9.4 Å and 7.2–7.4 Å mixed in single  
217 particles (Fig. 4c-e), which are frequently parallel to each other (Fig. 4d-e). Lamellae of 7.2–7.4 Å  
218 spacings are curved, more regular and thicker (from six to sixteen 7.2–7.4 Å spacings) when  
219 compared to the 9.2–9.4 Å lamellae (four fringes mostly, and up to six) (Fig. 4c-e). This is supported  
220 by electron diffraction patterns, showing that the sample is composed of crystalline lizardite with a  
221 7.2 Å spacing, coexisting with a lower crystallinity talc-like phase (Fig. 4f). No polygonal serpentine  
222 was found and only one basal and one longitudinal section of chrysotile tubes were observed in the  
223 specimen. Similar features involving ~7 Å and ~10 Å structures were observed in a garnierite from  
224 Oregon (18.6 wt.% NiO,  $X_{\text{talc}} = 27\%$ ) by Uyeda et al. (1973). These values differ from those of the  
225 type III garnierite (29–50 wt.% NiO and average  $X_{\text{talc}}$  of 47 %, Villanova-de-Benavent et al.  
226 2014a). Garnierites from New Caledonia shown by Pelletier (1983) and references therein also  
227 displayed comparable features.

228 **Talc-like particles (type IV).** The specimen containing Ni-“kerolite”-“pimelite” garnierites is very  
229 homogeneous, as it consists uniquely of sets of superimposed and/or plaited, thin, apparent lamellae  
230 with two to six basal spacings of 9.5–9.7 Å (Fig. 5c, d, e), confirmed by the absence of ~7 Å fringes  
231 in this specimen. Identical textures were observed in 10 Å-type garnierites from Brazil by Esson and  
232 Carlos (1978), in which no other crystalline phases were detected either. These Brazilian garnierites  
233 under the optical microscope display botryoidal features and coexist with silica, like type IV  
234 garnierites (Villanova-de-Benavent et al. 2014a). The small particle size, together with the wide and

235 diffuse rings observed in the electron diffraction patterns (typical of a random polycrystalline  
236 pattern), suggest that the sample may have low crystallinity (Fig. 5f).

237 **Sepiolite-falcondoite (type V).** The obtained TEM bright field images demonstrate the presence of  
238 Ni-rich sepiolite roughly oriented in fine ribbons with lengths above several hundreds of nanometers  
239 (Fig. 6c, d) and frequently enclosed within an amorphous silica matrix (Fig. 6b, c, e). The high  
240 magnification image in Fig. 6e shows a normal section of the fiber with the well defined (110)  
241 crystalline planes. This image allows measuring the 12 Å spacing typical of sepiolite-falcondoite  
242 (Post et al. 2007, Tauler et al. 2009).

#### 243 **Mineral chemistry at the nanoscale**

244 The composition of garnierite-forming minerals obtained by AEM is represented in Fig. 7, compared  
245 to previous analyses obtained by EMP (Villanova-de-Benavent et al. 2014a), and in Table 2. The  
246 identification of the mineral species based on the lattice fringe spacing and that from the chemical  
247 composition were coherent. In addition, to increase the number of analysis of each mineral species,  
248 additional analyses were obtained on particles deposited on copper grids (e.g. López Munguira and  
249 Nieto 2000, Abad et al. 2001).

250 In general, Al is virtually absent in the Falcondo garnierites. Serpentine particles have low Ni, up to  
251 0.2 apfu in type I and 0.6 apfu in type II; and yield remarkable amounts of Fe, 0.1–0.4 apfu in type I  
252 and up to 0.3 in type II garnierites. Type III compositions are actually mixed analyses of lizardite  
253 lamellae and the talc-like phase, with Ni contents ranging from 2.7 to 3.3 apfu and Al and Fe below  
254 detection limits. Talc-like analyses from all the studied garnierite types yield highly variable Ni  
255 contents, between 1.2 to 2.3 apfu (1.7 on average), and Al and Fe are generally low (up to 0.13 apfu  
256 Al, 0.03 on average; and up to 0.04 apfu Fe, <0.01 on average). In addition, Ni content in talc-like  
257 particles increases gradually from type I to type IV garnierites (Fig. 7).

258 AEM results plot within or near the compositional fields previously obtained through EMP in the  
259 same samples (Villanova-de-Benavent et al. 2014a), although type III mixtures and some analyses of  
260 type II serpentines still deviate towards the “kerolite”-“pimelite” series. Even if in most cases it was  
261 impossible to obtain pure analysis of the constituent phases of the mixture due to the very small size  
262 of the individual packets (e.g. Fig. 4c, d, e), the AEM spot size of tens of nanometers versus the  
263 EMP spot size of one micrometer gives unique results of the preferential partitioning of Ni in talc-  
264 like particles rather than in the associated serpentines. In summary, the higher spatial resolution of  
265 AEM analyses allowed better discrimination than EMP analyses, with slight deviations from the  
266 average composition obtained by EMP toward the respective constituent end-members (Fig. 7).  
267 Finally, the EDS spectra in type V garnierite enabled distinction of the Ni-sepiolite ribbons from the  
268 surrounding matrix. The Ni-sepiolite fibers show a distinctive Ni and Mg content in addition to Si  
269 and O, and the presence of Si seems to be related not only with sepiolite but also with amorphous  
270 SiO<sub>2</sub> matrix. Ni contents varied from 1.2 to 2.2 apfu and Al and Fe were below detection.

## 271 DISCUSSION

### 272 **Nanotextural variability of garnierite from the Falcondo Ni-laterite**

273 As stated by Brindley (1978), most garnierites are a mixture of serpentine-like and talc-like phases,  
274 and previous TEM imaging studies reported the occurrence of some garnierites as intimate mixtures  
275 of 7.2–7.4 and 9.2–9.7 Å phases at the nanometer scale (e.g. Uyeda et al. 1973, Poncelet et al. 1979,  
276 Soler et al. 2008). Most garnierites from the Falcondo Ni-laterite deposit (Figs. 2-5) actually consist  
277 of mixtures of different relative proportions of serpentine and talc-like phases (“kerolite”-  
278 “pimelite”). TEM analysis distinguished a wide variety of textures and mineral species of the  
279 serpentine minerals in the different garnierite mixtures (chrysotile tubes in type I, polygonal  
280 serpentine in type II, lizardite lamellae in type III). In addition, TEM revealed the presence talc-like

281 particles in all the serpentine-bearing garnierites even when these had not been detected by XRD, as  
282 in type I, probably due to their small quantity and low crystallinity.

283 Another characteristic feature is that 7.2–7.4 Å fringes occurred in large numbers, whereas 9.2–9.7  
284 Å fringes occurred in smaller sets (as stated by Uyeda et al. 1973), suggesting a lower crystallinity of  
285 talc-like phases, which is coherent with their broad peaks, or absence of them, in powder X-ray  
286 diffraction (Villanova-de-Benavent et al. 2014a) and with the diffuse and weak ring-shaped electron  
287 diffraction patterns in this study. Kato (1961) also reported that talc-like garnierites from New  
288 Caledonia produced ring-shaped selected area electron diffraction (SAED) patterns whereas  
289 serpentine-like phases gave single crystal diffraction patterns with well-defined spots. Furthermore,  
290 Pelletier (1983) (and references therein) distinguished the electron diffraction pattern of the talc-  
291 willemseite, with a regular structure, from the “kerolite”-“pimelite”, showing concentric circles.  
292 Other studies reported unspecified disordered regions in the TEM photomicrographs as well (e.g.  
293 Uyeda et al. 1973, Brindley 1978, Esson and Carlos 1978).

