

25 elements including: Ca, Pb, REE+Y (hereafter Σ REY), Th, Fe, Si, P, Al, Mg, Mn, K, and many others
26 (e.g., Finch and Murakami, 1999). The ~1590 Ma Olympic Dam (OD) iron-oxide copper gold deposit
27 is unusually rich in uranium where uraninite (ideally UO_2) is one of three main U-minerals (the others
28 are coffinite and brannerite; Ehrig et al., 2012, and references therein). Two generations of uraninite,
29 comprising four main textural classes are identified at OD (Macmillan et al., 2016): 1) ‘primary’, 2)
30 ‘zoned’, 3) ‘cob-web’ and 4) ‘massive’. Of relevance here are the ‘early’ generation (Classes 1-3), all
31 characterized by Pb- and Σ REY-rich (up to 0.42 apfu collectively) single grains (tens to hundreds of
32 μm in size). Differentiation between classes of ‘early’ uraninite is based on textural and chemical
33 zonation patterns (Macmillan et al., 2016). The authors show that ‘primary’ uraninite represents the
34 least-altered, most pristine, crystalline uraninite, whereas ‘zoned’ and ‘cob-web’ types have undergone
35 chemical-textural modifications by *in-situ* alteration processes (i.e., single grains of ‘primary’ uraninite
36 are progressively altered via solid-state diffusion followed by interaction with hydrothermal fluids to
37 form ‘zoned’ through to ‘cob-web’ uraninite). The ‘cob-web’ class consists of rhythmic intergrowths of
38 uraninite and sulfides from core to margin within any given grain. In contrast, changes in chemical
39 zonation patterns relative to grain morphologies used to define ‘zoned’ uraninite as distinct from the
40 ‘primary’ uraninite is less well constrained, and is the subject of the present study. The main question
41 addressed here, is whether there is a link between the chemical zoning and microstructures within
42 grains that show such modifications, and if so, could this provide clues in understanding the
43 crystallization and alteration history of uraninite.

44 **BACKGROUND AND RATIONALE**

45 Morphological changes between internal zoning and the margin outline as seen in grains from the
46 ‘zoned’ class of uraninite (Macmillan et al., 2016) can result either from primary crystallization
47 processes, during which rates and growth orientation change (e.g., in garnet; Allen and Buseck 1988),
48 or from dynamic recrystallization in a broad range of geological environments (e.g., Urai et al., 1986;

49 Steffen and Selverstone, 2006). The combined use of electron back-scatter diffraction (EBSD) and
50 orientation contrast (OC) imaging to study microstructures within minerals can be applied to any
51 mineral at a range of scales (Prior et al., 1999). From EBSD and OC data it is possible to quantify
52 microstructures empirically and constrain dislocation slip systems, and this coupled with other micro-
53 and nanoscale observations can lead to a more rigorous understanding of the formation of the observed
54 mineral textures. The presence of microstructures, boundaries and interfaces (i.e., slip systems, sub-
55 grain boundaries) have been shown to be important in controlling alteration processes, and the
56 formation of micro- and meso-textures (Prior et al., 1999 and 2002).

57 The use of microstructural data from EBSD and OC imaging has been the subject of numerous
58 studies to quantify the formation of various mineral textures. In zircon, microstructural features may
59 form as a result of a combination of primary growth characteristics, degree of radiation damage, and
60 recovery from crystal-plastic deformation, e.g., formation of dislocations, low-angle grain boundaries,
61 and movement of slip systems (e.g., Reddy et al., 2006). In other minerals (e.g., garnet, spinel) there
62 are strong crystallographic preferred orientations and these microstructural features are shown to be
63 linked to slip systems, dislocation creep and recovery (Boyle et al., 1998; Prior et al., 2002).

64 Defining grain-scale characteristics in radiogenically modified minerals like zircon or uraninite is
65 important for constraining their geologic evolution or U-Pb ages, particularly when zoning, porosity,
66 fractures and microstructures are present. Incorporation and release of daughter products of ^{235}U and
67 ^{238}U decay such as ^{207}Pb and ^{206}Pb , respectively, depend upon the robustness of the crystal lattice to
68 radiation damage (i.e., amorphization and healing rates), as well as external factors such as exposure to
69 hydrothermal fluids. In contrast to zircon, healing rates are considered much faster for uraninite and the
70 mineral does not suffer from amorphization due to radiation damage (e.g., Janeczek and Ewing, 1991).

71 There are three slip systems identified for UO_2 : $\{001\}\{1\bar{1}0\}$, at room temperature and atmospheric
72 pressure, and $\{110\}\{1\bar{1}0\}$ or $\{111\}\{1\bar{1}0\}$, at higher temperatures (Kelly et al., 2012). Any of these

73 could be activated to generate microstructures during various processes, including concentration and/or
74 release of minor/trace elements in uraninite. Here we employ EBSD analysis to study uraninite that has
75 the necessary prerequisites to show relationships between minor element redistribution (modification of
76 chemical zoning, fracture infill and mineral inclusions; Macmillan et al., 2016) and microstructures
77 resulting from lattice distortion.

