

1 **Revision 1**

2

3 **Combined Fe-Mg chemical and isotopic zoning in olivine constraining magma**  
4 **mixing-to-eruption timescales for the continental arc volcano Irazú (Costa Rica)**  
5 **and Cr diffusion in olivine**

6

7 **Martin Oeser**<sup>1,\*</sup>, **Philipp Ruprecht**<sup>2,3</sup>, and **Stefan Weyer**<sup>1</sup>

8 <sup>1</sup> Leibniz Universität Hannover, Institut für Mineralogie, Callinstr. 3, 30167 Hannover, Germany

9 <sup>2</sup> University of Nevada, Department of Geological Sciences, 1664 N. Virginia Street, Reno,  
10 NV 89557, USA.

11 <sup>3</sup> Columbia University, Lamont-Doherty Earth Observatory, 61 Route 9W, Palisades, NY, USA

12

13 \* corresponding author

14

15 email addresses:

16 Martin Oeser: [m.oeser@mineralogie.uni-hannover.de](mailto:m.oeser@mineralogie.uni-hannover.de)

17 Philipp Ruprecht: [pruprecht@unr.edu](mailto:pruprecht@unr.edu)

18 Stefan Weyer: [s.weyer@mineralogie.uni-hannover.de](mailto:s.weyer@mineralogie.uni-hannover.de)

19

20

### Abstract

21 Arc magmas commonly are mixtures of newly-arriving primitive melts, stored magmas at shallow  
22 levels, and xenolithic material added on ascent. Almost every eruption has a unique assembly of  
23 these components, which may record magmatic processes occurring in the plumbing system prior  
24 to an eruption. In this study, we focus on complexly zoned olivines (crystal xenocrysts) in order to  
25 obtain a better understanding of the magmatic processes and the assembly of the 1963-65 erupted  
26 magmas of Irazú volcano, one of the most voluminous active volcanoes in Costa Rica. We  
27 performed high-precision in-situ Fe-Mg isotope analyses by femtosecond-LA-MC-ICP-MS on  
28 these olivines, in order to unravel the origin of their complex chemical zoning (growth, diffusion,  
29 or a combination of both processes). This information was used to establish a refined diffusion  
30 model to explore magma mixing-to-eruption timescales. Furthermore, trace element analyses using  
31 LA-ICP-MS were performed. Chromium displays a chemical zoning in the investigated olivine,  
32 which coincides spatially as well as in terms of length scale and geometry with Fe-Mg zoning and  
33 which was used to constrain Cr diffusivity in natural olivine.  
34 Our findings show that Fe-Mg zoning in Irazú olivine mainly results from Fe-Mg inter-diffusion  
35 after two crystal growth episodes as indicated by strongly coupled chemical and isotopic zoning.  
36 Simulations of this diffusive process indicate that mixing of these crystals into ascending primitive  
37 melts occurred <600 days before their eruption, consistent with a previously reported diffusion  
38 study based on Ni zonation in Mg-rich olivines. Trace element characteristics of olivine suggest  
39 that the complex-zoned olivine crystals originate from a crystal mush/cumulate in the middle or  
40 lower crust and deeper than the shallow magma chamber and were mobilized by mantle-derived  
41 magma bearing Mg-rich olivines. Finally, modeling of the observed Cr zoning in the Irazú olivines  
42 indicates that the diffusion coefficient for Cr in olivine ( $D_{Cr}$ ) is smaller than  $D_{Fe-Mg}$  by a factor of

43  $4.9 \pm 2.9$  at the conditions experienced by these crystals consistent with Cr diffusion experiments at  
44 high silica activity in the melt.

45 Our results show that by combining elemental and isotope zoning studies in individual minerals we  
46 can refine the timing/assembly of magmatic eruptions and provide independent constraints on  
47 element diffusivities. Lastly, it confirms that primitive arc magmas at Irazú are not aphyric during  
48 ascent, but carry primitive phenocrysts from lower crust or Moho depth to the surface.

49

50 **Keywords:** olivine; Fe-Mg zoning; stable isotopes; laser ablation; diffusion modeling; Cr  
51 diffusivity; magma assembly

52

53

54

55

### Introduction

56 Intermediate arc magmas frequently contain diverse crystal populations, many of which are not in  
57 equilibrium with the host melt. As a result many crystals re-equilibrate with their host melt, either  
58 through chemical diffusion, crystallization/dissolution or a combination of these processes.

59 Snapshots of these transient processes, such as chemical zoning in crystals, are frozen at the time of  
60 eruption when magmas cool down rapidly. If diffusion is the dominant process to achieve  
61 equilibrium between crystal and melt, the chemical zoning can be used to obtain time information  
62 about the evolution of magmatic systems by diffusion modeling, provided that diffusion rates of  
63 elements in the minerals of interest are known. The diffusion of Fe and Mg in olivine has been  
64 intensely investigated in experimental studies for more than 30 years (reviewed by Chakraborty  
65 2010), and several studies have used Fe-Mg chemical gradients in magmatic olivine crystals in  
66 order to estimate timescales of magma evolution processes by diffusion modeling (e.g. Costa and

67 Chakraborty 2004; Costa and Dungan 2005; Kahl et al. 2011; Kahl et al. 2013; Hartley et al. 2016;  
68 Rae et al. 2016), relying on the parameterization for the Fe-Mg inter-diffusion coefficient given by  
69 Chakraborty (1997) and Dohmen and Chakraborty (2007). However, as crystal growth and  
70 elemental diffusion often show similar zoning patterns it remains questionable whether calculated  
71 diffusion timescales represent the timing of a specific magmatic process (Shea et al. 2015). As  
72 shown in recent studies, combining the information of Fe-Mg chemical variations and in-situ  
73 analyses of Fe-Mg stable isotope ratios in olivine represents a powerful means to elucidate  
74 diffusion and/or growth/dissolution processes (Sio et al. 2013; Oeser et al. 2015). As diffusion  
75 generates large kinetic isotope fractionations (e.g. Richter et al. 2003) diffusion-generated chemical  
76 zoning is coupled with pronounced isotopic zoning, in contrast to chemical zoning that developed  
77 in absence of diffusion (Oeser et al. 2015). Detailed chemical and isotopic profiles across  
78 magmatic olivines may thus be used to unravel complex crystal growth- and diffusion histories,  
79 which are indistinctly recorded by chemical zoning alone (e.g. Collinet et al. 2017; Sio and  
80 Dauphas 2017).

81 In this study, we aim to investigate the magmatic events recorded in complexly zoned olivine  
82 crystals in basaltic andesites of the 1963-65 eruption of the Irazú volcano, one of the most  
83 voluminous (e.g. Carr 1984) and active volcanoes in Costa Rica, with recurring explosive andesitic  
84 eruptions potentially endangering ~1.6 million people living within 30 km of this stratovolcano  
85 (Global Volcanism Program 2013). Previously, investigation of Ni zoning in primitive (magnesian)  
86 olivines from this eruption suggested that magma transport from the Moho to the surface occurred  
87 on timescales comparable to the duration of the eruption (~2 years), thus indicating that the  
88 eruption was likely fed (and triggered) by magmas extracted from the mantle (Ruprecht and Plank  
89 2013). This study provides a more comprehensive picture of the pre-eruptive mixing and  
90 assembling of erupted 1963-65 magmas, focusing on complexly zoned (in terms of Fo contents)

91 olivines. These crystals share the normal zoning history of the rims of the mantle-derived  
92 magnesian olivines. By unraveling the origin of their complex zoning we establish the basis for  
93 appropriate diffusion modeling. This will eventually yield reliable mixing-to-eruption timescales  
94 and provide a more detailed picture of the complex history of these crystals, leading to a better  
95 understanding of the magmatic processes and the assembly of the erupted magmas beneath Irazú  
96 volcano.

97 Moreover, through the refinement of timescale estimates using in-situ isotope records we can also  
98 assess the diffusivity of slow-diffusing elements, for which experimental studies are challenging  
99 (e.g. Milman-Barris et al. 2008). Natural samples have been used elsewhere (e.g. Mallmann et al.  
100 2009; Qian et al. 2010; Tollan et al. 2015) to compare results to experimentally determined  
101 diffusion coefficients (e.g. Ito and Ganguly 2006; Spandler and O'Neill 2010; Zhukova et al. 2017;  
102 Jollands et al. 2014, 2018). Notably, Cr diffusivities display a large range in the literature (Ito and  
103 Ganguly 2006; Milman-Barris et al. 2008; Spandler and O'Neill 2010; Jollands et al. 2018). The  
104 recent study of Jollands et al. (2018) suggested that this large range is a result of variations in SiO<sub>2</sub>  
105 activity ( $a_{\text{SiO}_2}$ ), an effect that has also been recognized for other trace elements in olivine (Jollands  
106 et al. 2014, 2016a; Zhukova et al. 2014). Independent timescale estimates (specifically more robust  
107 by combining elemental and isotopic zoning) can also be used to constrain diffusivities of such  
108 slow-diffusing elements and help to provide additional constraints to reconcile variations in  
109 experimentally determined diffusivities. The complexly zoned olivine crystals in basalts of the  
110 1963-65 eruption that are close to the SiO<sub>2</sub> activity buffer of olivine-enstatite are suitable for such  
111 an approach since Fe-Mg zoning coincides with Cr zoning (and to a lesser extent also V and Al  
112 zoning).

113  
114

115 **The Irazú magma system and samples**

116 The Irazú magmatic plumbing system has been characterized geophysically and geochemically  
117 (Alvarado et al. 2006; Dzierma et al. 2010; Ruprecht and Plank 2013). Seismic studies highlight a  
118 low-velocity region in the upper crust (5-10 km) commonly interpreted to represent the major  
119 magma storage region. Furthermore, the Moho depth has been constrained seismically to ~35-45  
120 km (Lizarralde et al. 2010; Gazel et al. 2015).

121 Geochemical constraints emphasize the hybrid nature of Irazú magmas, particularly the 1963-65  
122 eruption (Alvarado et al. 2006). Whole-rock compositions represent mixtures of basaltic and  
123 dacitic end-members. While geophysics and whole-rock geochemistry provide a simplified view of  
124 the magma plumbing system, more detail is recorded within the crystal cargo. Firstly, crystals from  
125 individual samples show a large compositional diversity that mimics the whole rock end members  
126 (An<sub>40-80</sub>, Fo<sub>72-91</sub>; Alvarado et al. 2006; Ruprecht and Plank 2013). However, a detailed look into  
127 the crystal cargo reveals that it is inadequate to explain the complex crystal cargo with mantle-  
128 derived magmas that simply rise to the shallow region where they differentiate and hybridize. In  
129 this study we focus on olivine crystals (with xenocrystic cores; see below) that provide a unique  
130 view into the magma assembly from the mantle to the surface. Ruprecht and Plank (2013) reported  
131 on the major and trace element geochemical diversity in olivines from the 1963-65 eruption. Based  
132 on their results, at least three different olivine types in the Irazú eruption can be distinguished: 1)  
133 magnesian olivines (Fo<sub>89-91</sub>) which were the focus of Ruprecht and Plank (2013), type A; 2)  
134 complex olivines that show significant reversals in their Fo contents and are the focus of this study,  
135 type B; and 3) intermediate to low Fo olivines at low Ni content that represent shallow crustal  
136 processes of fractionation and mixing (type C; Fig. 1). It is notable here, that these three types of  
137 olivines represent end-member compositions, and the full olivine population shows crystals that are  
138 intermediate between those end-members. Potentially, the “reservoirs” that are represented by the

139 three olivine end-members have either some internal variation, e.g., due to incomplete mixing  
140 (Ruprecht et al. 2008; Schleicher and Bergantz 2017) resulting in a range in composition, or they  
141 are partially open to interact with each other beyond just the processes directly driving the 1963-65  
142 eruption. In the “Results” section we are going to describe in greater detail the type B olivines,  
143 which are the main focus of this study. The samples analyzed herein (IZ-10-11/-12/-13) comprise  
144 crater rim deposits of the 1963-65 eruption on the SW side of the major Irazú crater which are  
145 juvenile scoria-rich tephra and fine lapilli tephra (Ruprecht and Plank 2013). From these samples,  
146 we investigated 10 olivine crystals (three of type A, seven of type B) for their major and trace  
147 elements characteristics and their Fe- and Mg isotopic compositions by microanalytical techniques  
148 (see next section).

149

150

151

## Methods

### **In-situ major and trace element analyses of olivines**

152 Major and trace element laser ablation data were analyzed using a VG PQ ExCell and NewWave  
153 196 nm ArF Eximer laser at Lamont-Doherty Earth Observatory during the study of Ruprecht and  
154 Plank (2013). The data acquisition was done in line mode and the continuous time series data was  
155 then processed into individual data points by combining a finite number of steps. Each sweep from  
156 Li-7 to Y-89 took 0.36 s, with dwell times of 10 ms on each mass (for all masses analyzed see  
157 Table S1). The raw data was processed following standard calibration techniques for LA-ICPMS  
158 (using a Matlab code developed by P. Ruprecht), but using Fe and Mg count rates and olivine  
159 stoichiometry to calculate the internal standard concentration of MgO. Comparisons with  
160 microprobe profiles show that for magnesian olivines this procedure produces Fo contents within  
161 0.5 mol units of electron microprobe results (Ruprecht and Plank 2013). San Carlos olivine San  
162

163 Carlos USNM 1113142/44 was used to normalize results from different days and to assess the  
164 measurement uncertainties. While this procedure may introduce a small systematic difference to  
165 other datasets in the literature related to potentially small heterogeneities in San Carlos trace  
166 element abundances across labs, it ensures that analyses are internally comparable. The  
167 concentrations used for San Carlos olivine were reported in Ruprecht and Plank (2013) and  
168 recalculation to other standards is trivial. A more detailed San Carlos olivine study is underway,  
169 but it is not the scope of this paper. The combined data points are recorded for every 5.8  $\mu\text{m}$ , which  
170 is less than the spot size of the laser (25  $\mu\text{m}$ ) and therefore does not represent the true spatial  
171 resolution of the analysis. Given the continuous movement across the sample (3  $\mu\text{m/s}$ ), some  
172 smearing of the signal will occur and the spatial resolution is therefore similar to the laser spot size  
173 (Bradshaw and Kent 2017). Recently we recognized that in our instrument Ca concentration data  
174 are affected by interference of  $^{28}\text{Si}$  and  $^{16}\text{O}$  on mass 44, thus the Ca data needs to be interpreted  
175 cautiously. We focus here on inferred Fo content as well as the cations V, Cr, and Al. Al counts  
176 were statistically indistinguishable from the background for one sample (IZ-10-13), which was  
177 measured on a different day than IZ-10-12, which was measured during a day of better instrument  
178 sensitivity. LA-ICP-MS results of the olivines investigated in this study can be found in the  
179 supplementary material (Table S1). Olivines were repolished after these LA-ICP-MS analyses in  
180 order to determine crystal orientations by electron backscatter diffraction (see below) and to  
181 acquire BSE images.