294 Such low crystallinity of the talc-like phase could be related to specific conditions, such as very low  
295 temperature, high water availability and/or high growth rate during crystallization. The fact that the  
296 10 Å fringes are less regularly defined than the 7 Å ones, could be also explained by a variable  
297 degree of hydration of their interlayers (Brindley and Hang 1973, Uyeda et al. 1973) and/or possible  
298 volatilization phenomena under the TEM vacuum.

299 Finally, sepiolite-falcondoite occurs as an independent phase, never mixed with serpentine nor talc-  
300 like garnierites. It displays the characteristic elongated ribbon shape of sepiolite, commonly related  
301 to amorphous silica and/or quartz, but has a remarkably higher Ni content (26.8 wt.% NiO) when  
302 compared to sepiolites examined under TEM in other localities (maximum 3.3 wt.% NiO in  
303 Indonesia; Kuhnel et al. 1978).

304 **Preferential Ni concentration in the talc-like structure**

305 According to AEM, talc-like phases yield higher Ni concentrations than serpentine, which is  
306 coherent with the good correlation between Ni content and the talc fraction in the garnierites from  
307 Falcondo, as suggested by Galí et al. (2012) and Villanova-de-Benavent et al. (2014a), and by Soler  
308 et al. (2008), in the Loma de Hierro Ni-laterite (Venezuela). The higher resolution of the AEM with  
309 respect to EMP enabled distinction of the serpentine particles from the “kerolite”-“pimelite”  
310 lamellae (Fig. 7), despite it being difficult to obtain pure, single-phase analyses of talc-like or  
311 serpentine in type III. Therefore, the distribution of Ni between serpentine and talc-like phases in  
312 type III could not be well established because of its finer particle size.

313 Despite Uyeda et al. (1973) stating that there was no correlation between Ni content and morphology  
314 of the particles under TEM, and between proportions of platy and elongated particles and the number  
315 of serpentine-like and talc-like layers (equivalent to serpentine or talc fraction, respectively), other  
316 studies reported that Ni is mostly concentrated in talc-like phases (e.g. Esson and Carlos 1978,  
317 Poncelet et al. 1979). Besides, Vitovskaya and Berkhin (1968) also showed some low-magnification  
318 and electron diffraction patterns of what they described as Mg- and Ni-bearing “kerolites” (actually  
319 garnierite mixtures). They identified typical, tiny clinochrysotile tubes in Mg-“kerolite” (consisting  
320 predominantly of serpentine with Ni below detection limit), and scales with tiny tubes of  
321 clinochrysotile in the Ni-“kerolite” (mostly a 10 Å mineral with 11.3 wt.% NiO). Consequently in  
322 this case Ni was also related to a 10 Å phase. Furthermore, Poncelet et al. (1979) showed by EMP  
323 analyses and by heating experiments that most of the nickel was concentrated in the octahedral layer  
324 of the 10 Å phase, and not homogeneously distributed in the octahedral layer of both the 7 Å and 10  
325 Å. During these experiments, metallic Ni particles were deposited onto the 10 Å flakes and rarely  
326 onto the serpentine fibers after heating the sample. In addition to these observations and  
327 interpretations, our AEM data confirm that the Ni was preferably contained in the talc-like phase.  
328 Possibly, the TOT talc structure is more likely to host Ni than the TO serpentine structure, as

329 demonstrated by the equilibrium constants of the simultaneous precipitation of Ni-serpentine and  
330 “kerolite”-“pimelite” (Galí et al. 2012).

331 AEM results of garnierites from the Falcondo Ni-laterite showed that Ni concentration in serpentine  
332 is lower than in the talc-like particles, and always below 50% of the total elements in octahedral  
333 coordination. Therefore, neither népouite nor pecoraite, the Ni analogues of lizardite and chrysotile  
334 respectively, are found in the garnierites from the Falcondo Ni-laterite deposit. However, the  
335 presence of népouite and pecoraite is widely reported in other localities. In particular, népouite has  
336 been extensively studied in New Caledonia (e.g. Brindley and Wan, 1975; Wells et al. 2009 and  
337 references therein). One possible explanation for the different mineralogy between New Caledonian  
338 and Dominican Ni-laterites is the lithology of the primary ultramafic rocks. In New Caledonia, the  
339 protolith is mainly harzburgite and dunite (e.g. Pelletier, 1983, 1996), whereas in the Dominican  
340 Republic the protolith is mostly clinopyroxene-rich harzburgite and lherzolite, commonly intruded  
341 by microgabbro and dolerite dykes (e.g. Marchesi et al., 2012). The greater pyroxene content may  
342 imply a greater availability of silica in the Falcondo Ni-laterite deposit, leading to the preferential  
343 formation of talc rather than serpentine during weathering.

344 When examined under TEM, népouite usually occurs as highly crystalline plates (34.8 wt.% NiO,  
345 Montoya and Baur 1963, 47.6 wt.% NiO, Manceau and Calas 1985), whereas pecoraite develops  
346 coils and spiral shapes with three to five revolutions about the spiral axis (Faust et al. 1969, 1973,  
347 Milton et al. 1983). In contrast, the serpentine particles in Falcondo present larger diameters and  
348 lengths than the Ni-dominant counterparts described in the literature. Roy and Roy (1954) and  
349 Milton et al. (1983) suggested that the substitution of Mg by Ni in chrysotile gives rise to non-  
350 tubular but also not well formed, platy crystals. The Ni substitution probably prevents the tubes to  
351 grow up to several hundreds of Å and to develop well rounded spirals or concentric tubes.  
352 Accordingly, despite that the serpentine particles in type III were not analysed by AEM due to their

353 small particle size, the possibility of these serpentine lamellae yielding higher Ni contents than the  
354 chrysotile tubes and polygonal serpentine in types I and II cannot be discarded.

### 355 **The formation mechanism of garnierites in the Falcondo Ni-laterite**

356 Low magnification and high resolution images are fundamental to understand the genetic  
357 relationships between serpentine (in most cases, polygonal fibers) and Ni-rich, talc-like phases. The  
358 talc-like lamellae are observed inside the serpentine central hole (Fig. 8a, b, e), surrounding the  
359 serpentines, forming at the very edges of the particles (Fig. 8a, c, d, f) and at the boundaries between  
360 adjacent sectors of the largest polygonal serpentines (Fig. 8e, f). In addition, HRTEM details indicate  
361 that, at the reaction front, the (001) talc-like planes are parallel to the 001 serpentine ones. This  
362 textural evidence suggests that talc-like phases form after serpentine, taking advantage of high stress  
363 sites, such as the outer rims (where the basal planes of serpentine are bent), the fiber cores (highest  
364 layer curvature), and the intersector boundaries. When replacement of serpentine by talc is more  
365 advanced, sectors become fully pseudomorphed by talc-like lamellae (Fig. 8g, h) and talc-like  
366 lamellae may develop rounded aggregates, mimicking the rounded shapes of former serpentine  
367 particles (Fig. 8i, j). It is worth noting that these poorly crystalline, talc-like layers are bent and  
368 characterized by extremely wide interlayer partings, giving rise to a highly porous nanotexture (Fig.  
369 5d, e; 8h).

370 The overall data indicates that the formation of the successive phases in the garnierite mixtures may  
371 be explained by an early formation of serpentine tubes. The diameter of the serpentine cores in type  
372 II is comparable to the outer diameter of narrow chrysotile tubes in type I. This may indicate that the  
373 diameter of the chrysotile increases in size until its curled structure is unstable. Then it is  
374 transformed into polygonal serpentine while preserving a cylindrical core, which may more likely be  
375 altered by later dissolution-precipitation than polygonal sectors. The serpentine particles are  
376 subsequently replaced by “kerolite”-“pimelite”, starting from high stress sites and structural



377 discontinuities, such as cores, rims and contact between sectors. The final product is possibly a  
378 garnierite formed exclusively by talc-like lamellae (Fig. 9).