78 ANALYTICAL METHODOLOGY

79 Quantitative analysis of uraninite was performed using a Cameca SX-Five Electron Probe Micro-
80 Analyzer (EPMA; Adelaide Microscopy, University of Adelaide) equipped with 5 tunable wavelength-
81 dispersive spectrometers (WDS). Twenty-eight elements were measured; methodologies for data
82 collection/analysis are given in Macmillan et al. (2016) and Appendix 1. The EPMA was also used to
83 generate WDS elemental maps including those for Pb ($M\alpha$) and Ce ($L\alpha$) in Fig. 1a and b, respectively.

84 Electron back-scatter diffraction (EBSD) data were collected using the EDAX-TSL™ EBSD system
85 on a FEI Helios NanoLab DualBeam™ FIB/SEM platform (Adelaide Microscopy). Analytical details
86 are given in Appendix 1. Three data processing methods are applied: Inverse Pole Figure (IPF), Grain
87 Reference Orientation Deviation (GROD) and Image Quality (IQ) mapping.

88 RESULTS

89 The uraninite studied is zoned with respect to minor elements, with Pb and Σ REY (Ce is a proxy for
90 LREE) having the highest concentrations, and these co-correlate with one another (Fig. 1a and b). The
91 presence of discrete REY-minerals as minute inclusions can also be inferred from the Ce map (Fig. 1b).
92 This grain was chosen because the chemical patterns show two, (Pb+ Σ REY)-poor, porous and fractured
93 domains with equant-rhombic shape, aligned along the long axis of the planar section. One side of the
94 section also displays edges parallel to the rhomb faces. Bornite and fluorite are also present as infill of
95 fractures (Fig. 1a). Such a zonation pattern relative to the morphology of the grain, could indicate a
96 type of sector zoning during primary growth, or alternatively, relate to secondary processes involving

97 minor element redistribution within the grain. Based on the color coding of the IPF map (Fig. 1c and
98 d), the grain orientation lies between the $\langle 111 \rangle$, $\langle 001 \rangle$ and $\langle 101 \rangle$ zone axes, and has been estimated as
99 $\langle 11\bar{2} \rangle$ (based on simulation of pole figures in Fig. A1, Appendix 2). Face indexing of the grain is
100 shown accordingly on Fig. 1c.

101 The subtle gradational color variations on the IPF map (Fig. 1c and d) indicate gradual changes in
102 crystallographic orientation but no apparent relationships with the chemical zoning. The boundaries
103 formed by a lattice rotation about the plane normal to several directions have been superimposed on the
104 IPF and IQ maps, as have low-angle grain boundaries on the GROD map (Figs. 1 to 3). This was
105 appropriate to test if there was any relationship between these boundaries and the known slip systems
106 for UO_2 . Slip systems such as $\{111\}\langle 1\bar{1}0 \rangle$, with rotation about the $\langle 11\bar{2} \rangle$ direction, and which
107 correspond to the present grain orientation, show only a few tilt boundary traces on the IPF map (Fig.
108 1c), indicating little correlation between microstructures and chemical heterogeneity in the grain. In
109 contrast, superimposed tilt boundaries formed by lattice rotation about $\langle 1\bar{1}0 \rangle$, normal to $\{001\}\langle 1\bar{1}0 \rangle$,
110 but different to the present grain orientation, correlate with bornite \pm fluorite infilled cracks, arrays of
111 inclusions, and some of the chemical zonation boundaries on the IPF map (Fig. 1d).

112 Intragranular orientation deviation as displayed by GROD mapping (Fig. 2a) occurs as a response to
113 deformation, or where there is stored strain (relative to a reference orientation) within a grain. There are
114 variations in intragrain crystallographic orientations of up to 18° , although the majority of the grain has
115 an orientation of between 3.5° and 7° (green) compared to the reference (blue, marked by a white
116 cross). Moreover, the misorientation profile (Fig. 2b) displays the range of orientations along the A-B
117 profile delineated on the GROD map. Low-angle grain boundaries ($<10^\circ$) are superimposed on the
118 GROD map, and correlate with the tilt boundaries plotted on the IPF and IQ maps (Figs. 1d and 3a).
119 Areas that appear to have higher relative stored strain (yellow-red) are found in regions that have a
120 higher concentration of, or are surrounded by, many low-angle grain boundaries (Fig. 2a). When

121 comparison is made between the tilt boundaries circled (black dotted lines on Fig. 1d) and the
122 compositional maps (Figs. 1a and b), the shape of the concentration boundaries of (Pb+ Σ REY) appear
123 similar to that of some of the tilt boundaries.

124 The IQ map (Fig. 3a) shows weak correlation with the chemical zonation pattern where zones of
125 higher and lower IQ correlate with domains of elevated and lower (Pb+ Σ REY), respectively. It is
126 unclear whether observed variation in IQ is due to chemical variability alone, or to a combination of
127 chemical variability and porosity/inclusion content, since the zones where IQ and chemical variability
128 correlate also have higher porosity/inclusion content.

