

1 **Revision 3**

2 **Effects of differential stress on the structure and Raman**
3 **spectra of calcite from first-principles calculations**

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9
10 **Abstract:** Differential stresses are expected to influence the properties of minerals. The structural and
11 Raman vibrational properties of calcite under hydrostatic and differential stresses were studied using a
12 first-principles method based on density functional theory. Our results show that the density of calcite
13 increases or decreases under different differential stress, relative to its value under hydrostatic pressure.
14 The calculated effects of differential stress on bond lengths are nominal. As pressure increases, the
15 frequencies of all Raman modes increase, with their pressure derivatives slightly depending on the
16 differential stress. The state of stress influences the Raman modes by shifting their frequencies to either
17 higher or lower values relative to the corresponding hydrostatic results. In particular, the largest and
18 smallest frequency shifts were predicted for E_g-156 and A_{1g}-1086 modes, respectively, when the
19 additional stress was applied along the *a*-axis. Visualization of atomic motions associated with Raman
20 modes suggests that the differential stress-induced shifts in Raman frequencies are controlled by
21 out-of-plane vibrations of atoms. The stress estimated on the basis of the experimentally measured shift
22 of the Raman frequency of calcite sample gathered from the Wuchuan earthquake fault by applying our
23 calculated dv/dP value of A_{1g}-1085 mode is 785 MPa, which appears to be comparable to the stress
24 inferred at the Wuchuan earthquake focus. Thus, the first-principles simulations and Raman

25 spectroscopy experiments together may help us in elucidating the state of stress in the Earth's interior.

26

27 **Keywords:** Differential stress, Structure, Raman vibrational properties, Calcite, First-principles

28

29

30 **Introduction**

31 Differential stress exists in the Earth's interior. It can be produced by tectonic compression or by
32 deep fluid movement, particularly at plate boundaries or minerals' grain boundaries thereby possibly
33 driving plate tectonics and causing earthquakes. The effects of differential stress on the properties of
34 minerals are shown or expected to be substantial (Demouchy et al. 2011; Sato et al. 2013; Liu et al.
35 2014). Therefore, the state of differential stress in the Earth's interior, particularly its distribution in
36 space and its time-dependent evolution, is a fundamental issue in geoscience. However, this issue has
37 yet to be fully understood (Solberg et al. 1978; Kidder 2012).

38 Calcite is one of the most abundant minerals in the Earth's continental crust. Understanding its
39 structural, spectral, and mechanical properties under differential stress is important for understanding
40 the mechanical behavior of rock and the state of stress in the deep Earth. The X-ray diffractive and
41 Raman spectral detection have been widely used to determine the crystal structure and Raman
42 spectrum of minerals. In particular, laser Raman spectroscopic analysis of minerals (such as quartz)
43 allows us to investigate the pressure conditions in metamorphic rocks (Enami et al. 2007). As an
44 important rock-forming mineral, calcite is usually abundant in fault and tectonic boundaries. So, if we
45 quantify how the crystal structure and Raman frequencies of calcite behave under different stress
46 conditions, we can infer the state of stress of fault or tectonic boundaries by measuring the structure
47 and/or Raman spectrum of calcite gathered from the sites. Several investigations of the stress state of
48 the Earth's interior have been reported (Enami et al. 2007), by focusing on the relationship between

49 pressure and Raman frequency shift of minerals (Liu and Mernagh 1990; Dean et al. 1982; Prencipe et
50 al. 2004; Schmidt and Ziemann 2000). In general, these investigations have been performed under
51 hydrostatic pressure.

52 However, determining the complex microstructures of multiple mineral phases in mineral
53 assemblages and rocks under non-hydrostatic stress or differential stress represents a formidable
54 challenge. For example, the questions of what non-hydrostatic stresses can be developed and how
55 much of the differential stress state can be preserved remain unanswered (Weathers et al. 1979; Christie
56 and Ord 1980; Demouchy et al. 2011; Angel et al. 2014). At present, quantitatively investigating the
57 effects of differential stress on the structure and Raman spectrum is difficult by experiments.
58 Alternatively, the first-principles methods can be used to accurately calculate various properties of
59 minerals at the atomic scale (Gillan et al. 2006 ; Jahn and Kowalski 2014), including the state of
60 differential stress and its effects (Parrinello and Rahman 1982; Liu et al. 2014). Here, we investigated
61 the structure and Raman spectra of calcite from first-principles calculations to quantify the effects of
62 differential stress on these properties.