182

### 183 **In-situ Fe and Mg isotope analyses of zoned olivines**

184 Profiles of Fe and Mg isotopic compositions in zoned olivines were acquired in-situ by  
185 femtosecond-laser ablation-MC-ICP-MS (fs-LA-MC-ICP-MS) at the Institut für Mineralogie of the  
186 Leibniz Universität Hannover, following the analytical procedures described in Oeser et al. (2014)



187 and Oeser et al. (2015). In summary, a Spectra-Physics Solstice fs-LA system is coupled to a  
188 Thermo-Finnigan Neptune Plus MC-ICP-MS, which is operated in high mass resolution mode in  
189 order to resolve molecular interferences of argon nitrides and argon oxides on Fe isotopes and  
190 isobaric interferences (e.g.  $^{52}\text{Cr}^{2+}$ ) on Mg isotopes. Laser repetition rate was between 30 Hz and 70  
191 Hz during Fe isotope analyses and between 20 Hz and 45 Hz during Mg isotope analyses,  
192 depending on the Fe and Mg concentrations in the ablated material. Fluence of the fs-LA system in  
193 Hannover is on the order of 1 J/cm<sup>2</sup>. Further details of the fs-LA system can be found in Oeser et  
194 al. (2014) and Lazarov and Horn (2015). The instrumental mass discrimination of the MC-ICP-MS  
195 was monitored during Fe isotope ratio determination by simultaneously analyzing the Ni isotope  
196 ratios of a Ni reference solution (NIST SRM 986) combined with a sample-standard bracketing  
197 protocol. For Mg isotope analyses, only the sample-standard bracketing technique was used to  
198 correct for drifts of the instrumental mass bias. With this analytical procedure, the LA analyses  
199 yield a long-term reproducibility of  $\pm 0.13\%$  (2 SD) for both  $\delta^{56}\text{Fe}$  and  $\delta^{26}\text{Mg}$ , based on replicate  
200 analyses of silicate reference glasses (e.g. BHVO-2G, ML3B-G) over a period of three years.  
201 During a single analytical session (1-2 days) the reproducibility of  $\delta^{56}\text{Fe}$  and  $\delta^{26}\text{Mg}$  values for the  
202 reference glasses is always better than  $\pm 0.10\%$  (2 SD). The LA analyses on olivines from Irazú  
203 were guided by previously acquired BSE images (Ruprecht and Plank 2013). Sample material was  
204 ablated with a laser spot size of 35-40  $\mu\text{m}$  (in diameter) by line scans either (i) parallel to the  
205 crystal rim (i.e. parallel to the chemical zoning; Fig. 2a and 2b) with a laser scan speed of 20  $\mu\text{m/s}$   
206 and a cycle integration time of  $\sim 1$  s or (ii) perpendicular to it with a laser scan speed of 2-3  $\mu\text{m/s}$   
207 and a cycle integration time of  $\sim 0.5$  s (Fig. 2c and 2d). The latter technique was applied if the  
208 exposed area on the crystal was not sufficient to perform line scans (with sizes of  $\sim 150 \times 40$   $\mu\text{m}$ )  
209 parallel to the chemical zoning. In case (i), one ablation line yields one  $\delta$ -value. In case (ii), the  
210 mean of 15-20 individual ratio measurements is used to calculate one  $\delta$ -value. As a consequence,

211 the precision ( $2\sigma$ ) of a single  $\delta$ -value determined in such a way is usually not as high as for case (i),  
212 i.e. better than  $\pm 0.10\%$ , but still better than the long-term reproducibility of  $\pm 0.13\%$ . For case (ii),  
213  $2\sigma$  is calculated by propagating the errors (SE) of the bracketing standards ( $n \approx 120$  cycles) and of  
214 the analyzed sample ( $n = 15$ -20 cycles), as follows:

$$215 \quad 2\sigma = 2 \times 1000 \times \left[ \left( \frac{SE_{std1}}{R_{std1}} \right)^2 + \left( \frac{SE_{std2}}{R_{std2}} \right)^2 + \left( \frac{SE_{sample}}{R_{sample}} \right)^2 \right]^{0.5} \quad (1)$$

216 where  $R$  is either  $^{56}\text{Fe}/^{54}\text{Fe}$  or  $^{26}\text{Mg}/^{24}\text{Mg}$ .

217

### 218 **Electron microprobe and electron-backscattered diffraction analyses**

219 After the in-situ isotope ratio analyses by fs-LA-MC-ICP-MS, all olivine crystals were carefully  
220 repolished, thus removing a layer of  $\sim 10 \mu\text{m}$  in thickness. Because the laser ablation lines have  
221 approximately a Gaussian shape (in cross-section; Fig. S1 and S2 in the supplementary material),  
222 this repolishing decreased the sizes of the LA lines significantly, enabling us to determine major  
223 and minor element concentrations by electron microprobe analyses (EMPA) very close to the  
224 isotopic profiles.

225 A Cameca SX-100 with 5-wavelength dispersive spectrometers at the American Museum of  
226 Natural History in New York was used for post-isotope analysis electron microprobe work.

227 Analyzes were performed utilizing a focused beam ( $1 \mu\text{m}$  diameter), 20 nA beam current, and 15  
228 kV accelerating voltage. The instrument was calibrated on forsterite (Mg (30 s, peak counting  
229 times), TAP (analyzer crystal)), plagioclase (Al (60 s), TAP), clinopyroxene (Si (30 s), TAP; Ca  
230 (80 s), PET), magnesiochromite (Cr (40 s), LPET), rhodonite (Mn (40 s), LPET), fayalite (Fe (30  
231 s), LLIF), and Ni-diopside (Ni (60 s), LLIF). Reference olivines San Carlos and OI174.1 ( $n=5$   
232 each) were measured together with unknowns. We estimate the reproducibility from the repeat  
233 measurements of reference olivines to be within  $\sim 2\%$  (2SD relative),  $<5\%$ ,  $<20\%$ , and  $<50\%$  for

234 major elements, Ni, Mn, and Ca, respectively. Calculated Mg# for San Carlos ( $89.79 \pm 0.09$ , 2SD  
235 absolute) and Ol174.1 ( $90.24 \pm 0.33$ ) are within error of published values (White 1966; Jarosewich  
236 et al. 1980). Cr and Al in reference materials were below the detection limits for this analytical  
237 protocol ( $\sim 200$  ppm), but analyzed to identify mixed analyses with spinel. A complete table of  
238 quantitative electron microprobe analyses together with BSE images can be found in the  
239 supplementary material (Table S2).

240 To account for diffusive anisotropy crystal lattice orientations relative to the 1D-zoning traverses  
241 were determined via electron-backscattered diffraction (EBSD). EBSD measurements were  
242 performed as part of the analytical session reported in Ruprecht and Plank (2013). A Zeiss Supra  
243 40VP scanning electron microscope, Oxford energy-dispersive X-ray spectroscopy and HKL  
244 electron back-scatter diffraction (EBSD) integrated system was used at the Marine Biological  
245 Laboratory in Woods Hole. Diffraction patterns were matched against a reference olivine lattice  
246 model and Euler angles were recorded for a rotation from the reference frame to the observed  
247 orientation. Rotation information is provided in the supplementary material (Table S3).

248

249

250

## Results

251 As described above, the Irazú 1963-65 samples investigated by Ruprecht and Plank (2013) and in  
252 this study contain principally three different olivine types. We define type B olivines as large  
253 crystals ( $>500 \mu\text{m}$ ) with intermediate Fo core ( $\text{Fo}_{80-87}$ ). Reverse zoning from the core to the mantle  
254 of the crystals links the crystallization history of type A and type B olivines (Fig. 1). The high Fo  
255 mantles never reach the most magnesian compositions recorded in type A olivines, but join the  
256 normal zonation near the rim at  $\sim \text{Fo}_{87}$  to  $\text{Fo}_{90.5}$ . The mantle zoning of major elements in type A  
257 and B is correlated with the same zoning features in minor and trace elements (e.g. Ni, Cr, Al, V;

258 Fig. 3). In contrast, the cores of these two types are significantly different also in trace element  
259 inventory (Fig. 4). Exemplary for this difference are the elevated Ni and low Cr concentrations, a  
260 trace element signature that is consistent with a xenocrystic origin for the type B cores. Such  
261 signatures are commonly associated with olivine cumulates (e.g. Otamendi et al. 2016), where  
262 partitioning and diffusive exchange of Ni and Cr leads to a preferential incorporation in olivine and  
263 pyroxenes, respectively.

264 Given the zoning characteristics of the different olivines in the 1963-65 eruption, exploring their  
265 relative contributions to individual samples shows that the crystal cargo changed as the eruption  
266 progressed. We have only investigated three samples (IZ-10-11/-12/-13) that represent early,  
267 middle, and late stages of the eruption (see Ruprecht and Plank, 2013, for the sampling levels and  
268 further sample details). Some coexisting lavas also were investigated petrographically (samples  
269 from Benjamin et al. 2007), but have not been investigated as detailed as the tephra samples. While  
270 all samples contain type A and C olivines at fractions in significant quantities, type B olivines are  
271 rare in the early sample (IZ-10-11).

272 More specifically, the seven complexly zoned olivine crystals from Irazú volcano investigated in  
273 this study show reversely zoned ferrous cores ( $Fe_{80-87}$  to  $Fe_{87-90}$ , Fig. 5, Fig. S3), likely indicating  
274 (at least) two episodes of crystal growth. The rims of these crystals are always normally zoned  
275 ( $Fe_{87-90}$  to  $Fe_{79-84}$ , Fig. 5). Hence, as mentioned above, they appear to share the history of the rims  
276 of primitive magnesian olivines in the samples of the 1963-65 eruption of Irazú volcano. This  
277 normal chemical zoning at the rims may indicate either (i) crystal growth during fractional  
278 crystallization of olivine and clinopyroxene (and thus an evolution of the melt to higher Fe-Mg  
279 ratios) or (ii) diffusion of Fe into and Mg out of olivine driven by a compositional contrast between  
280 olivine and melt or (iii) a combination of both processes. As shown in previous studies, Fe-Mg  
281 isotopic profiles in such chemically zoned olivines can unravel which of the three possibilities is

282 the most likely one (Sio et al. 2013; Oeser et al. 2015; Richter et al. 2016; Sio and Dauphas 2017).  
283 Other means to distinguish between growth- and diffusion-generated zoning in a crystal include,  
284 for example, the investigation (i) of zoning profile lengths of various elements with different  
285 known diffusivities (e.g. Costa and Dungan 2005), and (ii) of the width of the chemical zoning in  
286 different crystallographic directions expected from diffusion anisotropy (e.g. Costa et al. 2008).  
287 Three solely normally zoned (Mg-rich) olivine crystals investigated in this study show more or less  
288 homogeneous cores (Fo<sub>88-90</sub>) and strongly zoned rims (Fo<sub>88</sub> to Fo<sub>78-80</sub>, Fig. 6). One of these olivines  
289 (IZ-10-13 ol22) additionally shows reverse Ni zoning in its core, which has already been  
290 investigated by Ruprecht and Plank (2013).  
291 Our in-situ Fe-Mg isotope data reveal a strong coupling of Fe-Mg chemical and isotopic zoning for  
292 all investigated olivines (Fig. 5 and 6, Fig. S3), along with largely negative correlations between  
293  $\delta^{26}\text{Mg}$  and  $\delta^{56}\text{Fe}$  values (Fig. 7), indicating that Fe-Mg inter-diffusion is the dominant process  
294 responsible for the re-equilibration of these olivines (cf. Teng et al. 2011; Sio et al. 2013; Oeser et  
295 al. 2015). A few Mg isotope data points clearly deviate from the overall negative correlations  
296 between  $\delta^{26}\text{Mg}$  and  $\delta^{56}\text{Fe}$  values (Fig. 7). The origin of these "outliers" is unknown, but may be  
297 related to the fact that Mg and Fe isotope profiles for a certain olivine cannot be measured at the  
298 exact same positions in the crystal (see Fig. 2). Nevertheless, in these relatively Mg-rich olivines  
299 (Fo contents vary between 78 and 90 mol%), Fe isotope variations are much more pronounced than  
300 Mg isotope variations (Fig. 7), for mass balance reasons, and thus can put tighter constraints on the  
301 history of these crystals (see section "Discussion"). As a consequence, we will focus on Fe isotopic  
302 zoning, in combination with Fe-Mg chemical zoning, in the following sections. However, in the  
303 supplementary material we also show modeling results for Mg isotopic zoning in some olivine  
304 crystals (Fig. S4).