129 DISCUSSION

130 **Primary versus secondary minor/trace element patterns**

131 The continual production of Pb as a result of radioactive decay, will cause some alteration to the
132 chemistry and structure of uraninite and/or any other U-bearing minerals (Hazen et al., 2009, and
133 references therein). An important, self-induced alteration phenomenon which affects all U-bearing
134 minerals is the accumulation of long-term damage caused by α -decay events, or ‘metamictization’. The
135 most intense damage results in defect production and amorphization as, for example, has been shown
136 for modelling of energetic uranium recoil damage within zircon (Devanathan et al., 2006). In contrast,
137 uraninite is known to be quite resistant to α -recoil events because of relatively rapid annealing kinetics
138 (Eyal and Fleischer, 1985; Janeczek and Ewing, 1991), and has the ability to self-heal radiation
139 damage; the radioactive decay process can also induce redistribution of key elements such as Pb within
140 a given grain (e.g., Hazen et al., 2009).

141 Incorporation of Pb and Σ REY within the crystal lattice was shown for ‘primary’, oscillatory-zoned
142 uraninite which hosts the highest amounts of Pb (up to 0.2 apfu) and Σ REY (up to 0.2 apfu) and thus
143 was defined as the earliest uraninite generation at OD (Macmillan et al., 2016). Although oxidation of
144 U⁴⁺ to U⁶⁺ was calculated to compensate for substitutions and charge balance, no changes to crystal

145 symmetry or lattice defects were found in such highly-substituted uraninite despite prediction of
146 vacancies or other crystal structural modifications (Janeczek and Ewing, 1991). If all measured Pb is
147 assumed as radiogenic, the oscillatory zoning with respect to Pb (and Σ REY) typical of primary
148 uraninite at OD (Macmillan et al., 2016) is a self-induced diffusion patterning mechanism which traps
149 daughter isotopes formed during α -recoil events. Weak oscillatory zoning with respect to (Pb+ Σ REY)
150 is also observed in the Th-U sector zoned uraninite from Mesoproterozoic pegmatite in Southern
151 Norway, where the sector zoning is attributed to primary growth (Alexandre et al., 2015).

152 Zones of comparable low-(Pb+ Σ REY) concentrations with those discussed here for the square-
153 shaped-sectors (< 0.1 apfu for Pb and \sim 0.1 apfu for Σ REY) in the ‘zoned’ type were also reported, but
154 only as incipient sectorial zoning on $\langle h0l \rangle$ or equivalent directions in the ‘primary’ uraninite from OD
155 (Macmillan et al., 2016). Also documented by these authors, were the presence of rare, fine particles of
156 galena in parts of ‘primary’ uraninite grains affected by sub- μ m fractures, infilled with bornite \pm
157 fluorite. Zonation patterns (of Pb) which are oscillatory, sectorial, or a combination thereof, can be the
158 result of element redistribution during the same or sequential self-induced α -recoil ‘dry’ events, and/or
159 as a result of the interaction with fluids of differing chemistry (Cu, S, F) to uraninite. Bornite \pm fluorite
160 inclusions and infill, are more abundant in the ‘zoned’ uraninite, and these can be used to elucidate
161 fluid chemistry. Microstructural analysis is essential in being able to link the observed chemical
162 patterns and heterogeneity with micro- and meso-scale lattice defects that could have assisted ingress of
163 fluids during superimposed geological events.

164 **Lattice distortion and chemical heterogeneity**

165 The microstructural analysis of ‘zoned’ uraninite shows lattice rotation/dislocations tied to
166 preferential slip systems, low-angle boundaries and areas of high-strain. All these microstructures
167 correlate with directions/traces of chemical heterogeneity in the grain, i.e., (Pb+ Σ REY)-zonation, pores,
168 inclusions and (bornite \pm fluorite)-infilled cracks. The correlation between lattice rotation about the

169 $\langle 1\bar{1}0 \rangle$ direction on $\{001\}\langle 1\bar{1}0 \rangle$ slip system and the majority of tilt traces on the IPF map (Fig. 1d)
170 indicates that accumulation of dislocations and pile-up defects are attributable to the low-temperature
171 slip system in UO_2 (Boyle et al., 1998; Kelly et al., 2012). Such a system is active during superimposed
172 alteration since it is controlled by directions different to the grain orientation.

173 Lattice distortion, coincident with directions parallel to $\{201\}$ and $\{021\}$ faces of sector zoning
174 (indicated by dashed lines on Fig. 1d), and also with the orientation of infilled fractures, is highest in
175 areas of high-stored strain accommodated by an increase in the abundance of low-angle boundaries
176 (misorientation profile on the GROD map; Fig. 2). Therefore, fracturing of uraninite could be
177 considered the result of strain hardening by dislocation pile-up along chemical boundaries that impede
178 dislocation glide and prohibit recovery. This is plausible since there is a considerable amount of strain
179 energy stored in the region around a dislocation (Kelly et al., 2012).