63

64 **Computational method**

65 **Crystal structure of calcite**

66 Calcite is a carbonate mineral with trigonal symmetry (space group: $R\bar{3}c$, two formula units (Z)
67 per cell). The structure consists of layers of Ca ions alternating with stacks of carbonate layers along
68 the c axis of the hexagonal cell. The carbonate layers are composed of planar triangular shaped
69 carbonate ions (CO_3^{2-}), with a carbon atom at the center of the triangle and the three oxygen atoms at
70 each corner (Fig. 1). The irreducible representation of the optical vibrations of calcite for D3 factor

71 group (Fateley et al. 1972) is described by:

$$72 \quad \Gamma_{\text{op}} = A_{1g} + 3A_{2g} + 4E_g + 2A_{1u} + 4A_{2u} + 6E_u,$$

73 where $A_{1g} + 4E_g$ represents Raman-active, $1A_{2u} + 4E_u$ infrared active, $1A_{2u} + 1E_u$ acoustic, and $3A_{2g} + 2A_{1u}$
74 spectroscopically inactive (silent) modes.

75

76 **First-principles computational details**

77 The first-principles calculations were performed using density functional perturbation theory
78 (Refson et al. 2006), density functional theory (Hohenberg and Kohn 1964; Kohn and Sham 1965), and
79 plane wave pseudopotential method, as implemented in the CASTEP codes (Clark et al. 2005). Local
80 density approximation (Ceperley and Alder 1980; Perdew and Zunger 1981) was used for describing
81 exchange-correlation interaction. Norm conserving pseudopotentials (Hamann et al. 1979) were used to
82 model electron-ion interaction, with plane wave cutoff of 900 eV. A $4 \times 4 \times 1$ Monkhorst Pack grid of k
83 points was adopted for sampling Brillouin zone. A convergence criterion of 5×10^{-7} a.u. on the total
84 energy was used in the self-consistent field calculations.

85 The structure and Raman spectroscopy of calcite at given pressures were calculated by
86 simultaneously optimizing both atomic positions and lattice constants under Hellmann–Feynman forces
87 and stresses acting on nuclei and lattice parameters, respectively (Nielsen and Martin 1983). Spatial
88 derivatives of the macroscopic polarization were calculated numerically along eigenvectors of each
89 Raman active phonon mode according to the polarization for each displacement using linear response
90 formalism (Gonze 1997). Once these derivatives are determined, the Raman cross-section through
91 appropriate averaging space can be calculated. Further details can be found in Porezag and Pederson
92 (1996) and Refson et al (2006). The calculated results on the structural, elastic, and vibrational

93 properties of calcite are in good agreement with previous results (Supplementary Table 1.), thus
94 demonstrating the validity of our calculations.

95

96 **The setting and expression of the differential stress**

97 Different stresses were applied to the crystal along *a*-, *b*-, and *c*-axis directions (i.e., *x*, *y*, and *z*
98 directions) marked as σ_{xx} , σ_{yy} , and σ_{zz} (Fig. 1). Thus, the equivalent hydrostatic pressure (***P***, GPa)
99 applied to crystals is $(\sigma_{xx} + \sigma_{yy} + \sigma_{zz})/3$. When $\sigma_{xx} = \sigma_{yy} = \sigma_{zz}$, the pressure is hydrostatic; when $\sigma_{xx} \neq$
100 $\sigma_{yy} \neq \sigma_{zz}$, the pressure is differential stress. Therefore, if σ_{xx} is equal to ***P***_{*I*}, σ_{yy} is equal to ***P***_{*I*}, and σ_{zz} is
101 equal to ***P***_{*I*} + *x* GPa, then ***P*** is equal to ***P***_{*I*} + *x*/3 GPa, differential stress is $\Delta\sigma_{xx} = \sigma_{xx} - \mathbf{P} = -x/3$ GPa,
102 $\Delta\sigma_{yy} = \sigma_{yy} - \mathbf{P} = -x/3$ GPa, and $\Delta\sigma_{zz} = \sigma_{zz} - \mathbf{P} = 2x/3$ GPa. The largest stress difference among σ_{zz} , σ_{yy} ,
103 and σ_{xx} is *x* GPa. Totally, five sets of different crystal structure and Raman spectrum of calcite were
104 calculated under differential stresses and hydrostatic pressure.