305 The olivines investigated here for Fe-Mg isotopes have a distinct trace element characteristic. In  
306 particular, the olivines with intermediate Fo, high Ni cores overgrown by high Fo, high Ni rims  
307 have unique Cr concentrations among all 1963-65 Irazú olivines. The cores have low Cr  
308 concentrations in a few crystals less than 10 ppm (Fig. 3), which is uncommon for primitive  
309 magmatic olivine phenocrysts, and stands in contrast to the high Ni content for the given Fo  
310 content. On the other hand, the high Fo rims have Cr contents that are more common in primitive  
311 arc olivines >100 ppm. This extreme difference between those cores and the rims and any other  
312 primitive Irazú olivine is best expressed in the Ni/Cr ratio, which for normal olivines ranges  
313 between 10 and 15, but reaches 370 in some cores of the intermediate Fo, high Ni olivines (Fig. 4).  
314 In addition to the unique Ni/Cr ratios, the zoning of trace elements generally correlates with major  
315 element zoning and therefore can potentially be linked to the timescales obtained here via the Fe-  
316 Mg isotopes and Fe-Mg elemental zoning. We note that many trace elements have zoning that  
317 correlates with major element zoning, however, we focus here solely on Cr, V, and Al zoning,  
318 because these have received particular attention given the recent controversy about their  
319 diffusivities in olivine (Ito and Ganguly 2006; Spandler et al. 2007; Milman-Barris et al. 2008;  
320 Spandler and O'Neill 2010). For all crystals that have zoning of Cr, V, and Al, the lengthscales of  
321 the zoning are shorter than the Fe-Mg zoning, however of a similar order of magnitude.

322

323

324

### **Diffusion modeling**

325 The simulation of observed Fe-Mg chemical and isotopic zoning in Irazú olivines by diffusion  
326 modeling generally follows the approaches presented in Oeser et al. (2015). In particular, in order  
327 to reproduce the reverse zoning of the complexly zoned olivines (hereafter referred to as diffusion  
328 episode 1) two models were applied: the first one comprises the internal homogenization of an

329 olivine by Fe-Mg inter-diffusion after two growth episodes which generate a step function in the  
330 Fo content profile (magnesian rim, ferrous core) but homogeneous Fe and Mg isotopic  
331 compositions (scenario 3 in Oeser et al. 2015). For this case, we assumed a plane sheet geometry  
332 and solved the following one-dimensional expression of the diffusion equation numerically by the  
333 method of finite differences (e.g. Crank 1975; Costa et al. 2008):

$$334 \quad \frac{\partial C_i(x,t)}{\partial t} = \frac{\partial}{\partial x} \left[ D_i \frac{\partial C_i(x,t)}{\partial x} \right] \quad (2)$$

335 where  $C_i$  is the concentration of element  $i$ ,  $t$  is time,  $D_i$  is the diffusion coefficient of element  $i$ , and  
336  $x$  is distance. The second possible model encompasses the growth of a Mg-rich rim around a  
337 homogeneous ferrous olivine crystal, but now associated with simultaneous Fe-Mg inter-diffusion  
338 (scenario 4 in Oeser et al. 2015). Here, the olivine is assumed to be a semi-infinite medium, and we  
339 use the same analytical solution to this moving boundary problem as given in Oeser et al. (2015):

$$340 \quad C(x,t) = C_0 + (C_{\text{rim}} - C_0) \times \exp\left[\frac{R(x-\frac{Rt}{2})}{2D}\right] \frac{1}{2} \exp(\lambda t) \left\{ \exp\left(-x\sqrt{\frac{\lambda}{D}}\right) \operatorname{erfc}\left[\frac{x}{2\sqrt{Dt}} - \sqrt{\lambda t}\right] + \right. \\ 341 \quad \left. \exp\left(x\sqrt{\frac{\lambda}{D}}\right) \operatorname{erfc}\left[\frac{x}{2\sqrt{Dt}} + \sqrt{\lambda t}\right] \right\} \quad (3)$$

342 where  $C_0$  is the initial concentration of the element of interest (as measured in the core of the  
343 olivine crystal),  $C_{\text{rim}}$  is the concentration of this element in the overgrowing rim,  $t$  is time,  $D$  is the  
344 diffusion coefficient,  $R$  is the growth rate,  $\lambda = R^2/(4D)$ , and  $x$  is distance.

345 Simple Fe-Mg exchange between olivine and melt (scenario 1 in Oeser et al. 2015) was modeled  
346 by applying a fixed rim composition and Fe-Mg inter-diffusion in olivine with a plane sheet  
347 geometry. In this case, the diffusion equation (eqn. 2) was solved numerically by the method of  
348 finite differences. This model was used to reproduce the normal zoning at the rims of the  
349 investigated olivines from Irazú volcano (hereafter referred to as diffusion episode 2). Further

350 initial and boundary conditions are given in the following paragraph and in section "Diffusion  
351 episode 1".

352 According to the parameterization given by Dohmen and Chakraborty (2007), which we applied in  
353 this study, the diffusion coefficient for Fe-Mg diffusion in olivine depends on oxygen fugacity  
354 ( $fO_2$ ), the mole fraction of the fayalite component ( $X_{Fe}$ ), pressure ( $P$ ), temperature ( $T$ ), and the  
355 crystallographic orientation, i.e. the orientation of each measured profile relative to the  
356 crystallographic axes determined by electron backscatter diffraction. The latter was used to  
357 calculate the diffusion coefficient for individual profiles, following the approach given in, e.g.,  
358 Costa and Chakraborty (2004). The presence of water in an olivine crystal can also significantly  
359 affect  $D_{Fe-Mg}$  in olivine (Wang et al. 2004; Hier-Majumder et al. 2005). We have not investigated  
360 whether the Irazú olivines contain significant amounts of water (i.e. >11 ppm, as reviewed by  
361 Chakraborty 2010), and thus we do not account for effects from hydrogen. However, Irazú magmas  
362 commonly contain ~ 3 wt% H<sub>2</sub>O (Benjamin et al. 2007) and coexisting olivines may contain  
363 therefore significant amounts of water. Thus, our modeled timescales represent maximum  
364 estimates, as the presence of H<sup>+</sup> in olivine increases  $D_{Fe-Mg}$  by up to one order of magnitude relative  
365 to anhydrous conditions (e.g. Jollands et al. 2016b). The durations of the diffusive processes were  
366 estimated at isothermal conditions, and temperature estimations in this study rely on thermometry  
367 data for Irazú magmas presented in Alvarado et al. (2006), Benjamin et al. (2007) and Ruprecht  
368 and Plank (2013). In particular, for diffusion episode 1 we assumed a temperature of 1150°C,  
369 consistent with the temperature range (1100-1200°C) used for simulating Ni diffusion profiles in  
370 the magnesian olivines from Irazú (Ruprecht and Plank 2013). For the second diffusion episode, a  
371 temperature of 1080°C was used for diffusion modeling, consistent with temperature estimations  
372 based on whole rocks and melt inclusions of more evolved compositions (Benjamin et al. 2007), as



373 this diffusion event likely occurred in the plumbing system in the shallow crust (see below). The  
374 oxygen fugacity was assumed to be  $\Delta \log fO_2$  (NNO) = +1 (Benjamin et al. 2007), and its absolute  
375 value at the given temperature was calculated following the parameterization of Schwab and  
376 Küstner (1981). As the compositional variability concerning  $X_{Fe}$  is rather limited, i.e. fayalite  
377 contents range from 0.22 to 0.12 for the investigated olivines,  $X_{Fe}$  was presumed to be constant at  
378 an average value of 0.15 in all simulations. The uncertainty introduced by this simplification on  
379  $D_{Fe-Mg}$  (and, hence, on the calculated timescales) is smaller than a factor of 1.6. Pressure variations  
380 have the least effect on  $D_{Fe-Mg}$ , compared to variations in  $T$ ,  $fO_2$ ,  $X_{Fe}$ , or orientation, and given the  
381 pressure range experienced by the investigated olivine crystals (~0.1-500 MPa). Following the  
382 pressure estimation made for the magnesian Irazú olivines (Ruprecht and Plank 2013) we assumed  
383  $P$  to be 500 MPa and 100 MPa during diffusion episode 1 and 2, respectively.  
384 The (kinetic) isotope fractionation associated with the Fe-Mg inter-diffusion was modeled by  
385 applying the following empirical formula (e.g. Richter et al. 1999):

386 
$$\frac{D_q}{D_r} = \left( \frac{M_r}{M_q} \right)^\beta \quad (4)$$

387 where  $D$  represents the diffusion coefficient,  $q$  and  $r$  are isotopes of a certain element (e.g.,  $^{54}Fe$   
388 and  $^{56}Fe$ ), and  $M$  is the atomic mass in a.m.u. The exponent  $\beta$  is an empirical constant and equal to  
389 0.5 in ideal monoatomic gases (Van Orman and Krawczynski 2015). For olivine,  $\beta$ -values appear  
390 to depend on crystal composition and diffusion direction relative to the orientation of the  
391 crystallographic axes (Van Orman and Krawczynski 2015) and, as the systematics of these  
392 dependencies is not clear yet, they are thus difficult to predict; however, they can be estimated by  
393 fitting observed isotope diffusion profiles across (natural) olivine crystals. Studies by Sio et al.  
394 (2013), Oeser et al. (2015) and Richter et al. (2016) indicate that  $\beta$ -values for olivine fall in a range

395 of 0.055-0.16 for Mg and of 0.075-0.30 for Fe. In this study, Fe and Mg isotopic profiles were  
396 fitted by adjusting the respective  $\beta$ -values, i.e. modifying the diffusivity ratio of two isotopes of the  
397 same element (e.g.  $^{54}\text{Fe}$  and  $^{56}\text{Fe}$ ) until the amplitude of the modeled isotope fractionation matched  
398 that of the observed one. As a consequence, the used  $\beta$ -values are purely empirical and  
399 characteristic for the given simulated case. In this study,  $\beta_{\text{Fe}}$ -values used range from 0.12 to 0.25  
400 (Table 1), and therefore fully agree with previous estimates of  $\beta_{\text{Fe}}$  in olivine (Sio et al. 2013; Oeser  
401 et al. 2015; Collinet et al. 2017).

402

403

404

## Discussion

### 405 **Sequence of events recorded by complexly and normally zoned Irazú olivines**

406 The zoning characteristics of type B olivines provide unique constraints on the spatial origin of  
407 these crystals and the temporal incorporation of different end-members into the eruption. As type A  
408 and type B crystals have a shared history within the outer zones of the crystals and type B olivines  
409 merge onto the general geochemical evolution recorded in type A olivines we suggest that the  
410 xenocrystic type B cores were remobilized in the crust by ascending mantle-derived melts carrying  
411 the magnesian type A olivines. This is evidence that the mixing event recorded by Ni zonation in  
412 magnesian type A olivines occurred at deeper levels than the storage depth of the xenocrystic type  
413 B olivines (which is likely to be >15 km beneath the volcano's summit based on the trace element  
414 characteristics of type B olivines; see below), further corroborating the assumption of near-Moho  
415 origin of the primitive type A olivines (Ruprecht and Plank, 2013). In addition to the compositional  
416 differences, type B complex olivines are also not uniformly distributed within the 1963-65  
417 eruption. Of the three samples studied in the past, IZ-10-11 (the earliest phase of the eruption

418 studied by Ruprecht and Plank, 2013) lacks significant amounts of type B olivines, while type A  
419 and C are both common. Type B olivines are most common in IZ-10-12, but also present in IZ-10-  
420 13, suggesting that the assembly of the 1963-65 eruption is complex where some components wane  
421 and wax.

422 The complex zoning of the Irazú type B olivines investigated in this study points to at least two  
423 episodes of crystal growth under diverse crystal-melt equilibrium conditions. The intermediate Fo  
424 contents ( $Fo_{80-87}$ ), high Ni concentrations (2600-3200 ppm Ni), and anomalously high Ni-Cr ratios  
425 (Fig. 4) in the cores of these olivines indicate that these crystals were formed and equilibrated in  
426 the middle or lower crust beneath Irazú volcano, probably in a crystal mush or mafic cumulate (see  
427 section "Assembling the Irazú magma"). The Mg-rich zone near the rims of the complexly zoned  
428 olivines may then have formed by crystal growth when these crystals were entrained into ascending  
429 mantle melts and brought into the plumbing system in the shallow crust beneath Irazú volcano. The  
430 sharp intra-mineral chemical gradient produced by this second growth event was eventually  
431 smoothed out by Fe-Mg inter-diffusion as implied by the strong coupling of Fe-Mg chemical and  
432 isotopic zoning. Modeling this diffusive process (diffusion episode 1) therefore yields information  
433 about the time span from the entrainment of these olivines in the crust to the eruption. Previous  
434 studies have used similar Fe-Mg zoning patterns (initial step function, subsequent diffusive  
435 relaxation) in olivine to estimate mixing-to-eruption timescales by diffusion modeling (e.g. Kahl et  
436 al. 2011, 2013; Longpré et al. 2014; Hartley et al. 2016). However, all of these studies assumed  
437 that the observed zoning was the result of instantaneous growth of a crystal layer with a different  
438 composition than the core and subsequent Fe-Mg inter-diffusion (cf. scenario 3 in Oeser et al.  
439 2015), which provides maximum timescales. In this study, we additionally apply a model where  
440 crystal growth and Fe-Mg inter-diffusion occur simultaneously (scenario 4 in Oeser et al. 2015). As  
441 shown in Oeser et al. (2015), Fe-Mg isotopic profiles as obtained in this study can provide a further

442 constraint which one of the two scenarios is more appropriate to describe the history recorded by  
443 intra-mineral zoning of olivine, leading to more reliable timescales obtained from diffusion  
444 modeling.