180 Any distortions to the crystal lattice within the diffracting volume are recorded by IQ, and can be
181 used as a qualitative indicator of sample microstructure (e.g., variable crystallographic orientations;
182 grain boundaries; chemical variability; impurities; porosity; Reddy et al., 2007). Correlation between
183 chemical heterogeneity and sample microstructure is observed from the diffuse patterns recorded from
184 the (Pb+ Σ REY)-low, high-U sectors on the IQ maps (orange regions in Fig. 3). Such areas should
185 display brighter patterns due to the higher atomic scattering effect produced by heavier elements
186 (Wright and Nowell, 2006), but the presence of μm to sub- μm -scale inclusions/pores as dense fields
187 instead induces diffuse diffraction patterns and thus lowers IQ.

188 All the above support the interpretation that the observed (Pb+ Σ REY)-sector zoning is a result of the
189 removal of these elements from pre-existing uraninite. Zones of structural weakness were formed as a
190 result of the accumulation of defects and dislocations into tilt boundaries, which formed via lattice
191 rotation about the plane normal to the active slip system in uraninite, permitting the ingress of a
192 hydrothermal fluid into uraninite. Where dislocations and defects pile-up (i.e., along active slip

193 systems), high-diffusivity pathways can be formed, aiding element mobility (Reddy et al., 2006).
194 Replacement of uraninite by bornite \pm fluorite occurred along these planes, and these share common
195 crystallographic orientations (both pink/orange in Fig. 1c and d). The same fluids are likely responsible
196 for the presence of other trace elements, such as Ca and As within the (Pb+ Σ REY)-depleted sectors
197 (Macmillan et al., 2016), as well as increased pore/inclusion content. The documentation of low-angle
198 boundaries associated with high strain areas provides evidence for a dynamic rather than a static
199 recovery process. Thus the ability for uraninite to progressively self-anneal radiation damage (in
200 contrast to static temperature-driven annealing processes) may have modified the chemical zoning, but
201 more importantly, the interaction between uraninite and ingressing fluids must have been pivotal in
202 forming the modified zonation patterns observed in ‘zoned’ uraninites.

203 The results here are further evidence that the ‘zoned’ uraninite is a distinct, intermediate stage
204 during *in-situ* transformation of uraninite from ‘primary’ to ‘cob-web’ stages (Macmillan et al., 2016).
205 In the last stage (Fig. A2, Appendix 2) pseudomorphic sulfide replacement of uraninite is more intense
206 and leads to extremely modified forms of ‘zoned’ type, with the microstructural features (i.e., GROD,
207 IQ maps) for ‘cob-web’ uraninite being a variant of those displayed for ‘zoned’ uraninite (Figs. 1 to 3).

208 **IMPLICATIONS AND OUTLOOK**

209 Uraninite at OD has been exposed to a prolonged geologic history and has undergone multiple fluid-
210 rock interaction events at variable flow rates, fluid-pressures, temperatures and rheologic contexts since
211 early stage deposit formation at \sim 1590 Ma (Ciobanu et al., 2013). The interpretation of the evolution of
212 uraninite is important in constraining mineralizing stages at OD and elsewhere. Future U-Pb uraninite
213 geochronology requires a detailed knowledge of the inherent heterogeneity within these uraninites,
214 since these dating methods assume chemical homogeneity at the scale of the microprobe beam. Thus,
215 without combined microchemical and microstructural studies such as this, characterization of the
216 heterogeneity is not possible, and erroneous chemical ages may be attained. This type of work should

217 be applied to other U-bearing minerals featuring comparable chemical-textural complexity such as
218 hematite (Ciobanu et al., 2013), an intrinsic hydrothermal mineral in IOCG deposits.

219 **ACKNOWLEDGEMENTS**

220 This work forms part of the Ph.D. studies of EM and is supported by BHP Billiton. NJC, KE and AP
221 acknowledge support from the ARC Research Hub for Australian Copper-Uranium. Staff at Adelaide
222 Microscopy are thanked for instrument training. We appreciate insightful comments and suggestions from
223 reviewers Mark Pearce, Alan Boyle and Paul Alexandre, and Editor Ian Swainson, which assisted us with
224 revision of the manuscript.

225 **REFERENCES CITED**

- 226 Alexandre, P., Peterson, R. and Joy, B. (2015) Sector zoning in uraninite. *The Canadian Mineralogist*, 53, 1-11.
- 227 Allen, F.M. and Buseck, P.R. (1988) XRD, FTIR, and TEM studies of optically anisotropic grossular garnet.
228 *American Mineralogist*, 73, 568-584.
- 229 Boyle, A.P., Prior, D.J., Banham, M.H. and Timms, N.E. (1998) Plastic deformation of metamorphic pyrite: new
230 evidence from electron backscatter diffraction and foreshatter orientation contrast imaging. *Mineralium*
231 *Deposita*, 34, 71-81.
- 232 Ciobanu, C.L., Wade, B.P., Cook, N.J., Schmidt Mumm, A. and Giles, D. (2013) Uranium-bearing hematite
233 from the Olympic Dam Cu-U-Au deposit, South Australia: A geochemical tracer and reconnaissance Pb-Pb
234 geochronometer. *Precambrian Research*, 238, 129-147.
- 235 Devanathan, R., Corrales, L.R., Weber, W.J., Chartier, A. and Meis, C. (2006) Molecular dynamics simulation
236 of energetic uranium recoil damage in zircon. *Molecular Simulation*, 32, 1069-1077.
- 237 Ehrig, K., McPhie, J. and Kamenetsky, V. (2012) Geology and mineralogical zonation of the Olympic Dam Iron
238 Oxide Cu-U-Au-Ag deposit, South Australia. Special Publication Number 16 - Geology and Genesis of
239 Major Copper Deposits, Society of Economic Geologists, 237-267.
- 240 Eyal, Y. and Fleischer, R.L. (1985) Timescale of natural annealing in radioactive minerals affects retardation of
241 radiation-damage-induced leaching. *Nature*, 314, 518-520.
- 242 Finch, R.J. and Murakami, T. (1999) Systematics and paragenesis of uranium minerals. In P.C. Burns and R.J.