105

106 **Results and discussion**

107 **Effects of stress on cell parameters**

108 Figure 2 shows the calculated lattice parameters of calcite as a function of pressure. The *a*- and
109 *c*-axes decrease by 0.63 and 5.09% from 0 to 5 GPa. The *c*-axis is thus much more compressible than
110 the *a*-axis, thus implying highly anisotropic compression of calcite.

111 The lattice constant *a* increases and *c* decreases under differential stresses of $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} =$
112 $-1/6:-1/6:2/6$ and $= -1/3:-1/3:2/3$ relative to their equivalent hydrostatic values (Fig.2). The
113 differences of *a*- and *c*-axis from their equivalent hydrostatic values when $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/6:1/3:$
114 $-1/6$ and $= -1/3:2/3: -1/3$ are opposite to those when $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/6:-1/6:2/6$ and $=$

115 $-1/3:-1/3:2/3$. The lattice constants are smaller (larger) than their hydrostatic counterparts when the
116 positive (negative) differential stresses are applied along the axis direction. The effect on lattice
117 constant increases linearly with increasing differential stress.

118 As shown in Fig.2, the density increases by 0.23% (0.0067 g/cm^3) and 0.46% (0.0134 g/cm^3)
119 $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz}$ of $-1/6:-1/6:1/3$ and $= -1/3:-1/3:2/3$, respectively, with respect to its hydrostatic
120 value. On the other hand, the density decreases by 0.13% (0.0037 g/cm^3) and 0.20% (0.0058 g/cm^3)
121 when $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/6:1/3:-1/6$ and $= -1/3:2/3:-1/3$, respectively. The magnitude of these
122 density shifts also increases with increasing differential stress. The differential stress-induced changes
123 in the density of calcite are larger than those previously predicted for forsterite (0.003 g/cm^3 for under
124 the differential stress of $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/3:2/3:-1/3$ (Liu et al. 2014).

125 When pressure increases from 0 to 5 GPa, the calculated C–O and Ca–O bond lengths decrease
126 by 0.34, and 1.84%, respectively, and the calculated Ca–O octahedral volume decreases by 5.33%
127 (Supplementary Fig.1). Compared to its value under hydrostatic pressure, the octahedral volume
128 slightly increases when $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/6:-1/6:2/6$ and $= -1/3:-1/3:2/3$ and decreases when
129 $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/6:1/3:-1/6$ and $= -1/3:2/3:-1/3$ (Fig.3). However, the effects of differential stress
130 on the bond lengths are negligible.

131

132 **Vibrational pattern of Raman modes**

133 The calcite structure can be viewed in terms of independent Ca^{2+} -ion and carbonate-ion (CO_3^{2-})
134 units. The Raman modes of calcite given by $\Gamma_{\text{op}} = A_{1g} + 4E_g$ can be grouped into three categories:
135 translation (T), rotation (R), and internal libration (I). The four E_g modes show all vibrational types (T ,
136 R , and I) but the A_{1g} mode shows only I -type vibration (Bhagavantam and Venkatarayudu 1969). We

137 identified the vibrational pattern of each calcite Raman mode by visualizing the associated atomic
138 motions (vibrational direction and intensity of atoms) with the aid of animation of the computed
139 vibrational modes built into the CASTEP package. These vibrational patterns are illustrated in Fig. 4.

140 The atomic vibrations associated with the E_g -156 mode include out-of-plane translations of two
141 O atoms of planar carbonate unit and in-plane vibrations of one O and one C atom parallel to the b -axis,
142 that is, (001) crystal face. The relative vibrational intensities (i.e., the magnitudes of displacements) of
143 the four atoms of the carbonate unit do not differ substantially. The E_g -284 mode includes out-of-plane
144 translations of O and C atoms, and the vibrational intensity of the O atom is greater than that of the C
145 atom. In the case of the E_g -712 mode, its vibrations mainly correspond to the nearly in-plane rotation of
146 two O atoms and in-plane libration of one O and one C atom parallel to the b -axis. Similar to the
147 E_g -156 mode, the vibrational intensity of the four atoms of a carbonate do not substantially differ. The
148 vibrational properties of the E_g -1434 mode are similar to those of the E_g -156 mode, involving
149 out-of-plane translations of two O atoms and in-plane libration of one O and one C atom with different
150 vibrational directions. However, in this case, the vibrational intensity of the O atom is much smaller
151 than that of the C atom. The vibration of the A_{1g} -1086 mode corresponds only to the stretching of C–O
152 bonds. The C atom remains stable, whereas the O atoms move. The Ca^{2+} ions are not involved in any of
153 the five Raman vibrational modes.