445 The similar zoning patterns at the rims of the complexly zoned olivines and of the primitive  
446 olivines (Fo<sub>-88</sub> to Fo<sub>-80</sub>) from Irazú indicate that both crystal populations experienced a comparable  
447 late-stage history prior to the eruption. This likely included mixing with resident evolved magmas  
448 in the shallow crust as well as fractional crystallization from and, thus, further differentiation of the  
449 hybrid melt. The relative enrichment of light Fe and heavy Mg isotopes in the normally zoned part  
450 near the olivine rim implies that diffusive Fe-Mg exchange between crystal and melt represents the  
451 main process for the re-equilibration of olivine during that stage. It has to be noted that in some  
452 olivines the width of the normal zoning is so narrow that inversely correlated Fe-Mg isotopic  
453 profile cannot unambiguously be resolved by the LA technique used in this study, given the spatial  
454 resolution of 30-40  $\mu\text{m}$ . However, olivines with wide normal zoning (e.g. IZ-10-13 ol22 and ol31  
455 in Fig. 6 and Fig. S3, respectively) clearly show anti-correlated Fe-Mg isotopic profiles, which  
456 strongly argues for a diffusion origin of the normal zoning. As a consequence, we interpret this  
457 normal zoning to mainly be the result of a second diffusion episode (whose duration can be  
458 estimated by diffusion modeling) occurring in the shallow crust (shortly) prior to eruption.

459

#### 460 **Timescales of diffusion episodes**

461 **Diffusion episode 1.** As described above, we applied two models to simulate the reverse zoning of  
462 the complexly zoned olivines (diffusion episode 1). The first one (instantaneous growth and  
463 subsequent Fe-Mg inter-diffusion) provides maximum timescales for the diffusive process, while in  
464 the second model (crystal growth and simultaneous Fe-Mg inter-diffusion) the shape of the  
465 simulated profiles depends on the timescale, as well as on the ratio of crystal growth rate ( $R$  in m/s)

466 and diffusion coefficient of Fe-Mg diffusion in olivine ( $D$  in  $\text{m}^2/\text{s}$ ; see Fig. S5 for further details).  
467 Since  $D$  is determined by the chosen model parameters ( $T$ ,  $P$ ,  $f\text{O}_2$ , etc.) and assumed to be identical  
468 in both models, the second model yields a timescale for the diffusive re-equilibration of the crystal  
469 and additionally the growth rate of the Mg-rich layer. Following the stepwise modeling approach  
470 presented in Kahl et al. (2011), the diffusion profiles generated during the first diffusion episode  
471 are used as the initial concentration (and isotopic) gradients for diffusion episode 2, which  
472 produced the normal zoning at the rims of the complexly zoned and primitive olivines investigated  
473 herein. Hence, the reverse zoning of diffusion episode 1 is not simulated to full completion, in  
474 order to allow for further relaxation during the second diffusion episode. The results of this  
475 diffusion modeling approach are shown in Tables 1 and 2, and representative simulated diffusion  
476 profiles for four complexly zoned olivines are illustrated in Figures 8 to 11. Our results indicate  
477 that the duration of diffusion episode 1 was between  $\sim 90$  days and  $\sim 530$  days (Table 1). These  
478 timescales imply that the transfer of material from the lower crust to the shallow plumbing system  
479 beneath an active volcano can occur in less than two years or even within a few months. This is  
480 consistent with observations made by Ruprecht and Plank (2013) for primitive magnesian olivines  
481 with reverse Ni zoning, which have been transported with mantle-derived magmas through the  
482 crust at an average rate of 50-80 meters per day. The fact that the complexly zoned olivines still  
483 record quite a range of timescales of diffusive re-equilibration (90-530 days) may indicate (i) that  
484 not all of them were transported with the same mantle-derived magma batch, and/or (ii) that they  
485 were picked up from various parts of the crust beneath Irazú volcano at different points in time.  
486 The latter possibility is supported by the fact that the cores of these olivines show a range of Fo  
487 compositions from Fo<sub>80</sub> to Fo<sub>87</sub>, along with some variability in Fe- and Mg isotopic compositions,  
488 suggesting slightly different equilibration environments. Nevertheless, six out of the seven  
489 complexly zoned olivines investigated herein record diffusion timescales of less than 300 days for

490 the first diffusion episode, and thus a rapid transport into the shallow crust after being entrained  
491 into ascending mantle melts. These timescales are even shorter by up to ~25% if we assume that  
492 the growth of the Mg-rich layer and the diffusive equilibration of this intra-mineral chemical  
493 gradient occurred simultaneously during diffusion episode 1 (Table 2). Coeval growth and  
494 diffusion could also explain slight offsets (within ~ 30  $\mu\text{m}$ ) of the profile step locations for  
495 different elements required to fit Fe-Mg, Cr, V, and Al zoning profiles (Fig. 14). Unfortunately,  
496 neither the elemental zoning nor the Fe(-Mg) isotopic profiles can unequivocally unravel which of  
497 the two scenarios is the more likely one for the investigated olivines, i.e. both models  
498 (instantaneous growth + subsequent diffusion vs. growth + simultaneous diffusion) yield equally  
499 good fits for the Fe isotopic profiles in most olivines (Fig. 10). Only for one crystal (IZ-10-12 ol29)  
500 the second model reproduces the Fe isotopic profile much better (Fig. 11). The fairly good fits, that  
501 are obtained if the first model (instantaneous growth + subsequent diffusion) is applied, indicate  
502 that assumptions made in previous studies for similar Fe-Mg chemical profiles in olivine (e.g. Kahl  
503 et al. 2011; Longpré et al. 2014; Hartley et al. 2016) are reasonable (see above). In other words,  
504 based on the observations of this study, potential simultaneous crystal growth and diffusion appears  
505 to play a minor role in modifying Fe-Mg chemical and isotopic gradients in olivine. Nevertheless,  
506 only the combination of in-situ Fe-Mg concentration and isotopic analyses in olivine is capable of  
507 unraveling such complex growth+diffusion histories.

508 As described above, our second model (growth + simultaneous diffusion) does not only provide a  
509 timescale for the diffusion process, but also a growth rate for the Mg-rich layer. Based on the  
510 diffusion profiles observed in this study, these growth rates range from  $3 \times 10^{-12}$  m/s to  $2 \times 10^{-11}$  m/s  
511 (Table 2). Compared to experimentally determined growth rates of olivines in basaltic melts ( $10^{-10}$ -  
512  $10^{-7}$  m/s, Donaldson 1975; Jambon et al. 1992), such growth rates seem to be exceptionally low.  
513 However, in natural systems olivine growth rates of  $10^{-11}$ - $10^{-9}$  m/s appear to be more realistic

514 (Maaløe 2011; Watson et al. 2015), probably due to a smaller degree of supercooling prevailing  
515 under natural conditions (cf. Jambon et al. 1992), and/or due to a decrease in the degree of  
516 supersaturation with time in a natural magma reservoir (cf. Maaløe 2011), which is less  
517 pronounced in experimental studies because of the limited runtime of the experiments.

518

519 **Diffusion episode 2.** The second diffusion episode, which generated the normally zoned rims of  
520 the complexly zoned and the primitive olivines investigated herein, lasted between a few days and  
521 ~450 days (Tables 1 and 2). One olivine records an exceptionally long timescale of diffusive re-  
522 equilibration in the second diffusion stage (~1450 days, IZ-10-13 ol21). It has to be noted again  
523 that these timescales – as well as the timescales for diffusion episode 1 – likely represent  
524 maximum estimates since we assume that the observed normal zoning is purely diffusion-generated  
525 (as implied by the inversely correlated Fe-Mg isotopic profiles, Fig. 6) and that the investigated  
526 olivines do not contain significant amounts of  $H^+$  (which would strongly enhance  $D_{Fe-Mg}$ ; e.g.  
527 (Hier-Majumder et al. 2005). However, we cannot exclude that crystal growth during fractional  
528 crystallization and further differentiation of the melt also had some influence on the development  
529 of that normal zoning, especially near the surface where cooling and partial degassing drive  
530 crystallization more effectively. Unfortunately, the spatial resolution of the LA technique used in  
531 this study is not sufficient to resolve the extent of this influence. As a consequence, we prefer to  
532 provide conservative, i.e. maximum, estimates for the timescales of processes beneath Irazú  
533 volcano. The variability of observed diffusion times for diffusion episode 2 may be an indication of  
534 a complex plumbing system beneath Irazú volcano where various crystal populations are stored in  
535 diverse magma batches for a certain amount of time before eruption. Still, short diffusion  
536 timescales, i.e. <200 days, appear to prevail, implying that magma mixing events in the shallow  
537 crust and following eruptions take place within a matter of months, if not days.

538

539 **Mixing-to-eruption times and magma ascent rates.** The stepwise modeling approach applied in  
540 this study (see above) also allows us to estimate the time spans between the entrainment of the  
541 complexly zoned olivines into ascending mantle melts and the eruption ( $t_{\text{total}}$  in Tables 1 and 2).  
542 These timescales range from ~100 days to ~1750 days, with short timescales of <600 days ( $n = 6$ )  
543 being more common than longer ones ( $n = 1$ ). Assuming that these crystals were picked up  
544 somewhere in the middle crust, i.e. ~15 km beneath the volcano's summit, given a crustal thickness  
545 of ~35 km (Dzierma et al. 2010), the average ascent rate of the magmas carrying these olivines  
546 through the crust would be 25 meters per day. Again, this represents a conservative estimate as (i)  
547 several olivines record mixing-to-eruption timescales much shorter than 600 days, and (ii) the  
548 complexly zoned olivines may have been entrained into ascending mantle melts at greater depth,  
549 i.e. >15 km beneath the volcano's summit (see above). Also, such rates are averages, while  
550 transient transport rates are likely much faster (e.g. Rubin 1995). Nevertheless, an average ascent  
551 rate of 25 m/day is largely consistent with magma ascent rates of 50-80 m/day determined by  
552 Ruprecht and Plank (2013) based on diffusion modeling of reverse Ni zoning in magnesian olivines  
553 from the same sample suite and using a temperature range of 1100 to 1200°C.

554

555 **Uncertainties of the modeled timescales.** As illustrated in, e.g., Oeser et al. (2015) and Kahl et al.  
556 (2015), uncertainties of timescales calculated by Fe-Mg diffusion modeling mainly result from the  
557 uncertainty of temperature estimations. In our case, we chose a temperature of 1150°C for diffusion  
558 episode 1 which is well within the range of reported pre-eruptive temperatures of basalts from  
559 Irazú volcano (Alvarado et al. 2006). If this assumption was incorrect, and the temperature was  
560 50°C lower (or higher), the timescales obtained from diffusion modeling for diffusion episode 1  
561 would be longer (or shorter) by a factor of ~2.4. For diffusion episode 2, we assumed a temperature



562 of 1080°C, based on olivine-melt thermometry for basaltic andesites erupted from Irazú in 1723  
563 (Benjamin et al. 2007). If this temperature was actually 50°C lower for the samples investigated in  
564 our study (basaltic andesites erupted in 1963-65), the timescales obtained from diffusion modeling  
565 for diffusion episode 2 would be longer by a factor of ~2.6. As summarized in Costa et al. (2008),  
566 at any given temperature the uncertainty of  $D_{\text{Fe-Mg}}$  resulting solely from the experimental  
567 calibration is a factor of ~2. Incorrect assumptions for the oxygen fugacity prevailing during the  
568 diffusion stages may also lead to over- or underestimations of diffusion times (Oeser et al. 2015);  
569 however, these are minor compared to those arising from errors of temperatures estimates. For  
570 example, if the oxygen fugacity was at  $\Delta \log f\text{O}_2$  (NNO) = 0 (instead of +1 as presumed in our  
571 model) during diffusion episode 1, modeled timescales would be longer by a factor of ~1.5. Given  
572 these uncertainties, we are confident that the overall uncertainty of the timescales determined by  
573 diffusion modeling in this study is not larger than a factor of 4 which is typical for the calculation  
574 of timescales from diffusion modeling in magmatic systems (Costa et al. 2008).

575

576

### 577 **Assembling the Irazú magma prior to the 1963-65 eruption**

578 The new results from Fe-Mg isotopes generally corroborate the timescales of magma transport  
579 through the crustal magma system (Ruprecht and Plank 2013). By expanding our analysis to  
580 crystals with intermediate Fo, high Ni cores we can provide a more complete picture of the  
581 architecture of the magma system beneath Irazú and the assembly of the magma prior to the 1963-  
582 65 eruption (Fig. 12). Comparison with other crustal cumulate olivine data from crustal sections  
583 (e.g. Otamendi et al. 2016) suggests that the intermediate Fo, high Ni cores are not simple olivine  
584 phenocrysts that represent a different magma batch in the crust. Instead, they are best explained as  
585 xenocrystic olivine that was picked up by the primitive magma as it intersected an olivine

586 cumulate. The high Ni/Cr ratio in these cores are consistent with this interpretation and suggest that  
587 during olivine cumulate formation pyroxenes co-crystallized and through time both mineral phases  
588 equilibrated so that Ni was incorporated into the olivine at the expense of pyroxene and Cr behaves  
589 in reverse being concentrated in the pyroxenes. Typical distribution coefficients for Cr and Ni  
590 between olivine and clinopyroxene are  $<0.01$  and  $\sim 4$ , respectively (Otamendi et al. 2016). The  
591 observed concentrations are consistent with this model as Cr content in the xenocrystic cores is one  
592 to two orders of magnitude lower than in common olivines and Ni is about a factor of 2 higher.  
593 Our model suggests that the primitive magma intersected a mafic (maybe ultramafic) crustal  
594 cumulate on its way to the surface. The fact that olivine cores are overgrown by primitive Ni-rich,  
595 high Fo mantles further indicates that this contamination of the erupted magma occurred at a depth  
596 well below the shallow Irazú magma plumbing system. We have not attempted to estimate the  
597 depth for this cumulate, but comparing our results to a crustal section study that exposes olivine  
598 cumulates of similar geochemical characteristics within the section suggests at least mid-crustal  
599 levels ( $>15$  km depth; Otamendi et al. 2016) and potentially lower crustal conditions. Those  
600 depths/pressures are constrained through phase assemblages and compositions studied in this  
601 crustal section. Irrespective of the exact location of olivine cumulate contamination, crystal zoning  
602 correlations between type A and B olivines provides evidence that the primitive high Fo olivines  
603 must originate from greater depth and likely crystallized near the Moho (as it was previously  
604 postulated by Ruprecht and Plank, 2013), precluding aphyric magma ascent of primitive melts to  
605 shallow levels with ensuing crystallization near the shallow crustal magma reservoirs (Fig. 12).