- 243 Finch, Eds., Uranium: Mineralogy, Geochemistry and the Environment, 38, p. 91-179. Reviews in
244 Mineralogy and Geochemistry, Mineralogical Society of America, Chantilly, Virginia.
- 245 Hazen, R.M., Ewing, R.C. and Sverjensky, D.A. (2009) Evolution of uranium and thorium minerals. American
246 Mineralogist, 94, 1293-1311.
- 247 Janeczek, J. and Ewing, R.C. (1991) X-ray powder diffraction study of annealed uraninite. Journal of Nuclear
248 Materials, 185, 66-77.
- 249 Kelly, A.A., Knowles, K.M. and Kelly, A. (2012) Crystallography and crystal defects. 2nd ed., John Wiley and
250 Sons, Chichester, West Sussex, UK.
- 251 Macmillan, E., Cook, N.J., Ehrig, K., Ciobanu, C.L. and Pring, A. (2016) Uraninite from the Olympic Dam
252 IOCG-U-Ag deposit: linking textural and compositional variation to temporal evolution. American
253 Mineralogist, 101, 1295-1320.
- 254 Prior, D.J., Boyle, A.P., Brenker, F., Cheadle, M.C., Day, A., Lopez, G., Peruzzo, L., Potts, G.J., Reddy, S.,
255 Spiess, R., Timms, N.E., Trimby, P., Wheeler, J. and Zetterstrom, L. (1999) The application of electron
256 backscatter diffraction and orientation contrast imaging in the SEM to textural problems in rocks. American
257 Mineralogist, 84, 1741-1759.
- 258 Prior, D.J., Wheeler, J., Peruzzo, L., Spiess, R. and Storey, C. (2002) Some garnet microstructures: an
259 illustration of the potential of orientation maps and misorientation analysis in microstructural studies. Journal
260 of Structural Geology, 24, 999-1011.
- 261 Reddy, S.M., Timms, N.E., Trimby, P., Kinny, P.D., Buchan, C. and Blake, K. (2006) Crystal-plastic
262 deformation of zircon: A defect in the assumption of chemical robustness. Geology, 34, 257-260.
- 263 Reddy, S.M., Timms, N.E., Pantleon, W. and Trimby, P. (2007) Quantitative characterization of plastic
264 deformation of zircon and geological implications. Contributions to Mineralogy and Petrology, 153, 625-645.
- 265 Shore, M. and Fowler, A.D. (1996) Oscillatory zoning in minerals: a common phenomenon. Canadian
266 Mineralogist, 34, 1111-1126.
- 267 Steffen, K.J. and Selverstone, J. (2006) Retrieval of P-T information from shear zones: thermobarometric
268 consequences of changes in plagioclase deformation mechanisms. Contributions to Mineralogy and
269 Petrology, 151, 600-614.

270 Urai, J.L., Means, W.D.M. and Lister, G.S. (1986) Dynamic recrystallisation of minerals. In B.E. Hobbs and
271 H.C. Heard, Eds., Mineral and Rock Deformation: Laboratory Studies - The Paterson Volume, Geophysical
272 Monograph 36, p. 161-199. American Geophysical Union.

273 Wright, S.I. and Nowell, M.M. (2006) EBSD image quality mapping. *Microscopy and Microanalysis*, 12, 72-84.

274 **FIGURE CAPTIONS**

275 Fig. 1: **(a, b)** EPMA-WDS maps showing Pb and Ce zonation in uraninite (Urn; adapted from Macmillan et al.,
276 2016), with cracks infilled by bornite (Bn) and fluorite (Fl). Red-dotted lines demarcate (Pb+ Σ REY)-poor
277 sectors); **(c-d)** EBSD-derived $\langle 001 \rangle$ Inverse Pole Figure (IPF) map of UO_2 with superimposed tilt boundaries
278 formed by a lattice rotation about the plane normal to $\{111\}\langle 1\bar{1}0 \rangle$ in (c) and $\{001\}\langle 1\bar{1}0 \rangle$ in (d). UO_2 is of one
279 dominant orientation (close to $\langle 11\bar{2} \rangle$ zone axis) and represents a single grain with some gradational color
280 variation reflecting slight distortion of crystal lattice. White- and black-dotted circled areas highlight two
281 examples of where the tilt boundaries correlate to Ce zonation boundaries (Fig. 1b and d, respectively). Mean
282 compositions (wt%) of low-Pb zones: 75.5 UO_2 , 6.7 PbO_2 , 3.4 CaO , 7.1 $\Sigma\text{REY}_2\text{O}_3$, 1.2 As_2O_3 ; high-Pb zones:
283 69.8 UO_2 , 14.9 PbO_2 , 1.6 CaO , 11.0 $\Sigma\text{REY}_2\text{O}_3$, 0.3 As_2O_3 (as reported in Macmillan et al., 2016).