154

155 **Effects of hydrostatic stress on the Raman frequency**

156 The directions and intensities of the vibrations of all the atoms involved in the Raman modes
157 remain nearly unchanged with increasing pressure. Though pressure does not affect the vibrational
158 modes significantly, its effects on vibrational frequencies are substantial. The pressure derivatives

159 (dv_i/dP) of five Raman frequencies under hydrostatic pressure are listed in Table 1. All Raman
160 frequencies systematically increase with increasing pressure. Among the E_g vibrational modes, E_g -284
161 has the largest pressure derivative and E_g -712 has the smallest derivative. Our calculated pressure
162 derivatives of the E_g and A_{1g} modes compare favorably with previous experimental and calculated
163 results (Liu and Mernagh 1990; Gillet et al. 1993).

164 According to the anisotropic compression of calcite, pressure likely affects the out-of-plane
165 vibrations more strongly than the in-plane vibrations of Raman modes. Therefore, the E_g -284 mode
166 exhibits stronger pressure dependence. In the cases of the E_g -156 and E_g -712 modes, the O atoms and C
167 atoms have similar relative vibrational intensities and these modes exhibit relatively small pressure
168 derivatives. In the case of E_g -284 and E_g -1434, the relative vibrational intensities of the O atoms and C
169 atoms differ, resulting in relatively high values of frequency-pressure derivatives. Thus, the difference
170 between the relative vibrational intensities of the O atoms and C atoms can explain partly the
171 differences in the pressure derivatives of Raman frequencies. As the difference between the relative
172 vibrational intensities of O atoms and C atoms increases, Raman frequency depends more strongly on
173 pressure.

174

175 **Raman modes under differential stress**

176 The vibrational directions and intensities of atoms under differential stresses show patterns similar
177 to those under hydrostatic pressure (Fig. 5). Also, the pressure derivatives of all modes under
178 differential stresses are similar to those under hydrostatic pressure; the differences being less than 0.3
179 $\text{cm}^{-1}/\text{GPa}$. Some new Raman peaks appeared under differential stress but their intensities are small. For
180 the characteristic Raman modes of calcite, the frequencies change under differential stresses. Relative

181 to the values under hydrostatic pressure, the E_g -156 mode frequency shifts to higher value by 2.6 cm^{-1}
182 under $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/6:-1/6:1/3$, but to lower value by 1.1 cm^{-1} under $\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/6:1/3:$
183 $-1/6$ (Table 2). The E_g -284 and E_g -1434 modes show the similar shift pattern except their smaller
184 values under differential stresses. For all three modes, the two differential stress states show opposite
185 (positive and negative) shifts on Raman frequency. However, the E_g -712 mode shows negative shifts of
186 0.7 and 0.9 cm^{-1} under both the differential stress states ($\Delta\sigma_{xx}:\Delta\sigma_{yy}:\Delta\sigma_{zz} = -1/6:-1/6:1/3$ and $-1/6:1/3:$
187 $-1/6$). Finally, the differential stresses induce little or no frequency shifts in the case of the A_{1g} -1086
188 mode. Our results thus show that the differential stress has the strongest and weakest effects on the
189 frequency shifts in the cases of the E_g -156 and A_{1g} -1086 mode, respectively. For all modes except
190 A_{1g} -1086, the magnitude of the frequency shift increases with the increasing differential stress.

191 When the additional stress was applied along the c -axis, the differential stress affects the
192 in-plane vibrations of atoms much less than the out-of-plane vibrations (the plane refers to (001) crystal
193 face). As discussed above, the atomic vibrations associated with the E_g -156 and E_g -284 modes include
194 some out-of-plane vibration, but the atomic vibrations associated with E_g -712, E_g -1434, and A_{1g} -1086
195 modes mostly include the in-plane vibration. The differential stress does not affect the in-plane bond
196 distance (C-O bond) much (Fig.3), therefore, the E_g -156 and E_g -284 modes show larger frequency
197 shifts than the E_g -712, E_g -1434, and A_{1g} -1086 modes.