606

### 607 **Diffusion of chromium in olivine**

608 The coupling of Fe-Mg isotope and elemental zoning results provide valuable constraints for the  
609 growth and diffusion history of the Irazú olivines. As indicated by the Fe-Mg isotopic zoning,

610 chemical zoning was essentially driven by chemical diffusion following rapid (near instantaneous)  
611 growth. This finding allows to utilize Irazú olivines to calculate relative diffusion coefficients for  
612 Cr, Al, and V. We focus on Cr, Al, and V diffusivities as those are much less understood than  
613 divalent minor and major elements in olivine (Fe-Mg, Mn, Ca, Ni). In particular, Cr and Al  
614 diffusivities have been argued to vary over orders of magnitude (Cr: Ito and Ganguly 2006;  
615 Spandler and O'Neill 2010; Jollands et al. 2018; Al: Milman-Barris et al. 2008; Spandler and  
616 O'Neill 2010; Zhukova et al. 2017). The variations in Cr diffusion have been reconciled recently by  
617 Jollands et al. (2018) recognizing that low  $a_{\text{SiO}_2}$  systems (Ito and Ganguly 2006) may show slow Cr  
618 diffusion, while high  $a_{\text{SiO}_2}$  systems near the olivine-enstatite buffer (Spandler and O'Neill 2010)  
619 lead to significantly faster Cr diffusion. Moreover it was documented that Cr diffusion is strongly  
620 concentration dependent (Jollands et al. 2018), an effect that cannot be explored in this study  
621 meaningfully.

622 We do not attempt to estimate specific diffusivities for a specific temperature as we think that there  
623 are still significant uncertainties with the analysis of these natural olivines. Neither temperature nor  
624 oxygen fugacity in the Irazú magmatic system (thought to be near NNO+1, Benjamin et al. 2007,  
625 setting valence states of predominantly  $\text{Cr}^{3+}$  and  $\text{V}^{4+}$ , Papike et al. 2005; Mallmann and O'Neill  
626 2009) is tightly constrained when compared to experimental studies. However, as chemical zoning  
627 is primarily a result of chemical diffusion and not growth for the investigated olivine crystals, we  
628 can determine Cr and, with a higher uncertainty, V and Al diffusion relative to each other and to  
629 that of experimentally well-determined Fe-Mg inter-diffusion (Fig. 13).

630 For this approach to produce meaningful diffusivity estimates, we assume that the zoning profile in  
631 the olivines is generated by diffusion following a rapid growth period that generated a step function  
632 as starting conditions for diffusion and therefore represents a maximum diffusivity estimate. The  
633 previously discussed Fe-Mg isotope zoning supports in general this diffusion-dominated regime for

634 the studied olivines. Secondly, to extract diffusivity information on these elements one has to  
635 ensure that the finite (measured) lengthscale of the zoning profiles is primarily a true (diffusion-  
636 generated) zoning profile in the crystals and not an artifact of the acquisition method using a  
637 scanning laser with a continuous time series. The laser spot size of 25  $\mu\text{m}$  and the continuous  
638 sample movement underneath the laser intrinsically creates zonation patterns with a measurable  
639 lengthscale, where even true step functions would be smeared to resemble an evolving diffusion  
640 profile (Bradshaw and Kent 2017; Fig. S6 in the supplementary material). With increasing  
641 diffusion length scale (e.g., due to increasing diffusivity, Fig. S6) zoning profiles acquired during  
642 laser ablation and the true zoning profiles converge, and the sampling bias by a continuously  
643 moving laser track becomes negligible. To quantitatively compare true zoning and laser-generated  
644 profiles we extrapolate the maximum slope at the center where the curvature of the profile changes  
645 signs to the concentration plateaus of the step functions and obtain a conservative estimate  $\Delta x$  for  
646 the diffusion profile length for all elements of interest (Fig. S7, Tab. S4). Comparing the apparent  
647 diffusivity as a function of the lengthscale  $\Delta x$  one can see that as  $\Delta x$  approaches the spot size (25  
648  $\mu\text{m}$ ) the profile cannot be resolved unambiguously and any calculated diffusivity becomes  
649 increasingly overestimated. We can correct for this effect by calculating a smearing factor that  
650 accounts for the laser-generated smearing effects. All reported lengthscales  $\Delta x$  for all elements  
651 (Tab. S4) are significantly larger than the minimum spatial resolution, thus supporting the notion  
652 that these profiles are not simple step functions. As a result we can estimate diffusivities and apply  
653 a correction factor for the smearing that varies between 1.399 ( $\Delta x = 32 \mu\text{m}$ ) and 1.071 ( $\Delta x =$   
654  $71 \mu\text{m}$ ). No correction was applied for zoning lengthscales  $>100 \mu\text{m}$  as our analysis was limited to  
655  $\Delta x < 100 \mu\text{m}$  and the polynomial approximation is overestimating the smearing effect for large  $\Delta x$ .  
656 Lastly, we neglect potential effects of off-perpendicular (relative to the crystal-melt interface)  
657 sectioning of the crystals. Imaging of the crystals prior to and after LA-ICP-MS analysis, after

658 polishing (typically removal of ~10-20  $\mu\text{m}$ ) prior to MC-ICP-MS analysis, and again after  
659 polishing post MC-ICP-MS analysis shows that zoning evolution into the Z-direction of the crystal  
660 is persistent and crystal shapes in general change only to a small degree. This suggests that crystal  
661 sectioning was close to perpendicular relative to the crystal-melt interface for most crystals. In  
662 summary, the calculated zoning lengthscales and crystal imaging suggest that the profiles represent  
663 gradational zoning in Cr content from core to rim and, together with the Fe-Mg isotope zoning  
664 record, we can estimate elemental diffusivities for selected elements.

665 After the smearing correction is applied we find that  $D_{\text{Fe-Mg}} > D_{\text{V}} > D_{\text{Cr}} > D_{\text{Al}}$ . Our most robust  
666 results – given that we find similar diffusivities for 7 crystals and 9 different traverses (figures 13,  
667 14 and S8) – are for  $D_{\text{Cr}}$ , which is smaller than  $D_{\text{Fe-Mg}}$  by a factor of  $4.9 \pm 2.9$ , with the uncertainty  
668 representing 1SD (standard deviation) of the calculated population of diffusion coefficients.

669 Therefore, the results from these natural olivines are consistent with fast Cr diffusion under high  
670  $a_{\text{SiO}_2}$  conditions (Spandler and O'Neill 2010; Jollands et al. 2018). Al and V are less well  
671 constrained, because zoning profiles are not consistently correlated with Fo content and, in the case  
672 of Al, our analytical uncertainties were high relative to Cr concentrations in all of our IZ-10-13  
673 olivines. Nonetheless, the two olivine crystals for which Al zoning was distinct suggest that Al  
674 diffusivity may still be faster than inferred from other studies (Milman-Barris et al. 2008; Spandler  
675 and O'Neill 2010) consistent with results for Al diffusion in olivine under high  $a_{\text{SiO}_2}$  (Zhukova et  
676 al. 2017). A more extensive analysis will be required in the future, especially given that the  
677 comparison (Fig. 13) with data from Spandler and O'Neill (2010) is done at different temperatures.

678 While our diffusivity ratios are calculated for a temperature of about ~1150°C, Spandler and  
679 O'Neill (2010) ran their experiments at 1300°C. Thus, potential differences in the activation energy  
680 (~ 200 kJ/(mol K) for Fe-Mg inter-diffusion in olivine, Dohmen et al. 2007; >300 kJ/(mol K) for  
681 Cr in olivine, Ito and Ganguly 2006, Jollands et al. 2018) are not accounted for in this comparison.

682 The lower activation energy for Fe-Mg inter-diffusion suggests that  $D_{\text{Fe-Mg}}/D_{\text{Cr}}$  should increase  
683 with lower temperatures as Cr diffusion is slowed more effectively at low T. Lastly, V diffusivity,  
684 as determined by the zoning profiles of only four olivine crystals (Fig. 13 and 14), is smaller than  
685  $D_{\text{Fe-Mg}}$  by a factor of  $2.5 \pm 1.4$ . However, the activation energy for V diffusion is not known and an  
686 analysis for different temperatures cannot be performed. Given that V will not only be present as  
687 tetravalent species in terrestrial magmas, but also to some degree trivalent (Papike et al. 2005;  
688 Mallmann and O'Neill 2009) it will require more experimental work to fully characterize V  
689 diffusion in olivine under common oxygen fugacities.

690 These results for diffusivities in olivine of selected tri- and tetravalent cations are, with the  
691 exception of Al, in good agreement with those of the previous study of Spandler and O'Neill  
692 (2010) who also did not provide diffusivities for a range of temperatures, but rather limited their  
693 experimental investigation to 1300 °C using San Carlos olivine (Fig. 13). In the latter study the  
694 authors had excellent crystallographic control and provided diffusivity estimates for all three  
695 crystallographic orientations and all their diffusivities are of similar order of magnitude to  
696 diffusivities from this study. The olivines in this study experienced lower temperatures (<1200 °C,  
697 see Ruprecht and Plank 2013, for additional discussion on this) compared to that of the  
698 experimental study of Spandler and O'Neill (2010). Both studies utilized natural olivines, and the  
699 San Carlos olivines as well as the overgrowth mantles in the Irazú olivines have high Fo contents  
700 of ~ Fo90. However, given the limitations of our analysis, the determined diffusivities represent  
701 maximum diffusivity estimates and while they overlap with experimental results (Spandler and  
702 O'Neill 2010; Jollands et al. 2018) this agreement may break down if any made assumption is not  
703 maintained. The result would be smaller diffusivities. Given the good agreement however and the  
704 fact that  $a_{\text{SiO}_2}$  (e.g., Jollands et al. 2018) is a likely factor controlling the differences in diffusivities  
705 suggest that our analysis is robust. Nonetheless, additional similar studies on natural samples,

706 together with experimental studies, may enable to elucidate reasons for a range in diffusivity for  
707 specific elements.

708

709

710

### **Implications**

711 The findings of this study demonstrate that a combination of Fe-Mg chemical and isotopic profiles  
712 in complexly zoned olivines provides reliable information on the origin of zoning and thus, enables  
713 to unravel complex crystal growth and diffusion histories. This information can be used to optimize  
714 diffusion modeling in order to estimate the time scales of magmatic processes. In the case of  
715 olivines from Irazú volcano, such diffusion modeling yields the time span for the transfer of  
716 material from the middle or lower crust to the shallow plumbing system (<300 days for 6 out of 7  
717 olivine crystals), but also magma mixing-to-eruption timescales. These are in good agreement with  
718 the timescales of ascent of mantle-derived magmas, as determined by diffusion modeling of Ni  
719 zoning in primitive olivines (Ruprecht and Plank 2013). Mixing-to-eruption times for crustal  
720 olivines beneath the continental arc volcano Irazú appear to be short (<600 days in most cases) –  
721 despite a magma storage region in the shallow crust – and thus imply that material from the lower  
722 or middle crust can be transported to the shallow plumbing system and eventually to the vent of an  
723 arc volcano while the eruption is already underway.

724 Xenocrystic olivines may provide independent constraints on the diffusivity of commonly slow-  
725 diffusing elements such as Cr and Al for which the determination of diffusivities by experimental  
726 approaches is challenging. However, analysis of Irazú olivines with xenocrystic cores and Cr  
727 zoning suggests that under those natural conditions Cr diffusion in olivine may be faster than  
728 previously determined by experiments. Additional studies are required to fully understand the  
729 mechanisms of Cr diffusion and whether elevated trace impurities and defects or the chemical

730 activity of major elements are causing the enhanced diffusivity, as suggested for other elements  
731 and in olivine and other phases (e.g. Zhukova et al. 2014; Ferriss et al. 2016; Jollands et al. 2018).

732

733

734

#### Acknowledgements

735 This study was supported by the project funding program "Wege in die Forschung II" of the  
736 Leibniz Universität Hannover. P.R. acknowledges support from the U.S. National Science  
737 Foundation (EAR 1426820/1719687). We are grateful to M. Jollands and C. Sio for very  
738 constructive reviews and we thank R. Almeev for efficient editorial handling.

739

740

741

#### References

742

743 Alvarado, G.E., Carr, M.J., Turrin, B.D., Swisher, C.C., Schmincke, H.-U., and Hudnut, K.W.

744 (2006) Recent volcanic history of Irazú volcano, Costa Rica: Alternation and mixing of two  
745 magma batches, and pervasive mixing. Geological Society of America Special Papers , 412,  
746 259–276.

747 Benjamin, E.R., Plank, T., Wade, J.A., Kelley, K.A., Hauri, E.H., and Alvarado, G.E. (2007) High  
748 water contents in basaltic magmas from Irazú Volcano, Costa Rica. Journal of Volcanology  
749 and Geothermal Research, 168, 68–92.

750 Bradshaw, R.W., and Kent, A.J.R. (2017) The analytical limits of modeling short diffusion  
751 timescales. Chemical Geology, 466, 667–677.