284 Fig. 2 **(a)** EBSD-derived Grain Reference Orientation Deviation (GROD) map of UO_2 with superimposed $<10^\circ$
285 low-angle grain boundaries (solid black lines). Each pixel is colored from reference orientation (blue, defined
286 by white cross) with misorientation of up to 18° (red). Misorientation profile (Fig. 2b) plotted along section
287 A-B; **(b)** EBSD-derived misorientation profile displaying the highest misorientation ($\sim 9^\circ$) correlates to zones
288 of higher strain (yellow on Fig. 2a).

289 Fig. 3: **(a)** EBSD-derived Image Quality (IQ) map of entire uraninite grain with superimposed tilt boundaries
290 formed by a lattice rotation about the plane normal to $\{001\}\langle 1\bar{1}0 \rangle$. Warmer colors (red-orange) represent
291 areas of higher IQ (uraninite), and cooler colors (blue-green) represent areas of lower IQ (bornite and
292 fluorite). Orange colored areas (uraninite) represent low-Pb uraninite with higher inclusion content whereas
293 red colored areas represent high-Pb uraninite with lower inclusion content. Black dotted circled regions are
294 further enlarged in (c) and (d), and area outlined with black rectangle is imaged in (b); **(b)** BSE image of
295 uraninite grain with reduced brightness and contrast to highlight increased Pb/ Σ REY and reduced porosity
296 towards rim (red dashed lines); **(c-d)** EBSD-derived Image Quality (IQ) maps of UO_2 (colored as for (a)).
297 The uraninite regions with lower porosity and elevated Pb \pm Σ REY have higher IQ (red) than zones with lower
298 porosity (orange).