379 The later formation of “kerolite”-“pimelite” from serpentine particles is in accordance to the  
380 garnierite precipitation model proposed by Galí et al. (2012). This model is based on the assumption  
381 that in an Al-free system, such as the Falcondo laterite profile, the stability of serpentine, kerolite–  
382 pimelite or sepiolite–falcondoite is mainly controlled by the silica activity. As a result, the ideal  
383 formation of the Ni ore occurs as a successive precipitation of mineral phases progressively enriched  
384 in Ni and Si, because silica activity increases with time and through the profile. Thus, the first  
385 garnierite-forming phase to precipitate is serpentine, then followed by “kerolite”-“pimelite”, and  
386 eventually by sepiolite–falcondoite and sepiolite–falcondoite with amorphous silica and/or quartz  
387 (Galí et al. 2012).

388

## CONCLUSIONS

389 ~~This work provides low and high resolution TEM images, showing that the five types of garnierites~~  
390 ~~present characteristic features at the nanometer scale. Most consist of an ultrafine mixture of~~  
391 ~~serpentine and a talc-like phases, but nanotextures and the relative amounts of serpentine and talc-~~  
392 ~~like particles are variable from sample to sample.~~

393 ~~Serpentine occurred as chrysotile tubes, polygonal serpentine and lizardite lamellae, and in general~~  
394 ~~was more crystalline than the associated talc-like particles. We note that all garnierite samples~~  
395 ~~contained at least a small portion of talc-like lamellae, even when this phase was not detected by~~  
396 ~~XRD, due to its low crystallinity and/or low abundance. These particles, which could be the only~~  
397 ~~component of garnierites, are in fact minerals of the “kerolite” “pimelite” series, a talc-like phase~~  
398 ~~with extra water in their structure and showing an expansion of  $d_{001}$ -spacing from 9.2 (talc) to 9.7 Å.~~  
399 ~~AEM data confirmed that Ni was mostly concentrated in “kerolite”-“pimelite” instead of in~~  
400 ~~serpentine particles. Low magnification and HRTEM images established an insight of the~~

401 ~~mechanisms by which Falcondo garnierites form, showing the preferential replacement of serpentine~~  
402 ~~by “kerolite”-“pimelite” at high stress sites in the serpentine structure, such as the inner walls of~~  
403 ~~cores, outer rims and intersector boundaries.~~

#### 404 **IMPLICATIONS**

405 The results obtained in this comprehensive TEM study on the garnierites from the Falcondo Ni-  
406 laterite deposit (Dominican Republic) reveal that the Ni-bearing serpentine, “kerolite”-“pimelite”  
407 and sepiolite-falcondoite components display a wide variety of textures and sizes at the nanometer  
408 scale. It is worth noting that most of the observed textures in Mg-serpentine (e.g. lamellae,  
409 chrysotile tubes, polygonal) are also found in their Ni-bearing analogues. However, the studied  
410 samples are characterized by low Ni contents in serpentines, whereas Ni is concentrated in the talc-  
411 like phases, reinforcing previous works (Villanova-de-Benavent et al., 2014). The results altogether  
412 provide further insight on the mineralogy of Ni-phyllsilicates as the highest grade Ni ores in a  
413 world-class hydrous silicate type Ni-laterite deposit.

414 The most noteworthy result of this work is the first evidence of replacement at the nanometer scale  
415 among garnierite-forming minerals. We have demonstrated that the Ni-enriched talc-like lamellae  
416 replace the polygonal serpentine and chrysotile tubes in the higher stress sites. In this model the Ni-  
417 enriched talc finally replaces the serpentine completely leaving some rolled and curved talc lamellae  
418 as remnants of the former serpentine (Fig. 9). This provides the direct proof that the crystallization of  
419 the Ni-bearing phyllosilicates at low temperature is associated with an increase in the silica activity  
420 of the system, which is coherent with previous observations at the micrometer scale (Villanova-de-  
421 Benavent et al., 2014) and the thermodynamic model for garnierites that predicted that Ni is  
422 preferably contained within the talc-like phase (Galí et al., 2012). As a consequence, Ni-enriched  
423 talc phases precipitate from already formed secondary serpentine particles. In the long term, in a  
424 more advanced stage of weathering, this will lead to the complete dissolution of serpentine particles

425 that will be entirely substituted by Ni-enriched talc-like and even sepiolite-like phases as the silica  
426 and Ni activity increase. In addition it is worth noting that this Ni-enrichment process leads to the  
427 formation of remarkable porosity in the material.

428 Therefore, the results of this work contribute to the knowledge of the formation of Ni-phyllsilicates  
429 under tropical conditions demonstrating that the Ni-poor serpentine particles are replaced by Ni-rich  
430 talc-like phases. This in turn has interesting implications in the field of material sciences. Recently  
431 (Sivaiah et al., 2011), synthesized Ni-containing serpentine-like and talc-like phyllosilicates were  
432 used as catalyst precursors for processing greenhouse gases (i.e. CO<sub>2</sub>, CH<sub>4</sub>). Thus the TEM  
433 characterization presented in this work also suggests that natural garnierites could be a good  
434 candidate to accomplish the requirements of a suitable catalyst precursor.

435

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448

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## 596 TABLE AND FIGURE CAPTIONS

597

598 **Table 1.** Summary of the studied garnierite samples from the Falcondo Ni-laterite (Dominican  
599 Republic). The talc fraction (X<sub>talc</sub>) in the serpentine-talc mixtures is calculated according to  
600 Brindley & Hang (1973).

601 **Table 2.** Representative normalized AEM analyses performed on Falcondo garnierites in atoms per  
602 formula unit (apfu).

603

604 **Figure 1.** Geological map of the Falcondo Ni-laterite (a) after Haldemann et al. (1979), location of  
605 the deposit in the Dominican Republic (b) and schematic profile of the Falcondo Ni-laterite (c)  
606 modified from Lithgow et al. (1979), Lewis et al. (2006), Villanova-de-Benavent et al. (2014a).

607 **Figure 2.** Type I garnierite as seen under the TEM (specimen LC-101): a) image of the sample, the  
608 specimen was detached from the area in the red rectangle; b) detail of the hole in the grid under the  
609 optical microscope (plane polarised light); c) low magnification images of the specimen showing  
610 short serpentine tubes (Srp) and minor talc-like lamellae (Tlc); d) detail of a short serpentine tube  
611 with a hollow core (red arrow), next to a long serpentine tube (yellow arrow); e) high resolution  
612 image of the area in the rectangle in d) showing the typical basal spacings of serpentine minerals of

613 7.3 Å, and a dislocation marked with a white arrow; f) electron diffraction pattern of the serpentine  
614 tubes. All images were obtained in a CM20 (CIC-UGR).

615 **Figure. 3.** Type II as seen under the TEM (specimen LC-100B): a) image of the sample, the  
616 specimen was detached from the area in the red rectangle; b) detail of the hole in the grid under the  
617 optical microscope (plane polarised light); c) low magnification image of the specimen showing  
618 circular sections of serpentine (Srp) within a matrix of talc-like platelets (Tlc), including an electron  
619 diffraction pattern with distinctive 7.2 Å spots and a less crystalline phase; d) high resolution image  
620 of polygonal serpentine, showing the angle between sectors and the basal spacings; e) kidney-shaped  
621 sectored serpentine; f) serpentine tubes with large holes filled with talc-like lamellae; g) oblique  
622 section of serpentine tube, the central hole being altered to talc as in f), and curled up talc-like  
623 lamellae aggregate (lower left). All images were obtained in a CM20 (CIC-UGR).