752 Carr, M.J. (1984) Symmetrical and Segmented Variation of Physical and Geochemical  
753 Characteristics of the Central American Volcanic Front. Journal of Volcanology and



- 754 Geothermal Research, 20, 231–252.
- 755 Chakraborty, S. (1997) Rates and mechanisms of Fe-Mg interdiffusion in olivine at 980°–1300°C.  
756 Journal of Geophysical Research, 102, 12317–12331.
- 757 Chakraborty, S. (2010) Diffusion Coefficients in Olivine, Wadsleyite and Ringwoodite. Reviews in  
758 Mineralogy and Geochemistry, 72, 603–639.
- 759 Collinet, M., Charlier, B., Namur, O., Oeser, M., Médard, E., and Weyer, S. (2017) Crystallization  
760 history of enriched shergottites from Fe and Mg isotope fractionation in olivine megacrysts.  
761 Geochimica et Cosmochimica Acta, 207, 277–297.
- 762 Costa, F., and Chakraborty, S. (2004) Decadal time gaps between mafic intrusion and silicic  
763 eruption obtained from chemical zoning patterns in olivine. Earth and Planetary Science  
764 Letters, 227, 517–530.
- 765 Costa, F., and Dungan, M. (2005) Short time scales of magmatic assimilation from diffusion  
766 modeling of multiple elements in olivine. Geology, 33, 837–840.
- 767 Costa, F., Dohmen, R., and Chakraborty, S. (2008) Time Scales of Magmatic Processes from  
768 Modeling the Zoning Patterns of Crystals. Reviews in Mineralogy and Geochemistry, 69,  
769 545–594.
- 770 Crank, J. (1975) The Mathematics of Diffusion, second edi., 421 p. Clarendon Press, Oxford.
- 771 Dohmen, R., and Chakraborty, S. (2007) Fe-Mg diffusion in olivine II: Point defect chemistry,  
772 change of diffusion mechanisms and a model for calculation of coefficients in natural olivine.  
773 Physics and Chemistry of Minerals, 34, 409–430.
- 774 Donaldson, C.H. (1975) Calculated diffusion coefficients and the growth rate of olivine in a basalt  
775 magma. Lithos, 8, 163–174.
- 776 Dzierma, Y., Thorwart, M.M., Rabbel, W., Flueh, E.R., Alvarado, G.E., and Mora, M.M. (2010)  
777 Imaging crustal structure in south central Costa Rica with receiver functions. Geochemistry,

- 778 Geophysics, Geosystems, 11, 1–21.
- 779 Ferriss, E., Plank, T., and Walker, D. (2016) Site-specific hydrogen diffusion rates during  
780 clinopyroxene dehydration. *Contributions to Mineralogy and Petrology*, 171, 1–24.
- 781 Gazel, E., Hayes, J.L., Hoernle, K., Kelemen, P., Everson, E., Holbrook, W.S., Hauff, F., van den  
782 Bogaard, P., Vance, E.A., Chu, S., and others (2015) Continental crust generated in oceanic  
783 arcs. *Nature Geoscience*, 8, 321–327.
- 784 Global Volcanism Program (2013) Irazu (345060). In E. Venzke, Ed., *Volcanoes of the World*.  
785 Smithsonian Institution. <http://volcano.si.edu/volcano.cfm?vn=345060>
- 786 Hartley, M.E., Morgan, D.J., MacLennan, J., Edmonds, M., and Thordarson, T. (2016) Tracking  
787 timescales of short-term precursors to large basaltic fissure eruptions through Fe–Mg  
788 diffusion in olivine. *Earth and Planetary Science Letters*, 439, 58–70.
- 789 Hier-Majumder, S., Anderson, I.M., and Kohlstedt, D.L. (2005) Influence of protons on Fe-Mg  
790 interdiffusion in olivine. *Journal of Geophysical Research B: Solid Earth*, 110, 1–12.
- 791 Ito, M., and Ganguly, J. (2006) Diffusion kinetics of Cr in olivine and  $^{53}\text{Mn}$ – $^{53}\text{Cr}$   
792 thermochronology of early solar system objects. *Geochimica et Cosmochimica Acta*, 70, 799–  
793 809.
- 794 Jambon, A., Lussiez, P., Clocchiatti, R., Weisz, J., and Hernandez, J. (1992) Olivine growth rates  
795 in a tholeiitic basalt: An experimental study of melt inclusions in plagioclase. *Chemical*  
796 *Geology*, 96, 277–287.
- 797 Jarosewich, E., Nelen, J.A., and Norberg, J.A. (1980) Reference Samples for Electron Microprobe  
798 Analysis. *Geostandards Newsletter*, 4, 43–47.
- 799 Jollands, M., Hermann, J., O'Neill, H.S.C., Spandler, C., and Padrón-Navarta, J.A. (2016a)  
800 Diffusion of Ti and some Divalent Cations in Olivine as a Function of Temperature, Oxygen  
801 Fugacity, Chemical Potentials and Crystal Orientation. *Journal of Petrology*, 57, 1983–2010.

- 802 Jollands, M.C., O'Neill, H.S.C., and Hermann, J. (2014) The importance of defining chemical  
803 potentials, substitution mechanisms and solubility in trace element diffusion studies: the case  
804 of Zr and Hf in olivine. *Contributions to Mineralogy and Petrology*, 168, 1055.
- 805 Jollands, M.C., Burnham, A.D., O'Neill, H.S.C., Hermann, J., and Qian, Q. (2016b) Beryllium  
806 diffusion in olivine: A new tool to investigate timescales of magmatic processes. *Earth and  
807 Planetary Science Letters*, 450, 71–82.
- 808 Jollands, M.C., O'Neill, H.S.C., Van Orman, J., Berry, A.J., Hermann, J., Newville, M., and  
809 Lanzirotti, A. (2018) Substitution and diffusion of Cr<sup>2+</sup> and Cr<sup>3+</sup> in synthetic forsterite and  
810 natural olivine at 1200–1500 °C and 1 bar. *Geochimica et Cosmochimica Acta*, 220, 407–428.
- 811 Kahl, M., Chakraborty, S., Costa, F., and Pompilio, M. (2011) Dynamic plumbing system beneath  
812 volcanoes revealed by kinetic modeling, and the connection to monitoring data: An example  
813 from Mt. Etna. *Earth and Planetary Science Letters*, 308, 11–22.
- 814 Kahl, M., Chakraborty, S., Costa, F., Pompilio, M., Liuzzo, M., and Viccaro, M. (2013)  
815 Compositionally zoned crystals and real-time degassing data reveal changes in magma  
816 transfer dynamics during the 2006 summit eruptive episodes of Mt. Etna. *Bulletin of  
817 Volcanology*, 75, 1–14.
- 818 Kahl, M., Chakraborty, S., Pompilio, M., and Costa, F. (2015) Constraints on the Nature and  
819 Evolution of the Magma Plumbing System of Mt. Etna Volcano (1991–2008) from a  
820 Combined Thermodynamic and Kinetic Modelling of the Compositional Record of Minerals.  
821 *Journal of Petrology*, 56, 2025–2068.
- 822 Lazarov, M., and Horn, I. (2015) Matrix and energy effects during in-situ determination of Cu  
823 isotope ratios by ultraviolet-femtosecond laser ablation multicollector inductively coupled  
824 plasma mass spectrometry. *Spectrochimica Acta Part B: Atomic Spectroscopy*, 111, 64–73.
- 825 Lizarralde, D., Holbrook, W.S., Van Avendonk, H.J., Mora Fernandez, M., Alvarado, G.E., and

- 826 Harder, S.H. (2010) Crustal structure along the active Costa Rican volcanic arc. In AGU Fall  
827 Meeting Abstracts p. T13A–2176.
- 828 Longpré, M.-A., Klügel, A., Diehl, A., and Stix, J. (2014) Mixing in mantle magma reservoirs prior  
829 to and during the 2011-2012 eruption at El Hierro, Canary Islands. *Geology*, 42, 315–318.
- 830 Maaløe, S. (2011) Olivine phenocryst growth in Hawaiian tholeiites: Evidence for supercooling.  
831 *Journal of Petrology*, 52, 1579–1589.
- 832 Mallmann, G., O'Neill, H.S.C., and Klemme, S. (2009) Heterogeneous distribution of phosphorus  
833 in olivine from otherwise well-equilibrated spinel peridotite xenoliths and its implications for  
834 the mantle geochemistry of lithium. *Contributions to Mineralogy and Petrology*, 158, 485–  
835 504.
- 836 Milman-Barris, M.S., Beckett, J.R., Baker, M.B., Hofmann, A.E., Morgan, Z., Crowley, M.R.,  
837 Vielzeuf, D., and Stolper, E. (2008) Zoning of phosphorus in igneous olivine. *Contributions to*  
838 *Mineralogy and Petrology*, 155, 739–765.
- 839 Oeser, M., Weyer, S., Horn, I., and Schuth, S. (2014) High-Precision Fe and Mg Isotope Ratios of  
840 Silicate Reference Glasses Determined In Situ by Femtosecond LA-MC-ICP-MS and by  
841 Solution Nebulisation MC-ICP-MS. *Geostandards and Geoanalytical Research*, 38, 311–328.
- 842 Oeser, M., Dohmen, R., Horn, I., Schuth, S., and Weyer, S. (2015) Processes and time scales of  
843 magmatic evolution as revealed by Fe–Mg chemical and isotopic zoning in natural olivines.  
844 *Geochimica et Cosmochimica Acta*, 154, 130–150.
- 845 Otamendi, J.E., Tiepolo, M., Walker, B.A., Cristofolini, E.A., and Tibaldi, A.M. (2016) Trace  
846 elements in minerals from mafic and ultramafic cumulates of the central Sierra de Valle Fértil,  
847 Famatinian arc, Argentina. *Lithos*, 240–243, 355–370.
- 848 Papike, J.J., Karner, J.M., and Shearer, C.K. (2005) Comparative planetary mineralogy: Valence  
849 state partitioning of Cr, Fe, Ti, and V among crystallographic sites in olivine, pyroxene, and

- 850 spinel from planetary basalts. *American Mineralogist*, 90, 277–290.
- 851 Qian, Q., O'Neill, H.S.C., and Hermann, J. (2010) Comparative diffusion coefficients of major and  
852 trace elements in olivine at 950 °C from a xenocryst included in dioritic magma. *Geology*, 38,  
853 331–334.
- 854 Rae, A.S.P., Edmonds, M., MacLennan, J., Morgan, D., Houghton, B., Hartley, M.E., and Sides, I.  
855 (2016) Time scales of magma transport and mixing at Kilauea Volcano, Hawai'i. *Geology*,  
856 44, 463–466.
- 857 Richter, F., Chaussidon, M., Mendybaev, R., and Kite, E. (2016) Reassessing the cooling rate and  
858 geologic setting of Martian meteorites MIL 03346 and NWA 817. *Geochimica et*  
859 *Cosmochimica Acta*, 182, 1–23.
- 860 Richter, F.M., Liang, Y., and Davis, A.M. (1999) Isotope fractionation by diffusion in molten  
861 oxides. *Geochimica et Cosmochimica Acta*, 63, 2853–2861.
- 862 Richter, F.M., Davis, A.M., DePaolo, D.J., and Watson, E.B. (2003) Isotope fractionation by  
863 chemical diffusion between molten basalt and rhyolite. *Geochimica et Cosmochimica Acta*,  
864 67, 3905–3923.
- 865 Rubin, A.M. (1995) Propagation of Magma-Filled Cracks. *Annual Review of Earth and Planetary*  
866 *Sciences*, 23, 287–336.
- 867 Ruprecht, P., and Plank, T. (2013) Feeding andesitic eruptions with a high-speed connection from  
868 the mantle. *Nature*, 500, 68–72.
- 869 Ruprecht, P., Bergantz, G.W., and Dufek, J. (2008) Modeling of gas-driven magmatic overturn:  
870 Tracking of phenocryst dispersal and gathering during magma mixing. *Geochemistry*  
871 *Geophysics Geosystems*, 9, 1–20.
- 872 Schleicher, J.M., and Bergantz, G.W. (2017) The Mechanics and Temporal Evolution of an Open-  
873 system Magmatic Intrusion into a Crystal-rich Magma. *Journal of Petrology*, 58, 1059–1072.

- 874 Schwab, R.G., and Küstner, D. (1981) The equilibrium fugacities of important oxygen buffers in  
875 technology and petrology. *Neues Jahrbuch für Mineralogie-Abhandlungen*, 140, 111–142.
- 876 Shea, T., Lynn, K.J., and Garcia, M.O. (2015) Cracking the olivine zoning code: Distinguishing  
877 between crystal growth and diffusion. *Geology*, 43, 935–938.
- 878 Sio, C.K.I., and Dauphas, N. (2017) Thermal and crystallization histories of magmatic bodies by  
879 Monte Carlo inversion of Mg-Fe isotopic profiles in olivine. *Geology*, 44, G38056.1.
- 880 Sio, C.K.I., Dauphas, N., Teng, F.-Z., Chaussidon, M., Helz, R.T., and Roskosz, M. (2013)  
881 Discerning crystal growth from diffusion profiles in zoned olivine by in situ Mg–Fe isotopic  
882 analyses. *Geochimica et Cosmochimica Acta*, 123, 302–321.
- 883 Spandler, C., and O’Neill, H.S.C. (2010) Diffusion and partition coefficients of minor and trace  
884 elements in San Carlos olivine at 1,300°C with some geochemical implications. *Contributions  
885 to Mineralogy and Petrology*, 159, 791–818.
- 886 Spandler, C., O’Neill, H.S.C., and Kamenetsky, V.S. (2007) Survival times of anomalous melt  
887 inclusions from element diffusion in olivine and chromite. *Nature*, 447, 303–6.
- 888 Teng, F.-Z., Dauphas, N., Helz, R.T., Gao, S., and Huang, S. (2011) Diffusion-driven magnesium  
889 and iron isotope fractionation in Hawaiian olivine. *Earth and Planetary Science Letters*, 308,  
890 317–324.
- 891 Tollan, P.M.E., O’Neill, H.S.C., Hermann, J., Benedictus, A., and Arculus, R.J. (2015) Frozen  
892 melt-rock reaction in a peridotite xenolith from sub-arc mantle recorded by diffusion of trace  
893 elements and water in olivine. *Earth and Planetary Science Letters*, 422, 169–181.
- 894 Van Orman, J.A., and Krawczynski, M.J. (2015) Theoretical constraints on the isotope effect for  
895 diffusion in minerals. *Geochimica et Cosmochimica Acta*, 164, 365–381.
- 896 Wang, Z., Hiraga, T., and Kohlstedt, D.L. (2004) Effect of H<sup>+</sup> on Fe-Mg interdiffusion in olivine,  
897 (Fe,Mg)<sub>2</sub>SiO<sub>4</sub>. *Applied Physics Letters*, 85, 209–211.