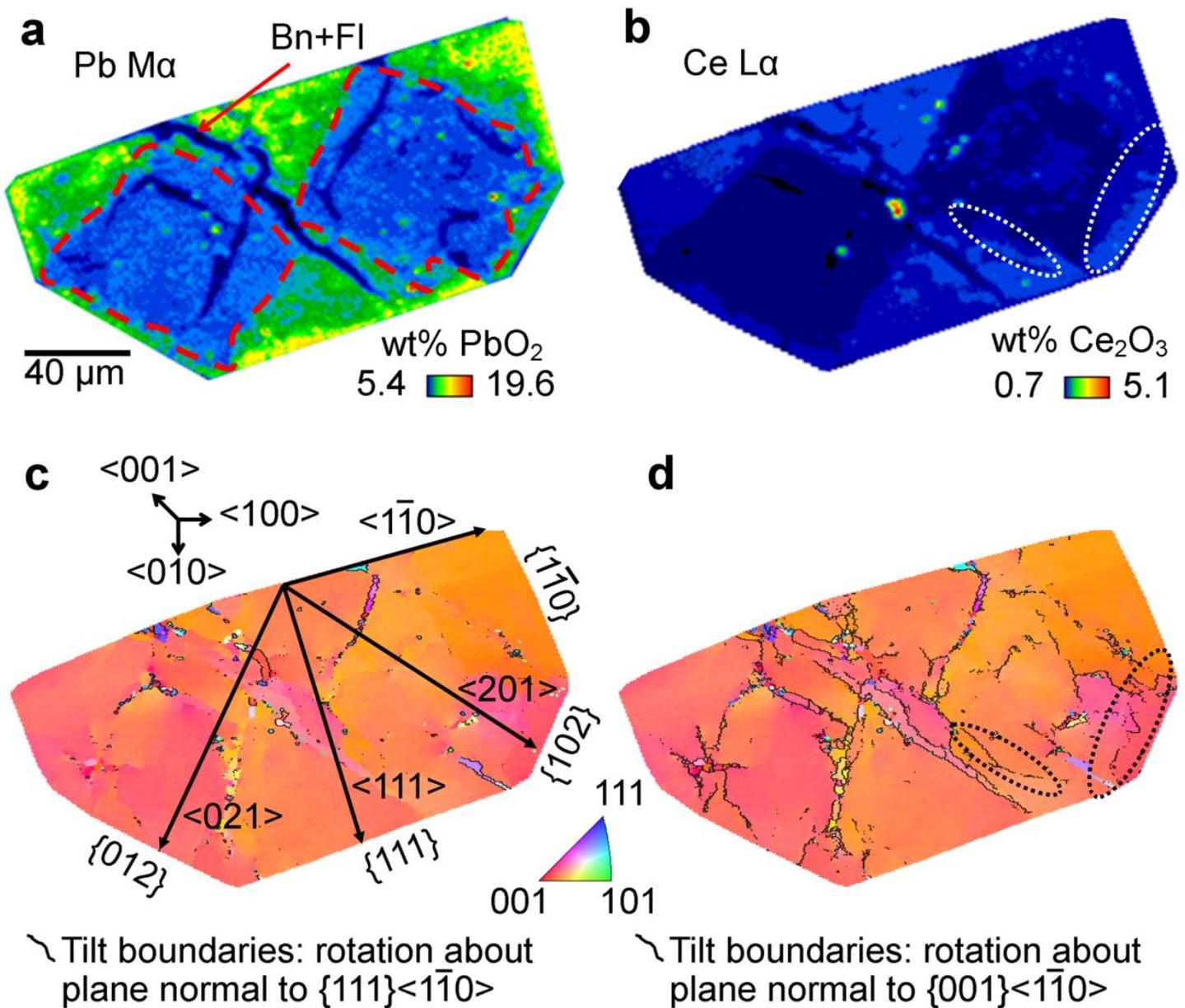


Figure 1 (Macmillan et al.)

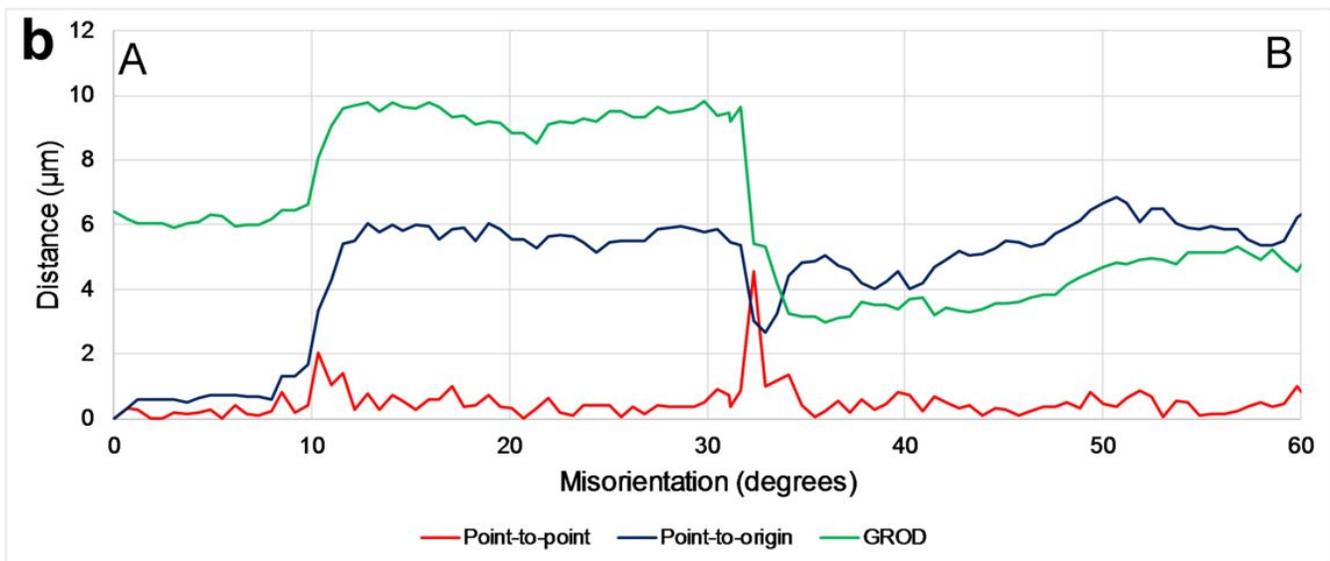
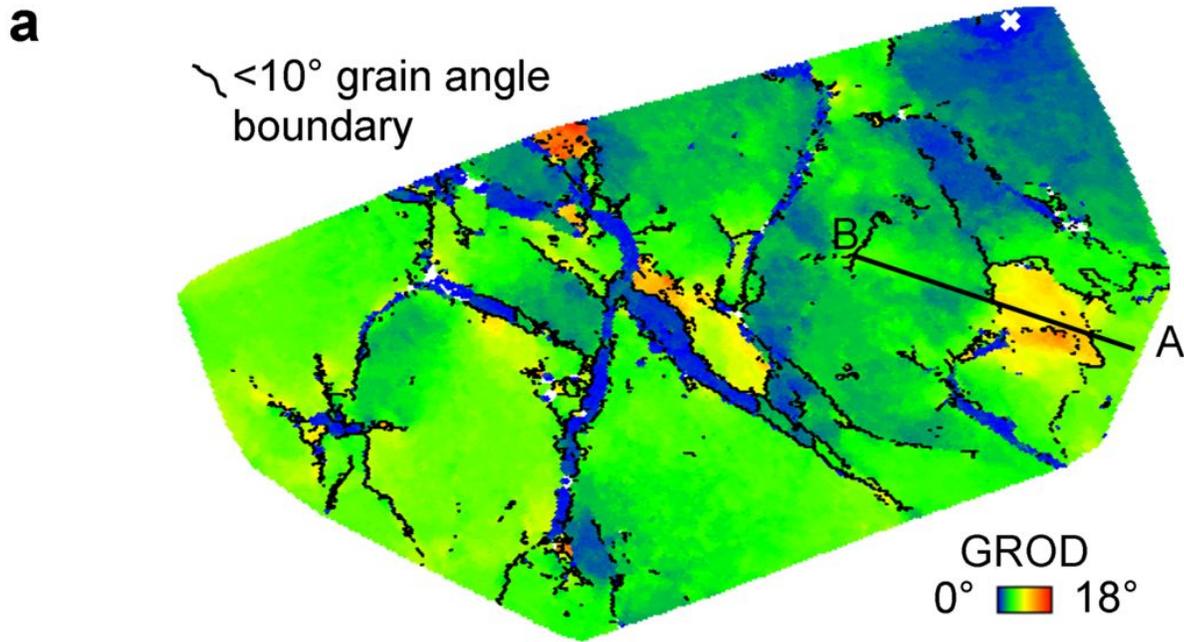