198 When the additional stress was applied along the c -axis, the c -axis becomes smaller than its
199 equivalent hydrostatic pressure value and the intensities of out-of-plane vibration of atoms increase
200 thereby shifting the E_g -156 and E_g -284 modes to higher frequency. On the other hand, when the
201 additional stress was applied along the a -axis, c -axis is larger than its equivalent hydrostatic pressure
202 value and the intensities of out-of-plane vibration of atoms decrease thereby shifting these modes to

203 lower frequency. This means that the differential stress-induced frequency shifts are mainly controlled
204 by the out-of-plane vibrations of atoms. For E_g -712, the frequency shifts to lower value for the cases of
205 additional stress.

206

207 **Conclusions and implication**

208 Our first-principles results on the structural and Raman vibrational properties of calcite under
209 pressure and differential stresses may help advance our knowledge about the state of stress or gravity in
210 the earth's interior. First, we summarize our main findings: The effects of differential stress are to
211 increase or decrease the lattice constants and the density of calcite depending on the stress type. The
212 calculated pressure derivatives of all Raman modes of calcite under differential stresses are positive
213 and similar to those under hydrostatic pressure. Our results show that the Raman vibrational
214 frequencies are sensitive to differential stress, with E_g -156 and A_{1g} -1086 modes showing the largest
215 and smallest shifts, respectively. The state of stress also affects the Raman frequencies, shifting to
216 either higher or lower value, that is, showing positive or negative shifts with respect to the
217 corresponding hydrostatic values. Therefore, the relationship between pressure and the Raman
218 frequencies of minerals investigated using hydrostatic pressure (Asell and Nicol 1968; Schmidt and
219 Ziemann 2000; Liu and Mernagh 1990; Gillet et al. 1993) is not valid under non-hydrostatic stress
220 conditions. To investigate state of stress in metamorphic rocks (Enami et al. 2007), laser Raman
221 spectroscopic analysis can relate Raman frequencies shifts with changes of differential stress.

222 Here, we show what the calculated effects of differential stress (and pressure) on Raman
223 frequency shifts can tell us about the state of stress of minerals. The stress determined by the core
224 differential strain analysis method in the Wenchuan-Earthquake Fault Scientific Drilling (WFSD)

225 project located at Hongkou town, Dujiangyan city, Sichuan province shows that in situ stress in the
226 Earth's interior increases with increasing depth and ranges from 69.1 MPa (maximum principle stress)
227 to 35.3 MPa (minimum principle stress) at a depth of 1.19 km (Peng et al. 2011). These measurements
228 also demonstrate that differential stress exists in the Earth's interior. We extrapolated the results to the
229 19-km focal depth of the Wenchuan earthquake (magnitude 7.9, Eastern Sichuan, China; USGS; May
230 12, 2008) and obtained the maximum and minimum principle stresses of 1110 MPa and 566 MPa,
231 respectively. The Raman peaks associated with A_{1g} -1085 mode of calcite collected from the Wenchuan
232 earthquake fault (Shenxigou fault at Hongkou town, Dujiangyan city, Sichuan province) were observed
233 to be shifted 2.7 cm^{-1} toward higher frequencies (Xie et al. 2015). Then the residual stress in calcite
234 crystal is 785MPa on the base of our calculated dv/dP value of A_{1g} -1085 mode. The value is close to the
235 stress at 19km depth in WFSD thereby suggesting that stress of seismic fault is similar to the stress of
236 hypocenter. Unfortunately, the effects of differential stress on A_{1g} -1085 mode are too small for
237 deducing and verifying the state of differential stress. If we know the shifts of other Raman modes such
238 as their E_g -156 and E_g -712 modes for the sample, the state of differential stress can be deduced. We
239 anticipate that this approach can be applied to other several mineral samples gathered at the same sites
240 (the fault or other tectonic zone), such as quartz, feldspar, etc. so that we can obtain more specific
241 information about the state of stress. Our study thus demonstrates that we can learn about the state of
242 stress in the Earth's interior by combining first-principles simulations and Raman spectroscopy studies.