898 Watson, E.B., Cherniak, D.J., and Holycross, M.E. (2015) Diffusion of phosphorus in olivine and  
899 molten basalt. *American Mineralogist*, 100, 2053–2065.

900 White, R.W. (1966) Ultramafic inclusions in basaltic rocks from Hawaii. *Contributions to*  
901 *Mineralogy and Petrology*, 12, 245–314.

902 Zhukova, I., O’Neill, H.S.C., Cambell, I.H., and Kilburn, M.R. (2014) The effect of silica activity  
903 on the diffusion of Ni and Co in olivine. *Contributions to Mineralogy and Petrology*, 168,  
904 1029.

905 Zhukova, I., O’Neill, H., and Campbell, I.H. (2017) A subsidiary fast-diffusing substitution  
906 mechanism of Al in forsterite investigated using diffusion experiments under controlled  
907 thermodynamic conditions. *Contributions to Mineralogy and Petrology*, 172, 1–12.

908

909

910

911

912

### Figure Captions

913

914 **Figure 1.** Schematic architecture of Irazú magma system (modified from Ruprecht and Plank  
915 2013) and the origin of distinct olivine types A-C. Green represents primitive magnesian olivines  
916 ( $\text{Fo}_{89-91}$ ), or the Mg-rich rims ( $\text{Fo}_{87-90}$ ) grown around crystal cores with intermediate Fo contents  
917 ( $\text{Fo}_{80-87}$ ; blue) when the latter were picked up by ascending mantle melts. Orange represents  
918 normally zoned ( $\text{Fo}_{87-90}$  to  $\text{Fo}_{79-84}$ ) outer rims of olivine crystals or intermediate to low Fo olivines  
919 from the shallow magma storage region.

920

921 **Figure 2.** BSE and reflected-light images of two olivine crystals investigated in this study. **(a)** and  
922 **(c)** BSE images of crystals IZ-10-12 ol16 and IZ-10-12 ol11, respectively, before in-situ analyses  
923 of Fe-Mg isotopic profiles by LA-MC-ICP-MS. **(b)** and **(d)** Reflected-light images of the same  
924 crystals after the in-situ analyses by LA-MC-ICP-MS.

925

926 **Figure 3.** Examples of type A **(left)** and type B **(right)** olivine crystals from Irazú. Type A olivines  
927 show reverse Ni zoning at constant and high Fo and Cr contents in their cores, while type B  
928 olivines display reverse zoning of Fo and Cr (and in some cases also of Al and V) as well as very  
929 low Cr concentrations in their cores..

930

931 **Figure 4.** Ni/Cr versus Fo content in olivines investigated in this study. Type B olivines are  
932 characterized by large Ni/Cr ratios, while type A olivines are constrained to low Ni/Cr ratios at  
933 high Fo content. All core-to-rim data for Type B crystals are labeled as xenocrysts and therefore  
934 xenocryst rims plot at low Ni/Cr ratios.

935

936 **Figure 5.** Fe-Mg chemical (expressed as Fo) and isotopic profiles for four complexly zoned  
937 olivines analyzed in this study. All olivines show strong coupling of Fe-Mg chemical and isotopic  
938 zoning (especially for Fe isotopes), indicating equilibration by Fe-Mg inter-diffusion. Sample  
939 material of these four olivine crystals was ablated by line scans parallel to the chemical zoning.  
940 Error bars represent the analytical uncertainty (2 SD) during one analytical session.

941

942 **Figure 6.** Fe-Mg chemical (expressed as Fo) and isotopic profiles for three primitive, normally  
943 zoned olivines investigated in this study. The normal zoning at the rim is strongly coupled with  
944 inversely correlated Fe-Mg isotopic profiles, implying diffusive re-equilibration prior to eruption.



945 Sample material of olivine IZ-10-13 ol22 was ablated by line scans perpendicular to the chemical  
946 zoning. Error bars represent the analytical uncertainty (2 SD) during one analytical session.

947

948 **Figure 7.**  $\delta^{56}\text{Fe}$  and  $\delta^{26}\text{Mg}$  values for all Fe-Mg isotopic profiles acquired in this study, illustrating  
949 that Fe isotope variations (up to 1.7‰) are much more pronounced than Mg isotope variations  
950 (<0.6‰) in all investigated olivine grains.

951

952 **Figure 8.** Results of the stepwise diffusion modeling approach used in this study for a complexly  
953 zoned olivine (IZ-10-12 ol16). Profile of Fo contents is based on the calibrated BSE (grayscale)  
954 image of this crystal. **(left)** Subfigures show the measured and modeled Fo contents **(a)** and the  
955 measured and modeled Fe isotopic compositions **(b)** for diffusion episode 1, assuming a model of  
956 instantaneous growth of a Mg-rich layer (distance 0-106  $\mu\text{m}$  in a) and subsequent Fe-Mg inter-  
957 diffusion. For diffusion episode 2 **(right)**, the diffusion profiles generated during the first diffusion  
958 episode are used as the initial concentration (and isotopic) gradients, and only the rim composition  
959 (at distance = 0  $\mu\text{m}$ ) is modified to simulate diffusion of Fe into (and Mg out of) the olivine.

960

961 **Figure 9.** Results of the stepwise diffusion modeling approach used in this study for a complexly  
962 zoned olivine (IZ-10-13 ol34), assuming a model of instantaneous growth of a Mg-rich layer  
963 (distance 0-156  $\mu\text{m}$  in a) and subsequent Fe-Mg inter-diffusion for diffusion episode 1 **(left)**. This  
964 example also illustrates that the second diffusion episode **(right)** with diffusion of Fe into (and Mg  
965 out of) the olivine is needed to reproduce the low Fo content **(c)** and relative enrichment of light Fe  
966 isotopes **(d)** near the rim. Note that the profile of Fo contents is shown as measured by EMPA. For  
967 further details, see figure caption of Fig. 8.

968

969 **Figure 10.** Comparison of modeling results obtained from two models for diffusion episode 1  
970 (instantaneous growth and subsequent diffusion [**left**] vs. growth and simultaneous diffusion  
971 [**right**]), both of which are followed by simple diffusive Fe-Mg exchange between crystal and melt  
972 (diffusion episode 2, lasting 448 days in this case). Both models for diffusion episode 1 are able to  
973 reproduce the Fe-Mg chemical (**a, c**) and Fe isotopic zoning (**b, d**) of this olivine reasonably well.

974

975 **Figure 11.** Comparison of modeling results obtained from two models for diffusion episode 1  
976 (instantaneous growth and subsequent diffusion [**left**] vs. growth and simultaneous diffusion  
977 [**right**]), both of which are followed by simple diffusive Fe-Mg exchange between crystal and melt  
978 (diffusion episode 2, lasting 11 days in this case). Both models for diffusion episode 1 are able to  
979 reproduce the Fe-Mg chemical zoning of this olivine (**a, c**); however, the second model yields a  
980 much better fit to the Fe isotopic zoning of this crystal (**d**), indicating that the shorter diffusion  
981 timescale (110+11 days) is more likely in this case.

982

983 **Figure 12.** Schematic architecture of Irazú magma system (modified from Ruprecht and Plank  
984 2013) with information about major and trace element characteristics of the three main olivine  
985 types and about timescales of transport ( $\Delta t$ ) as deduced in this study from diffusion modeling at the  
986 given temperature range (1080-1150°C). Transport timescale for primitive magnesian olivines  
987 (green) from MOHO depth is from Ruprecht and Plank (2013), based on diffusion modeling of  
988 reverse Ni zoning.

989

990 **Figure 13.** Comparison of Cr, Al, and V diffusivity relative to Fe-Mg inter-diffusion. Filled and  
991 open squares are results for smearing corrected and uncorrected profiles, respectively. Al data is  
992 limited to sample IZ-10-12, which was measured on a day with better instrument sensitivity. The

993 results for Cr and V diffusivity determined by Spandler and O'Neill (2010; SPO'N10) are shown  
994 for comparison, whereas Al diffusion was not quantified in that study. Experiments on Cr  
995 diffusion by Jollands et al. (2018) did not explore co-existing Fe-Mg inter-diffusion and therefore  
996 no diffusivity ratios are shown for this study. The results from our study are not plotted for specific  
997 crystallographic axis, because of random orientations of the diffusion profiles; see supplementary  
998 data (Table S3) for rotation angles. Diffusivities were determined by 1D diffusion profiles (see  
999 Figure 14).

1000

1001 **Figure 14.** LA-ICPMS rim-to-center analyses for Fo and Al (white squares) as well as Cr and V  
1002 (gray squares). Calculated diffusion profiles used for the relative determination of elemental  
1003 diffusivities are shown as black (Fo, Al) and gray (Cr, V) lines. Profiles were calculated separately  
1004 for each element with a step profile. Step location was chosen to minimize misfit for the respective  
1005 element. Near-rim zoning is ignored as it probably includes effects of fractional crystallization.  
1006 Additional Fo, V, and Cr zoning profiles are shown in the supplementary material (Fig. S8).

1007

1008

**Figure 1**

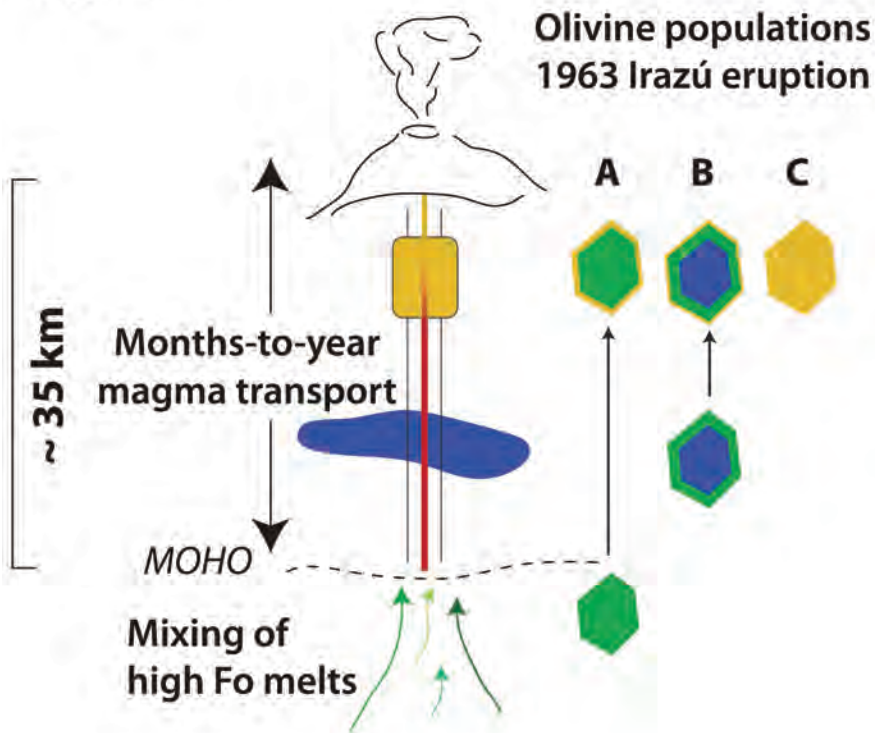
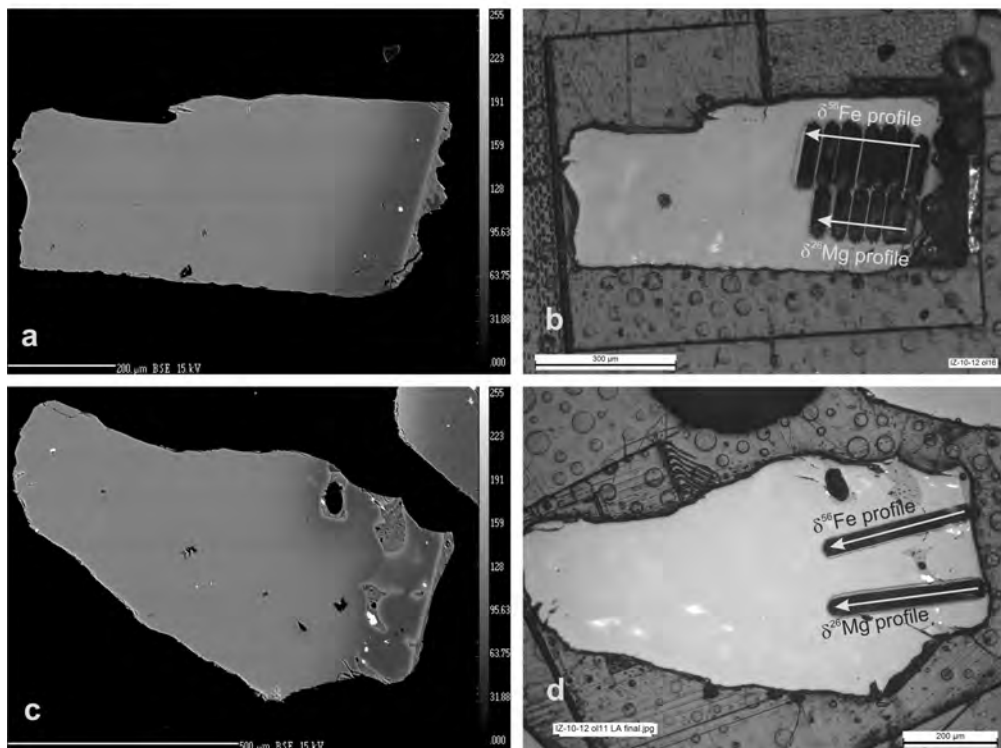
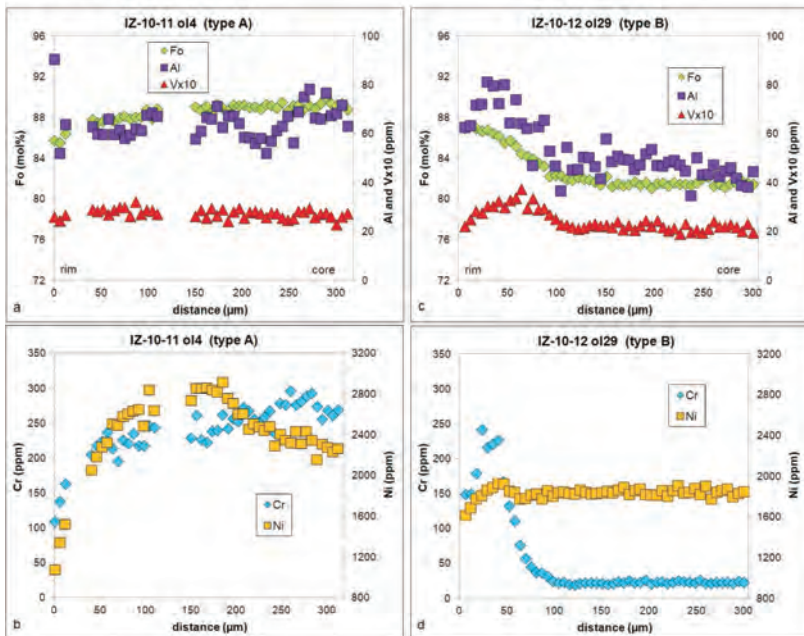