624 **Figure. 4.** Type III as seen under the TEM (specimen GAR-2.2): a) image of the sample, the  
625 specimen was detached from the area in the red rectangle; b) detail of the hole in the grid under the  
626 optical microscope (plane polarised light); c-e) high resolution images showing the presence of 7.2–  
627 7.4 Å and 9.2–9.4 Å basal spacings; f) electron diffraction image showing a higher crystallinity 7 Å  
628 spacing coexisting with a less crystalline material. All images were obtained in a Philips CM20  
629 (CIC-UGR).

630 **Figure. 5.** Type IV (kerolite-pimelite) as seen under the TEM (specimen LC-100B): a) image of the  
631 sample, the specimen was detached from the area in the red rectangle; b) detail of the hole in the grid  
632 under the optical microscope (plane polarised light); c) low magnification images of the specimen  
633 showing bundles of talc-like lamellae; d) high resolution images of talc-like lamellae; e) close-up of  
634 image d) displaying the typical basal spacings of talc-like minerals; f) electron diffraction image. All  
635 TEM images were obtained in a Philips CM20 (CIC-UGR).

636 **Figure 6.** Sepiolite-falcondoite (type V) as seen under the TEM (specimen GAR-7v): a) image of  
637 the sample, the specimen was detached from area in the red rectangle, b) detail of the hole in the grid  
638 under the optical microscope (plane polarised light), c) low magnification image of the specimen  
639 showing sepiolite ribbons in a matrix of amorphous silica (obtained in a Jeol 2010, CCiT-UB), d)  
640 detailed view of criss-cross sepiolite ribbons (in a Philips CM20, CIC-UGR), e) detail of 12 Å  
641 fringes of a Ni-sepiolite ribbon oriented normal to the beam, in the matrix of amorphous silica  
642 (Philips CM20, CIC-UGR).

643 **Figure 7.** Triangular diagram comparing AEM (circles, squares and triangles) with previous EMP  
644 (fields). Compositional fields from Villanova-de-Benavent et al. (2014a). Abbreviations from  
645 Whitney & Evans (2010) except falcondoite (Fal), kerolite (Ker), pimelite (Pim), pecoraite (Pec).

646 **Figure 8.** Replacement textures between Ni-phyllsilicates: a) sectored serpentine being replaced by  
647 talc-like lamellae, with talc-like lamellae inside the serpentine core (b) and growing from the  
648 serpentine edges (c) (sample LC-100AB-5); d) large 15-sectored polygonal serpentine, showing talc-  
649 like lamellae inside the core and at the edges of sectors (e), and at the border (f) (sample LC-100AB-  
650 5); g) polygonal serpentine being altered to talc-like lamellae; h) detail of (g) (sample 09GAR-2.1);  
651 i-j) rolled talc-like lamellae recording textures of former serpentine particles (i: sample 09GAR-2.1;  
652 j: LC-100AB-5). All these images were obtained in a Jeol 2010 (UniSi).

653 **Figure 9.** Replacement model which may explain the formation of Ni-enriched talc-like (“kerolite”-  
654 “pimelite”) lamellae from the alteration of Ni-poor serpentine particles (1a-3a) and HRTEM images  
655 representative of each precipitation stage (1b-3b): early formation of polygonal serpentine and  
656 chrysotile tubes (1a, 1b), which are subsequently replaced by talc-like lamellae mainly in the higher  
657 stress sites (inner cores, outer rims, borders between sectors) (2a, 2b), leading to a complete  
658 replacement of serpentine particles by talc-like, leaving some rolled and curved talc-like lamellae as

659 remnants of the former serpentine features (3a, 3b). The HRTEM images were obtained in a Jeol  
660 2010 (UniSi): sample LC-100AB-5 (1b), sample 09GAR-2.1 (2b), sample LC-100AB-4 (2c).  
661