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341 **Tables**

342 Table 1 Pressure shifts of the Raman modes

| Symmetry | E _g -156 | E _g -284 | E _g -712 | E _g -1434 | A _{1g} -1086 | Reference |
|---|---------------------|---------------------|---------------------|----------------------|-----------------------|----------------------|
| ν ₁₀ (cm ⁻¹) | 164.6 | 297.3 | 696.0 | 1441.2 | 1086.9 | This study |
| | 156 | 281 | 711 | 1434 | 1085 | Liu and Mernagh 1990 |
| dν _i /dP (cm ⁻¹ /GPa) | 3.2 | 5.7 | 1.9 | 4.9 | 3.4 | This study |
| | 2.5 | 5.3 | 2.2 | 5.9 | 9 | Liu and Mernagh 1990 |
| | 2.3 | 6 | 1.7 | 3 | 7.5 | Gillet et al. 1993 |

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345 Table 2 Shifts of Raman frequencies under differential stress with respect to its hydrostatic value

| | Symmetry | | | | | State of stress |
|---|---------------------|---------------------|---------------------|----------------------|-----------------------|--|
| | E _g -156 | E _g -284 | E _g -712 | E _g -1434 | A _{1g} -1086 | Δσ _{xx} : Δσ _{yy} : Δσ _{zz} |
| Shift of frequency (cm ⁻¹) | 2.6 | 1.2 | -0.7 | 0.6 | | -1/6 : -1/6 : 1/6 |
| | 5.3 | 2.5 | -1.1 | 0.9 | | -1/3 : -1/3 : 2/3 |
| | -1.1 | -1.3 | -0.9 | -0.1 | | -1/6 : 1/3 : -1/6 |
| | -2.7 | -2.3 | -2.3 | -0.3 | | -1/3 : 2/3 : -1/3 |

346 Notes: The positive values mean shift toward higher frequency, and negative values mean shift toward
347 lower frequency

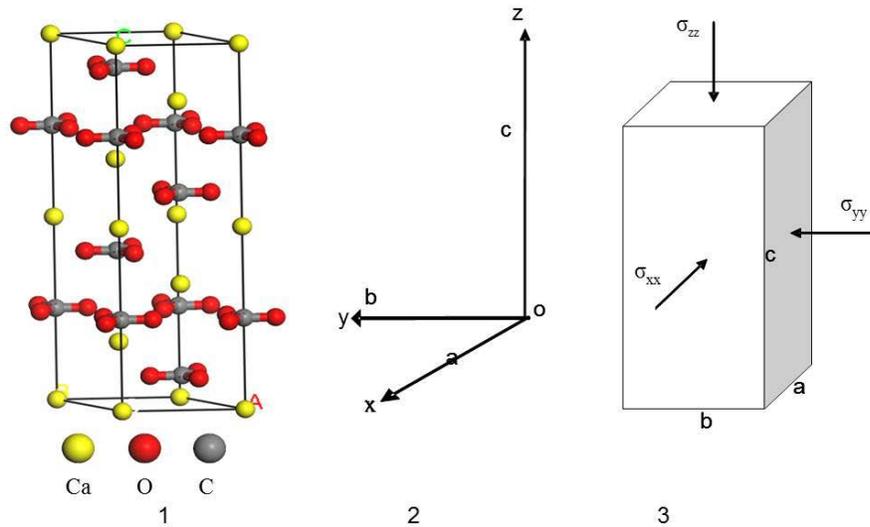
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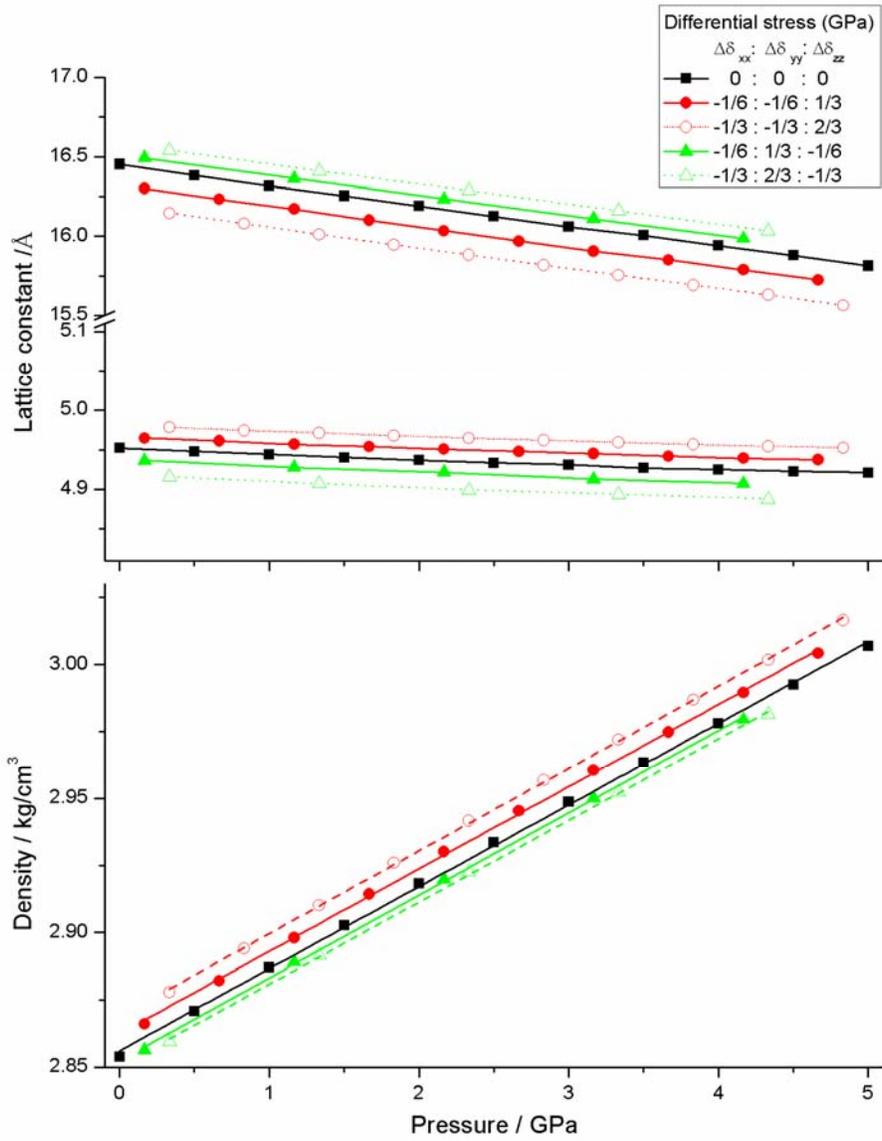
352 **Figures**