Figure 2 (Macmillan et al.)

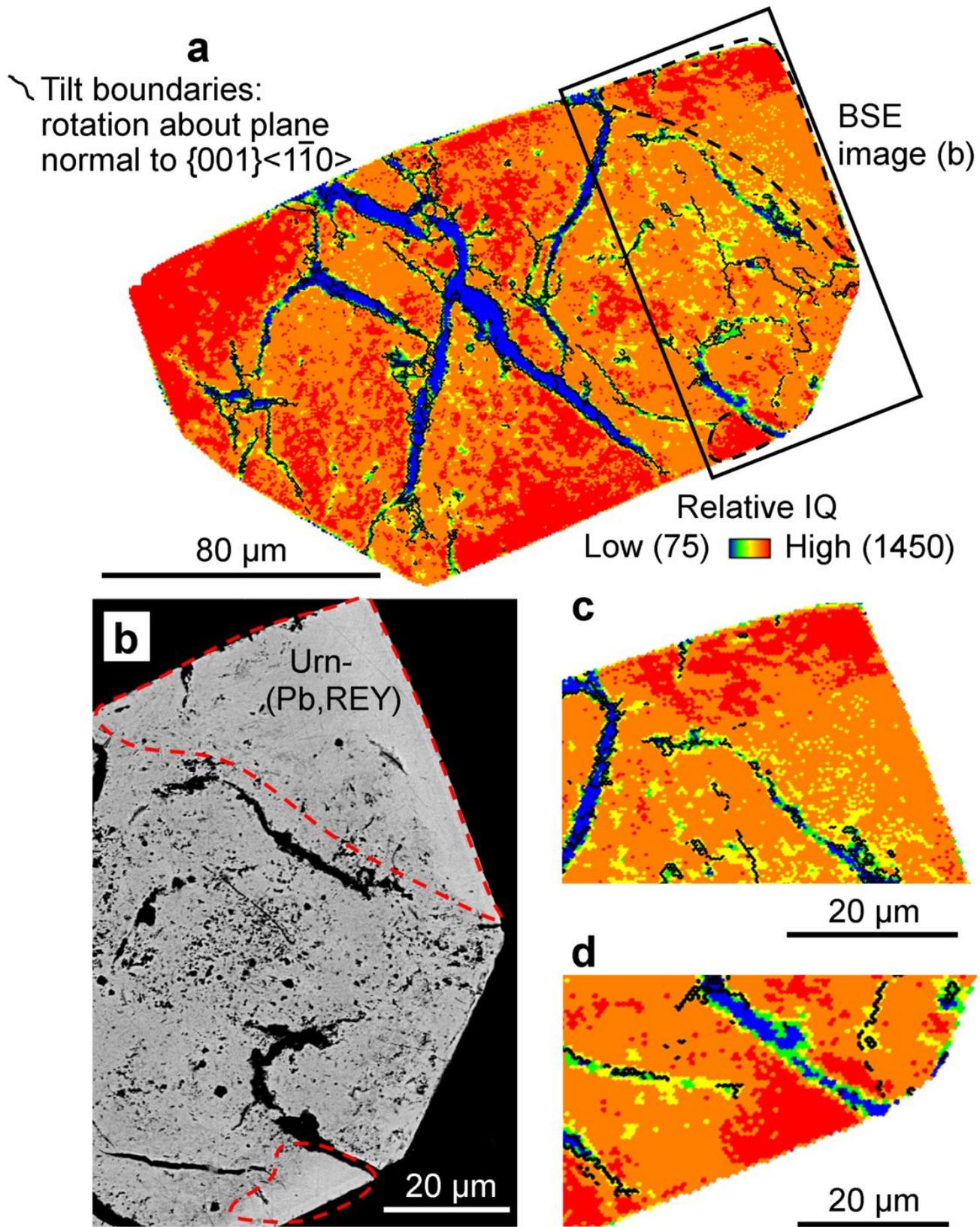


Figure 3 (Macmillan et al.)