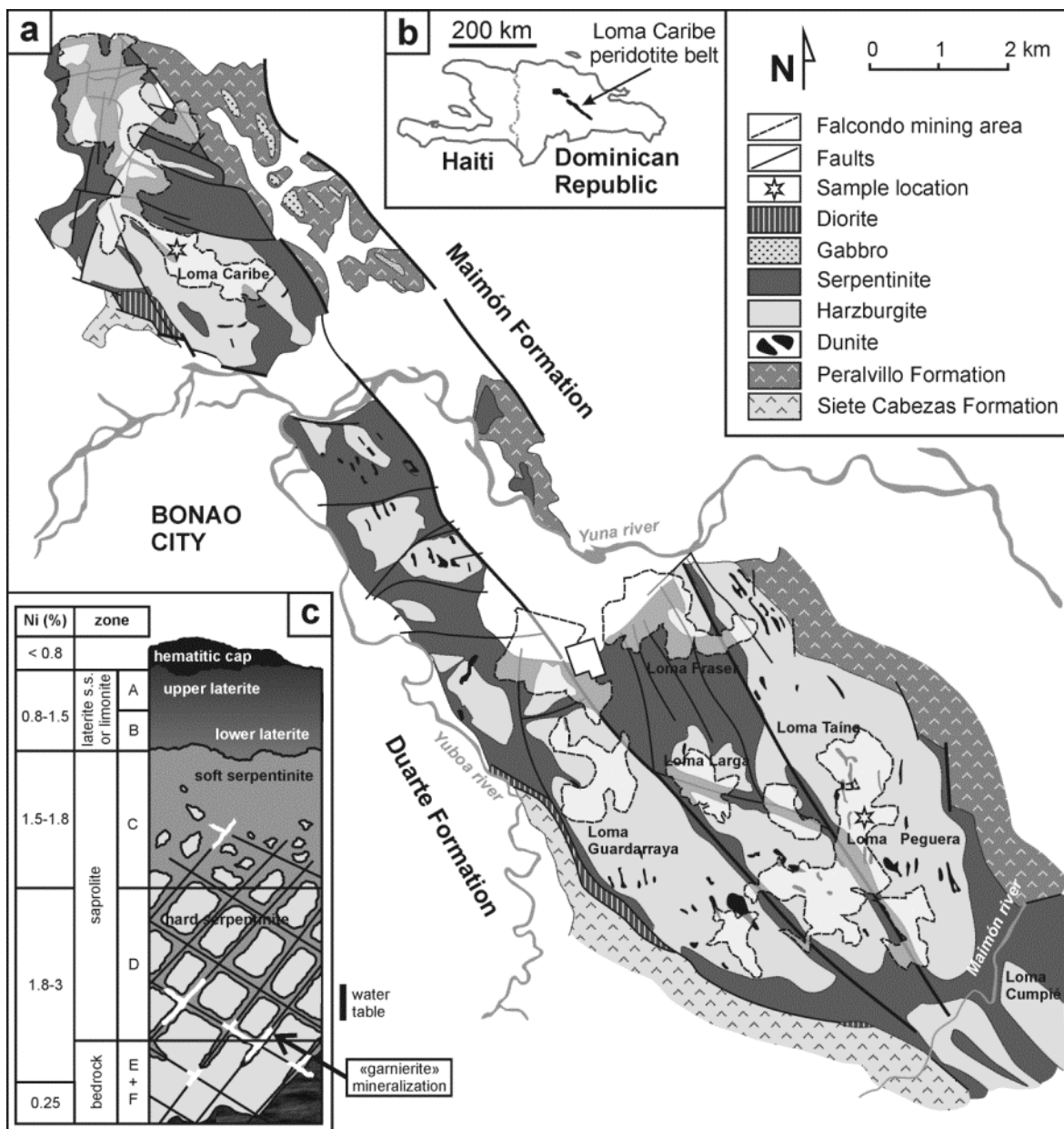
5

Type	Mineralogy (XRD)	Structural formula (EMP) [number of analyses averaged]	Xtal
TYPE I	serpentine + traces kerolite-pimelite (type I)	(Mg <sub>2.29</sub> Ni <sub>0.25</sub> Fe <sub>0.19</sub> )Si <sub>2.17</sub> O <sub>5.53</sub> (OH) <sub>3.27</sub> ·0.19H <sub>2</sub> O [23]	0.19
TYPE I	serpentine + traces kerolite-pimelite (type I)	(Mg <sub>2.34</sub> Ni <sub>0.48</sub> Fe <sub>0.15</sub> )Si <sub>2.48</sub> O <sub>6.27</sub> (OH) <sub>3.49</sub> ·0.25H <sub>2</sub> O [16]	0.25
TYPE II	serpentine + minor kerolite-pimelite (type II)	(Mg <sub>1.94</sub> Ni <sub>1.02</sub> Fe <sub>0.03</sub> )(Si <sub>2.56</sub> Al <sub>0.04</sub> )O <sub>6.49</sub> (OH) <sub>3.40</sub> ·0.30H <sub>2</sub> O [23]	0.30
TYPE III	serpentine + kerolite-pimelite (type III)	(Ni <sub>2.56</sub> Mg <sub>0.43</sub> )Si <sub>2.97</sub> O <sub>7.44</sub> (OH) <sub>3.02</sub> ·0.49H <sub>2</sub> O [27]	0.49
TYPE III	serpentine + kerolite-pimelite (type III)	(Ni <sub>2.42</sub> Mg <sub>0.56</sub> Fe <sub>0.01</sub> )Si <sub>2.71</sub> O <sub>6.81</sub> (OH) <sub>3.28</sub> ·0.36H <sub>2</sub> O [4]	0.36
TYPE IV	kerolite-pimelite (type IV)	(Mg <sub>1.90</sub> Ni <sub>1.09</sub> )Si <sub>3.89</sub> O <sub>9.76</sub> (OH) <sub>2.09</sub> ·0.91H <sub>2</sub> O [4]	0.91
TYPE IV	kerolite-pimelite (type IV)	(Mg <sub>1.72</sub> Ni <sub>1.27</sub> Fe <sub>0.01</sub> )Si <sub>3.84</sub> O <sub>9.61</sub> (OH) <sub>2.16</sub> ·0.92H <sub>2</sub> O [17]	0.92
TYPE IV	kerolite-pimelite (type IV)	(Ni <sub>1.95</sub> Mg <sub>1.04</sub> )Si <sub>3.69</sub> O <sub>9.24</sub> (OH) <sub>2.30</sub> ·0.85H <sub>2</sub> O [109]	0.85
TYPE V	sepiolite-falcondoite (type V)	(Mg <sub>4.60-7.05</sub> Ni <sub>1.01-2.39</sub> Fe <sub>0-0.05</sub> )(Si <sub>11.75-12.75</sub> Al <sub>0-0.02</sub> )O <sub>15</sub> (OH) <sub>2</sub> ·6H <sub>2</sub> O [30]	-
TYPE V	sepiolite-falcondoite (type V)	(Mg <sub>3.42-5.15</sub> Ni <sub>2.29-5.24</sub> Fe <sub>0-0.02</sub> )(Si <sub>11.57-12.55</sub> Al <sub>0-0.15</sub> )O <sub>15</sub> (OH) <sub>2</sub> ·6H <sub>2</sub> O [4]	-
TYPE V	sepiolite-falcondoite (type V)	(Mg <sub>6.51-7.94</sub> Ni <sub>0.34-0.58</sub> Fe <sub>0.01-0.27</sub> )(Si <sub>11.46-12.32</sub> Al <sub>0.01-0.03</sub> )O <sub>15</sub> (OH) <sub>2</sub> ·6H <sub>2</sub> O [16]	-

	Type I (Srp)	Type I (Srp)	Type I (Tlc)	Type I (Tlc)	Type II (Srp)	Type II (Srp)	Type II (Srp)	Type II (Tlc)	Type II (Tlc)
	LC-101 serp 1-2	LC-101 serp 1-8	LC-101 serp 1-10	LC-101 serp 1-11	LC-100B- Serp-12	LC-100B- Serp-15	LC-100B serp 3b-5	LC-100B- 2 (SERP)- 4	LC-100B- 2 (SERP)- 14
S 1	14	14	22	22	14	14	14	22	22
	2.737	2.509	2.374	2.111	2.756	2.556	2.590	1.482	1.008
	0.000	0.000	0.000	0.000	0.266	0.094	0.000	0.000	0.054
	1.825	2.085	3.869	3.846	1.888	1.899	2.132	3.905	3.907
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.386	0.160	0.158	0.072	0.023	0.305	0.067	0.018	0.018
	0.035	0.080	0.651	1.091	0.035	0.047	0.045	1.681	2.070
	3.158	2.750	3.183	3.273	2.814	2.908	2.702	3.182	3.097
	1.825	2.085	3.869	3.846	2.154	1.993	2.132	3.905	3.961
	type III	type III	type III	type IV	type IV	type IV	type V	type V	type V
	GAR- 2.3B-3	GAR- 2.3B-7	GAR- 2.3B-8	LC-100B- 4 (TLC)-2	LC-100B- 4 (TLC)-5	LC-100B- 4 (TLC)- 12	GAR-7v- 3	GAR-7v- 6	GAR-7v- 7
S 1	22	22	22	22	22	22	32	32	32
	0.364	0.417	0.413	1.440	1.494	1.725	1.961	2.138	2.013
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	3.770	3.757	3.966	3.997	4.015	4.026	6.065	6.106	6.090
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000	0.000
	3.096	3.068	2.656	1.566	1.476	1.222	1.910	1.649	1.806
	3.460	3.485	3.069	3.007	2.971	2.948	3.871	3.787	3.819
	3.770	3.757	3.966	3.997	4.015	4.026	6.065	6.106	6.090