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354 **Fig. 1.** Atomic structure of calcite and schematic diagram of differential stress project. (1) Atomic

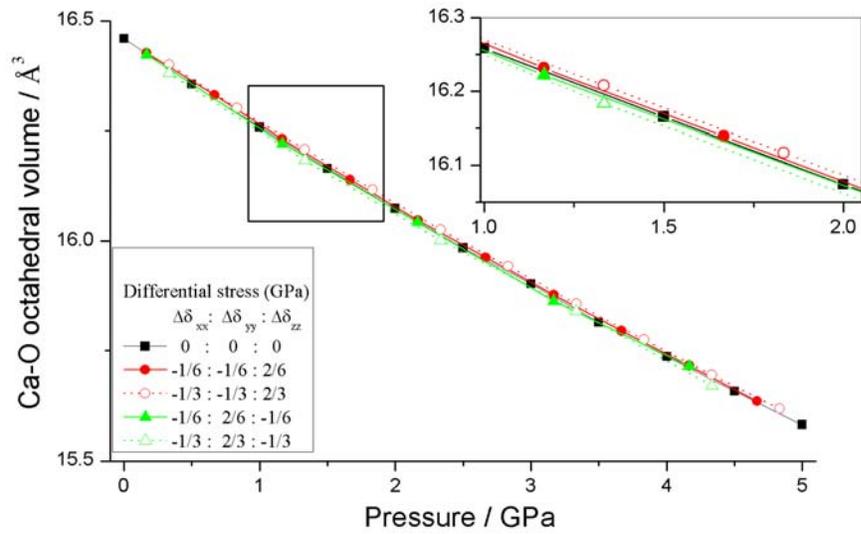
355 structure, (2) Lattice constants and directions, and (3) Applied stresses (σ_{xx} , σ_{yy} , and σ_{zz}).