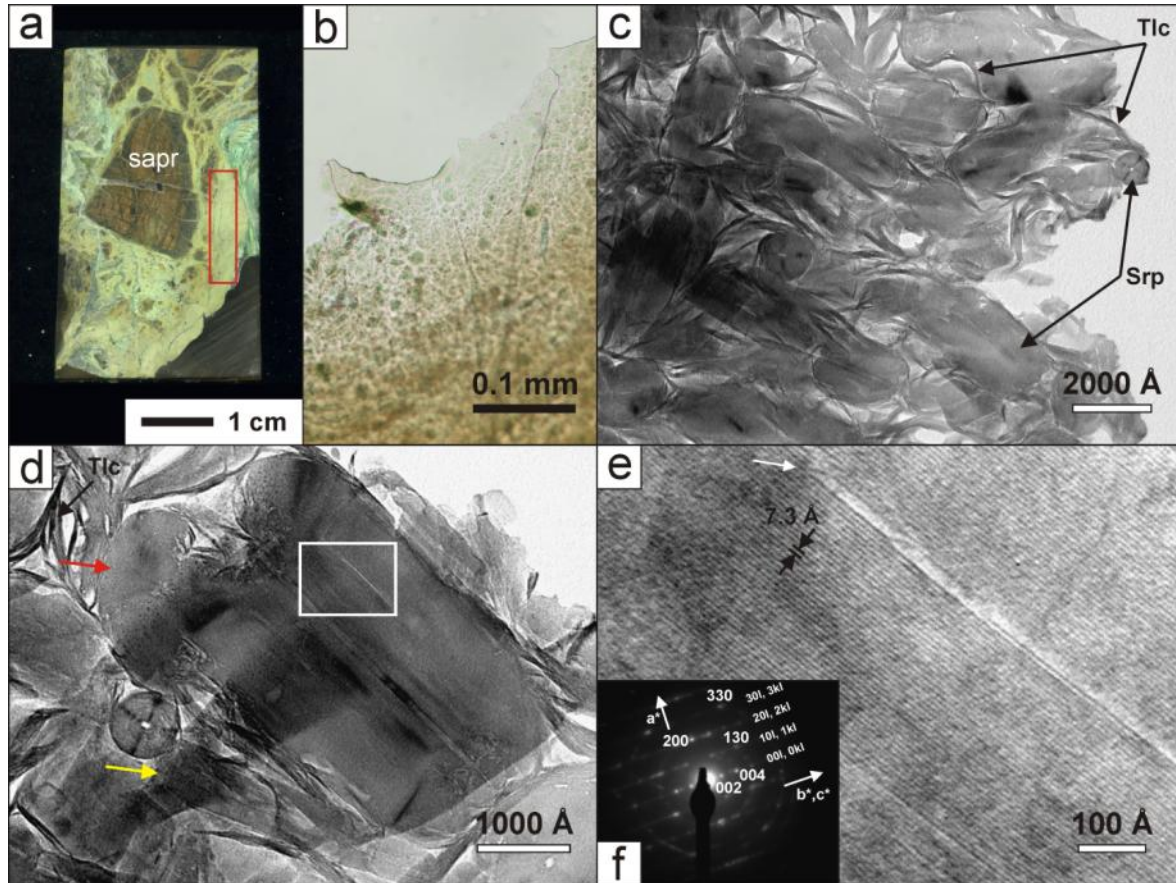
**FIGURES**

**Figure 1.**

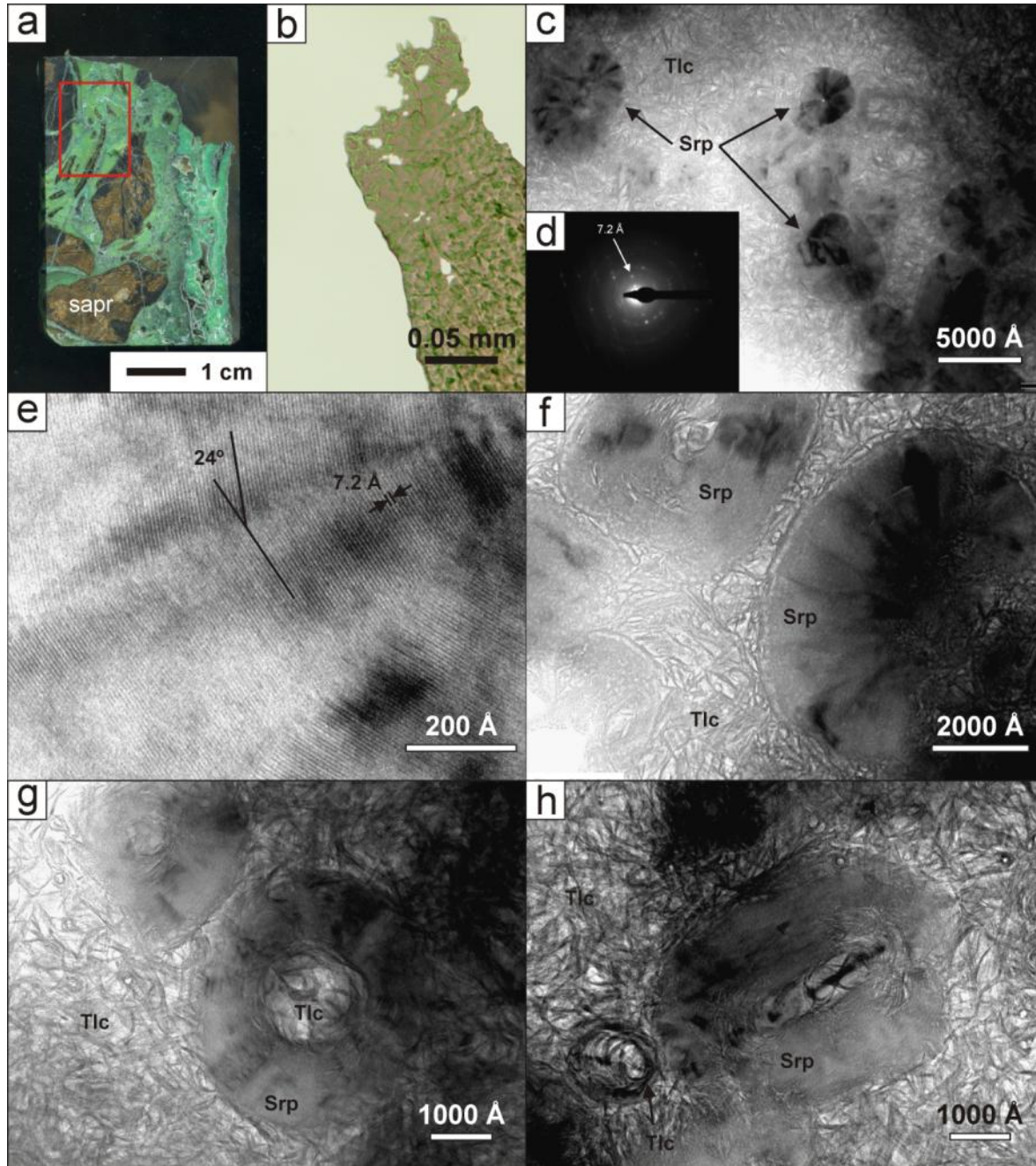




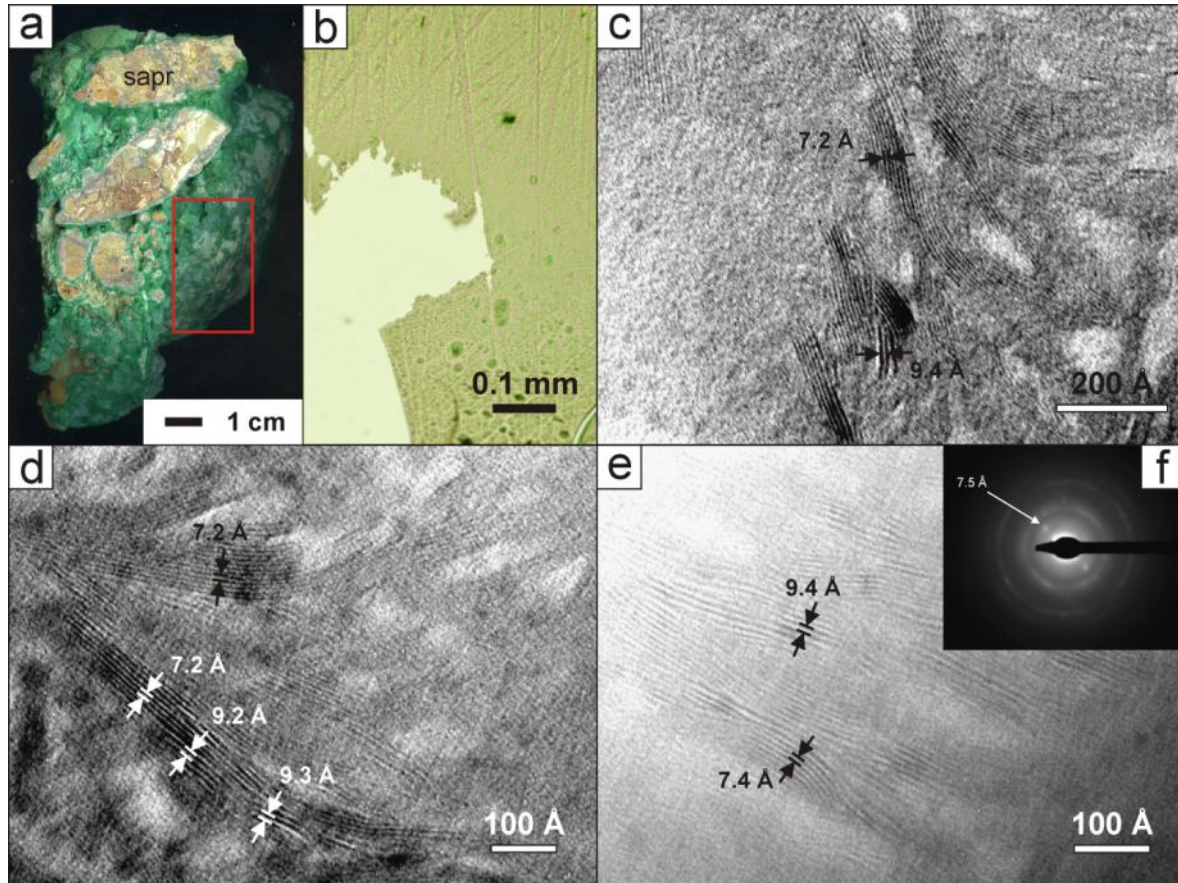
**Figure 2.**



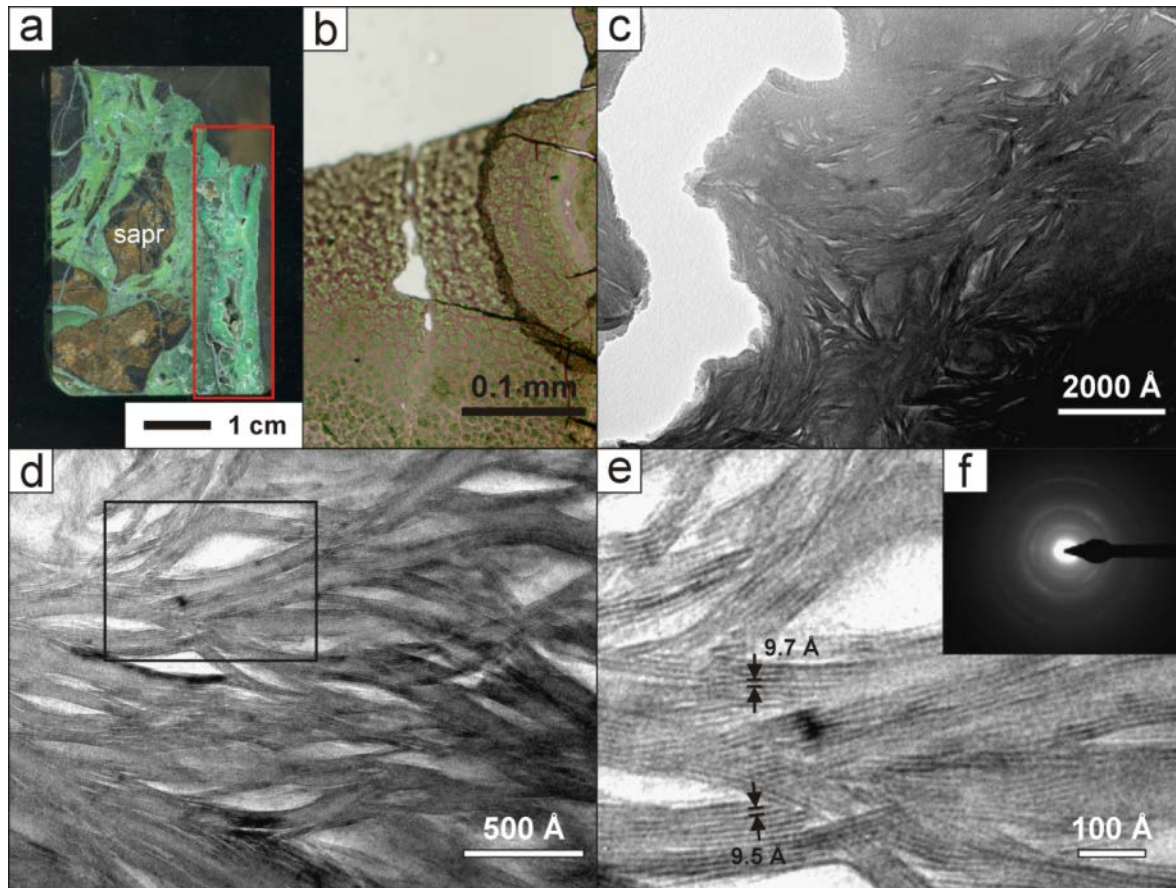
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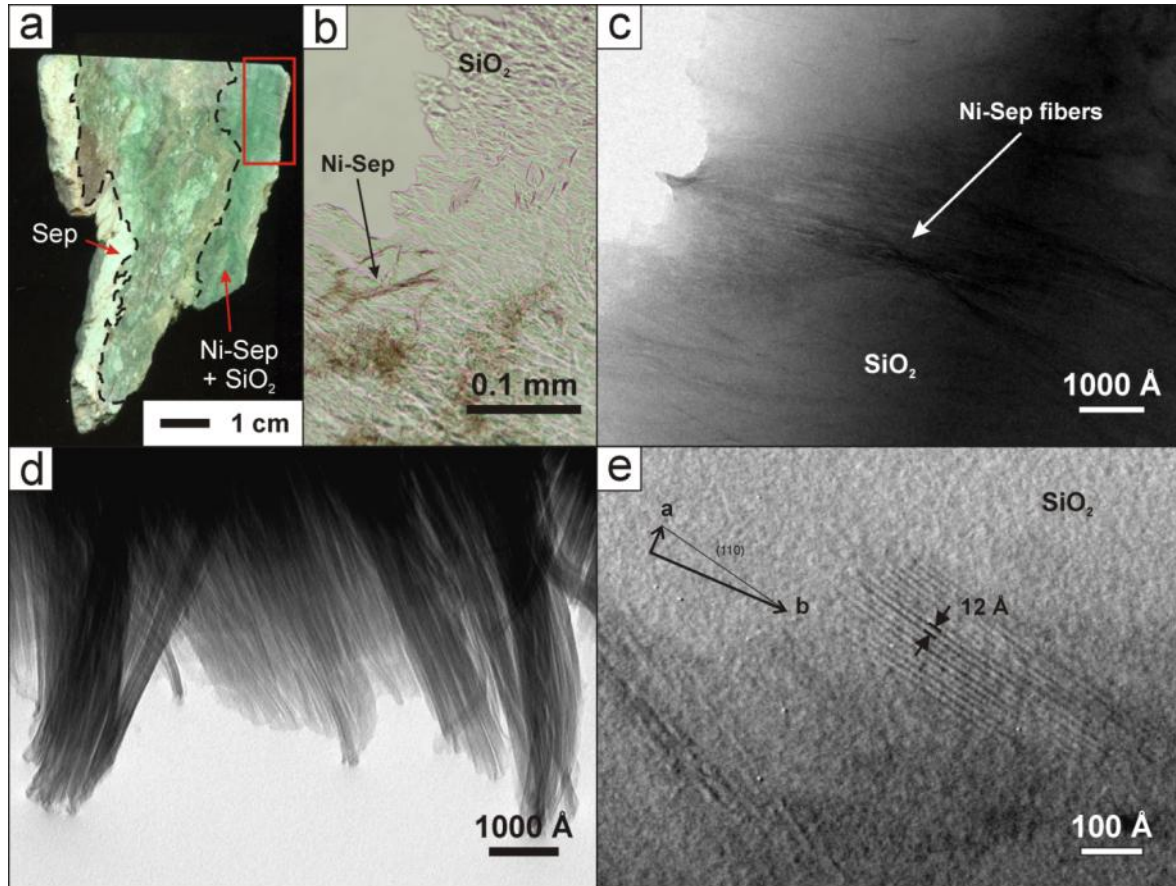
**Figure 4.**