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Fig. 2 Changes of lattice constants and density of calcite with pressure

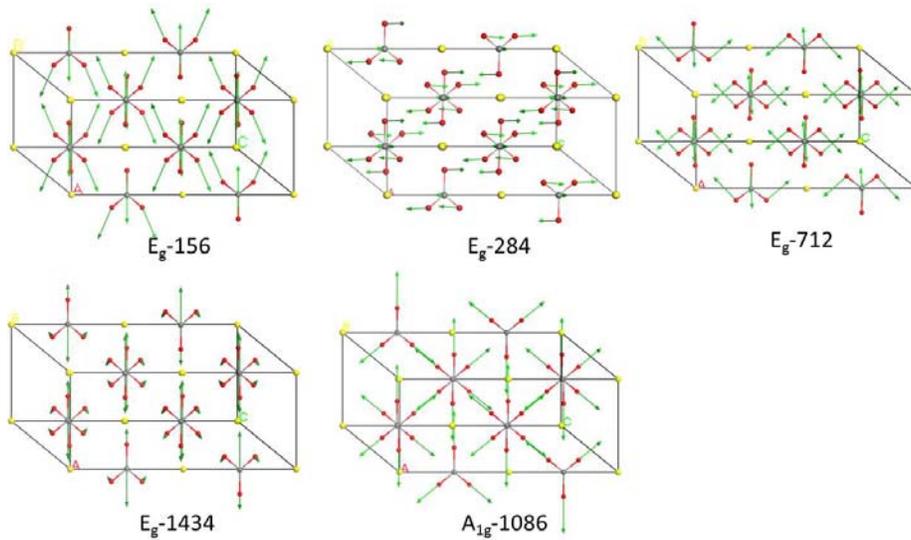


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Fig.3. Changes of bond length and polyhedral volume of calcite with pressure.

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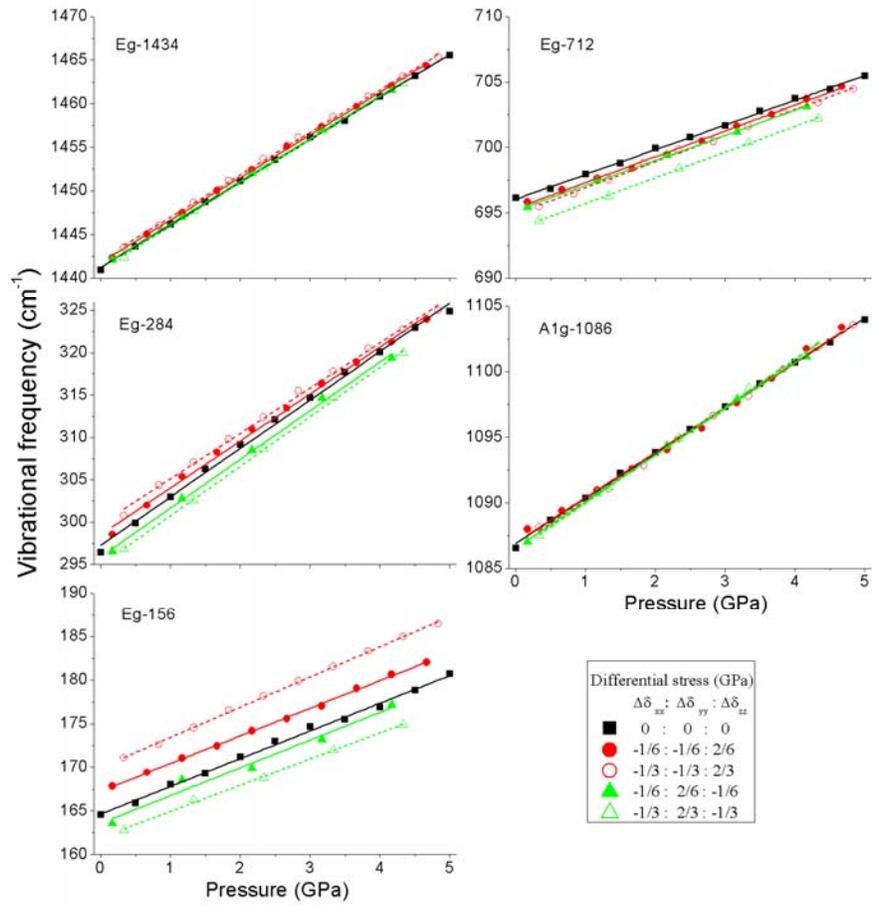
Fig. 4 Schematics of vibrational pattern of each calcite Raman mode (viewed from [101]). The arrows

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and their lengths indicate the direction of atomic motions and the relative vibrational intensities of

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atoms, respectively.



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Fig. 5 Pressure dependence of the Raman vibrational frequencies of calcite

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