**Figure 5.**



**Figure 6.**



**Figure 7.**

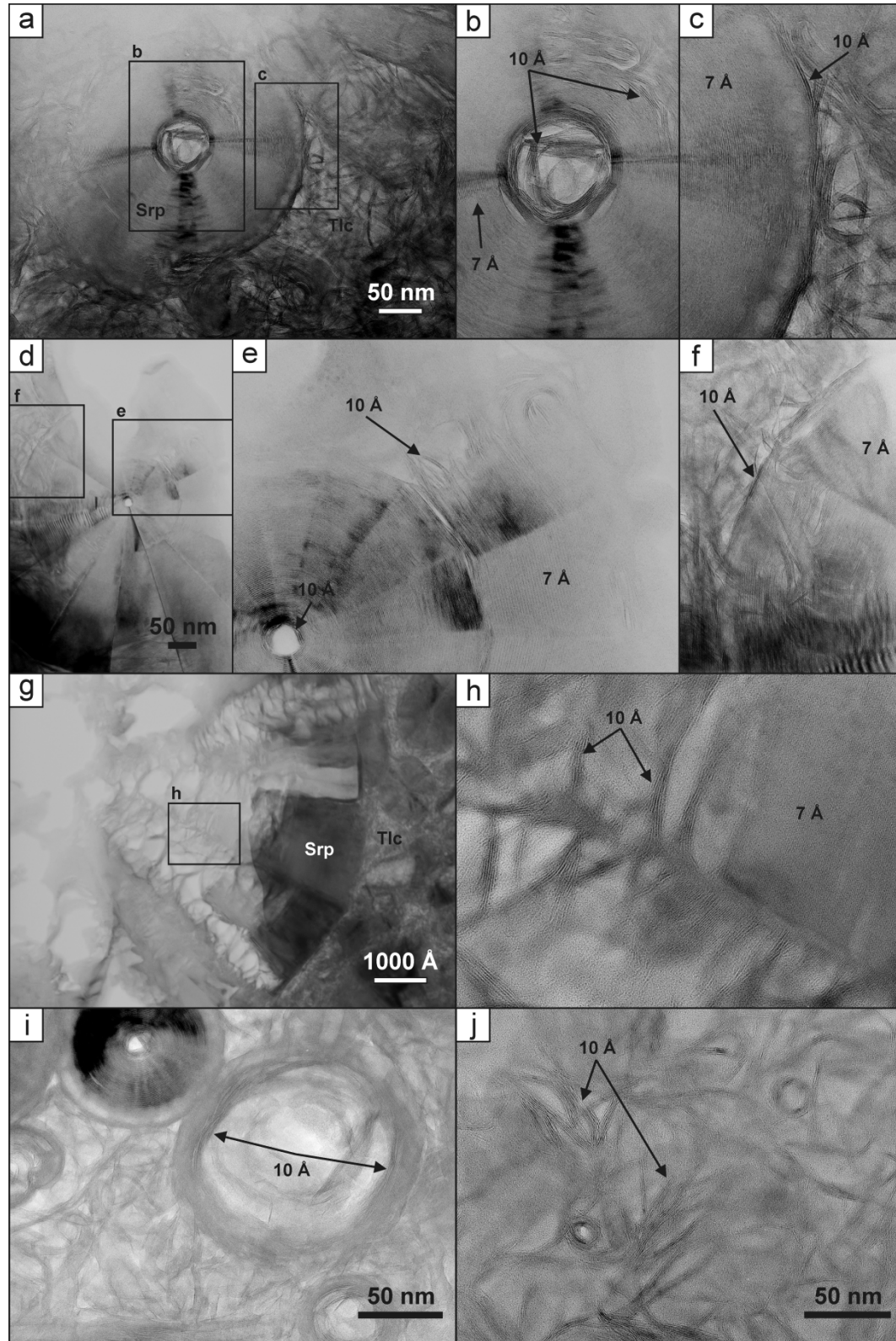
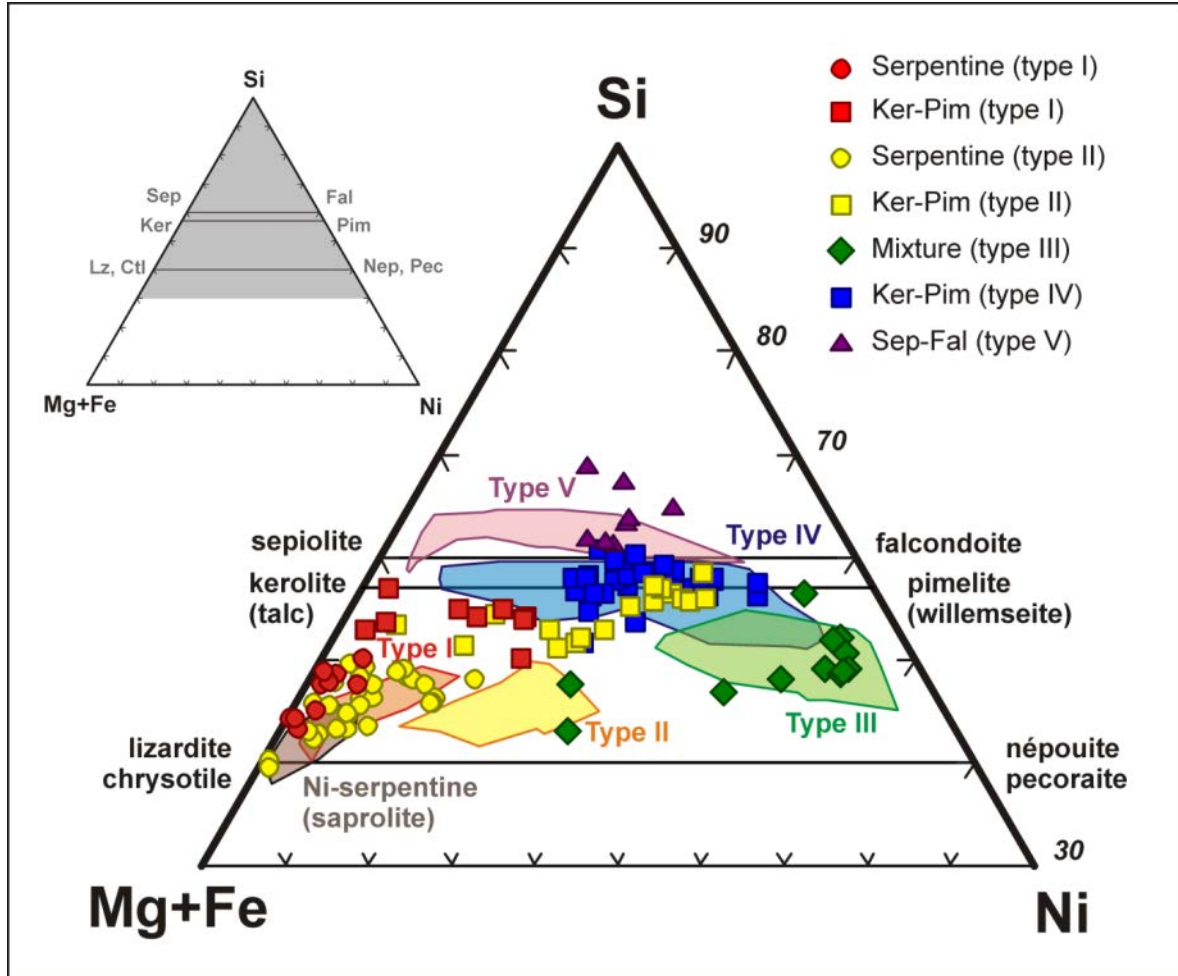


Figure 8.



**Figure 9.**