Figure 2



### Figure 3



**Figure 4**

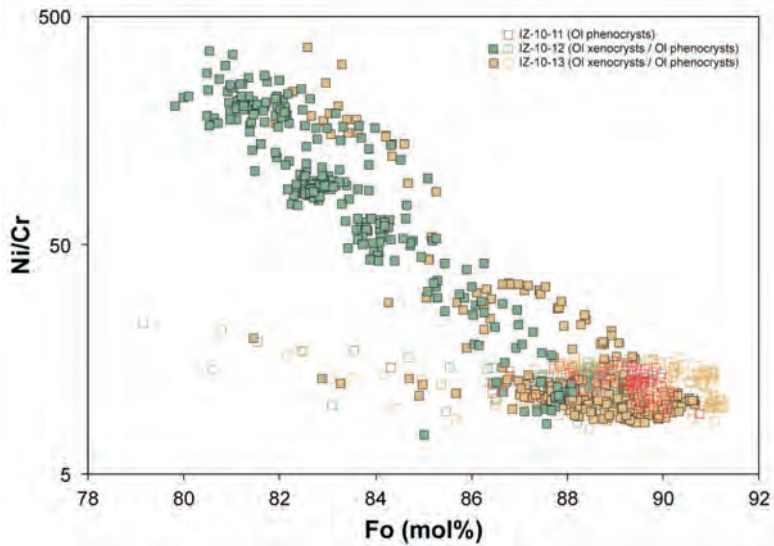
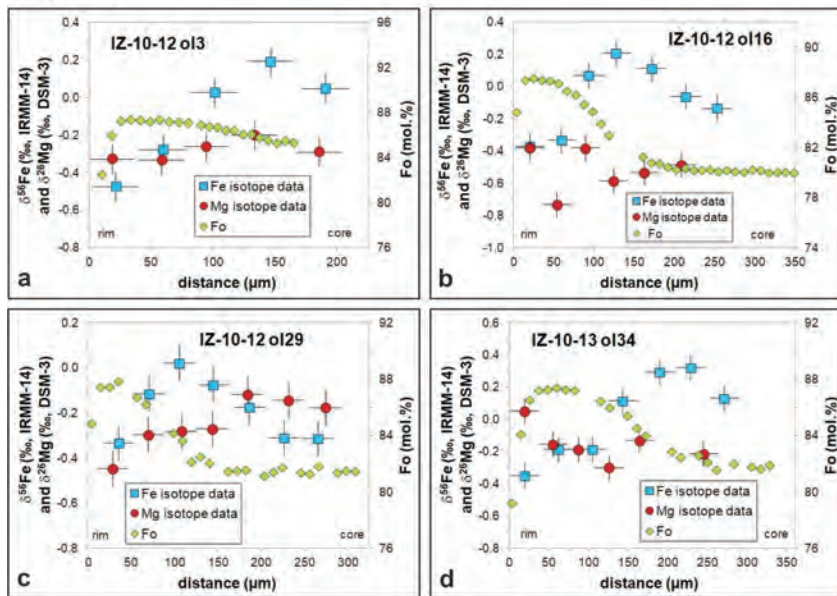


Figure 5





# Figure 6

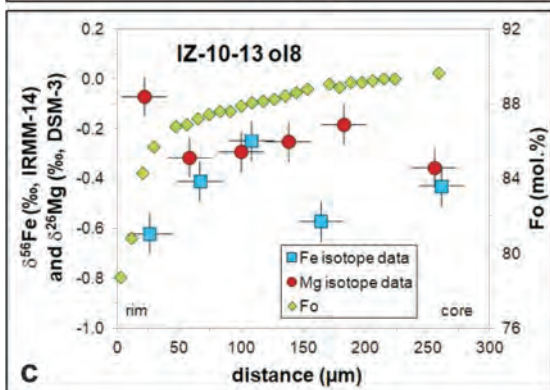
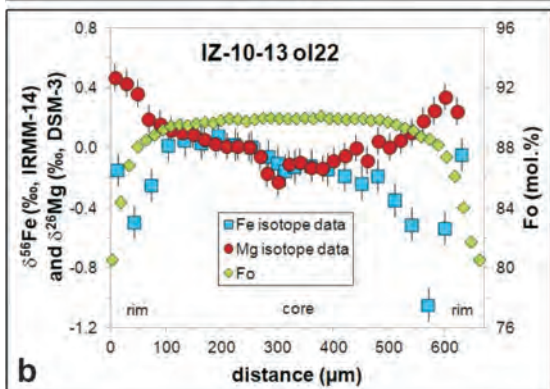
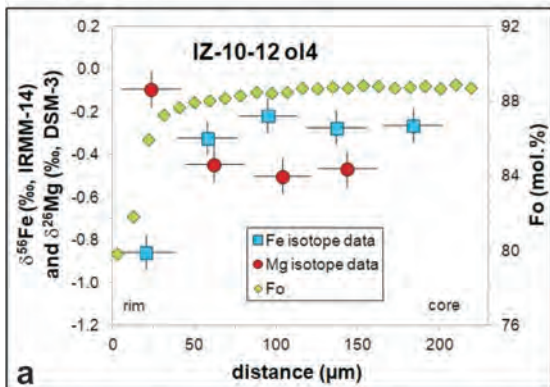


Figure 7

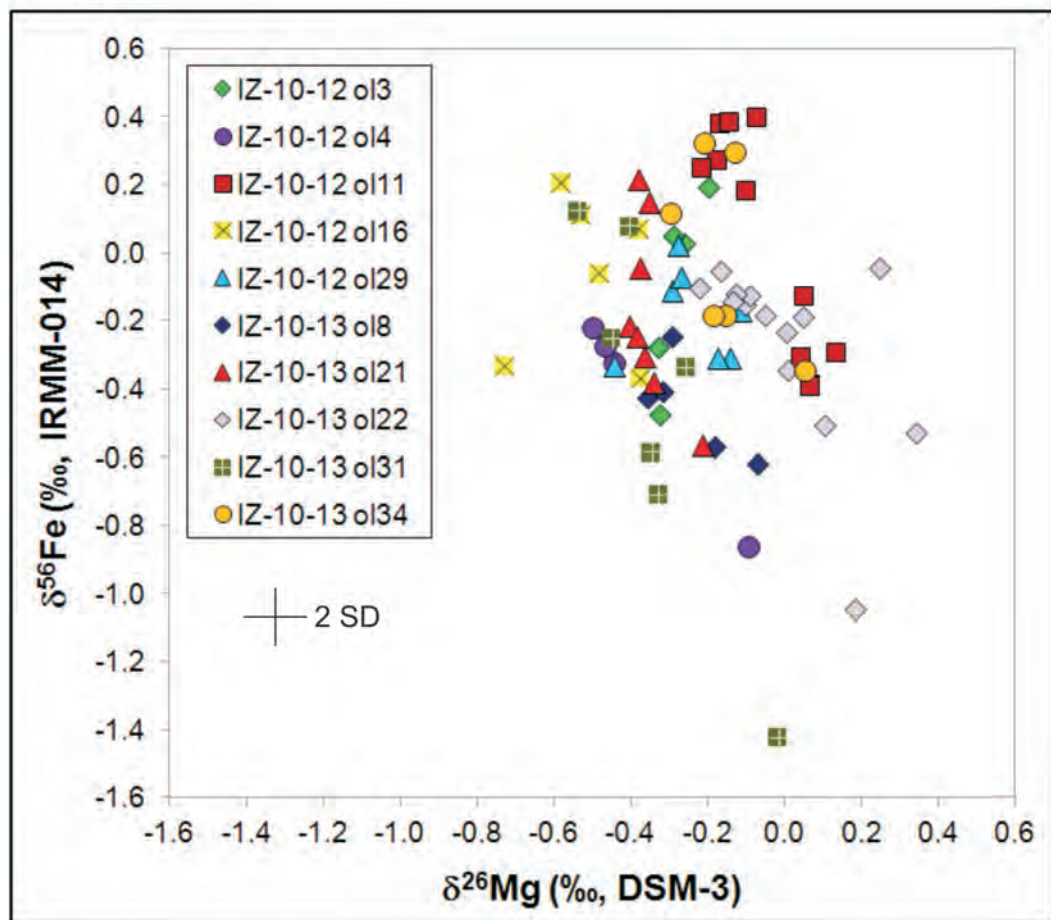


Figure 8

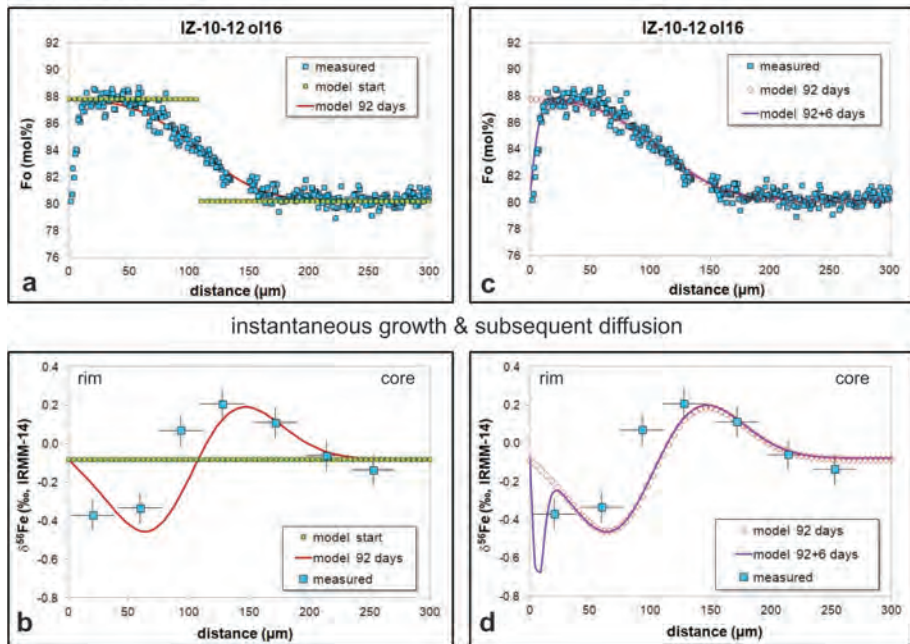


Figure 9

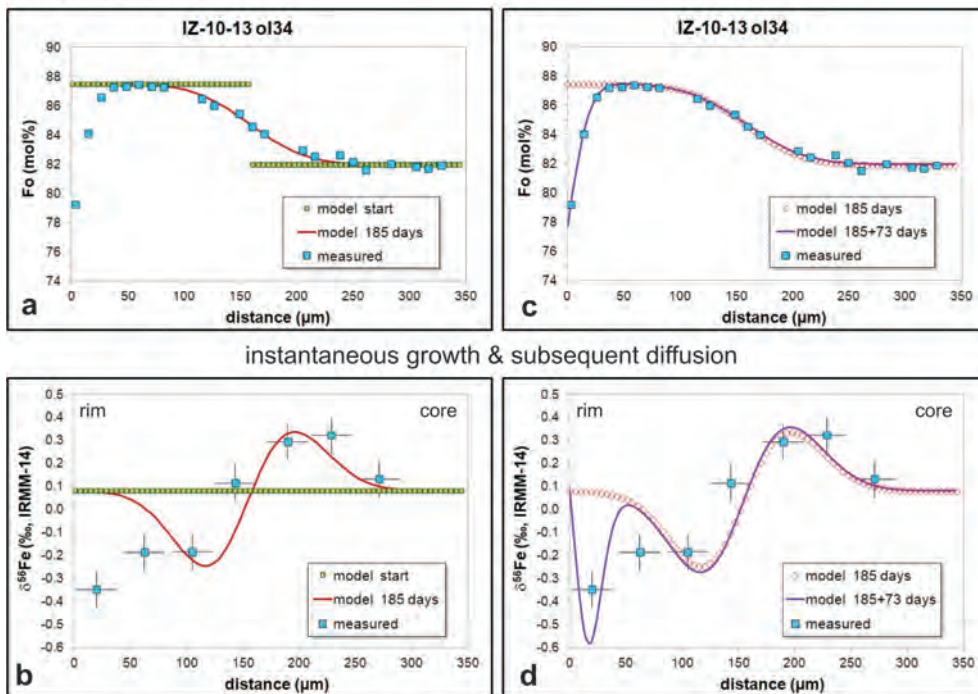


Figure 10

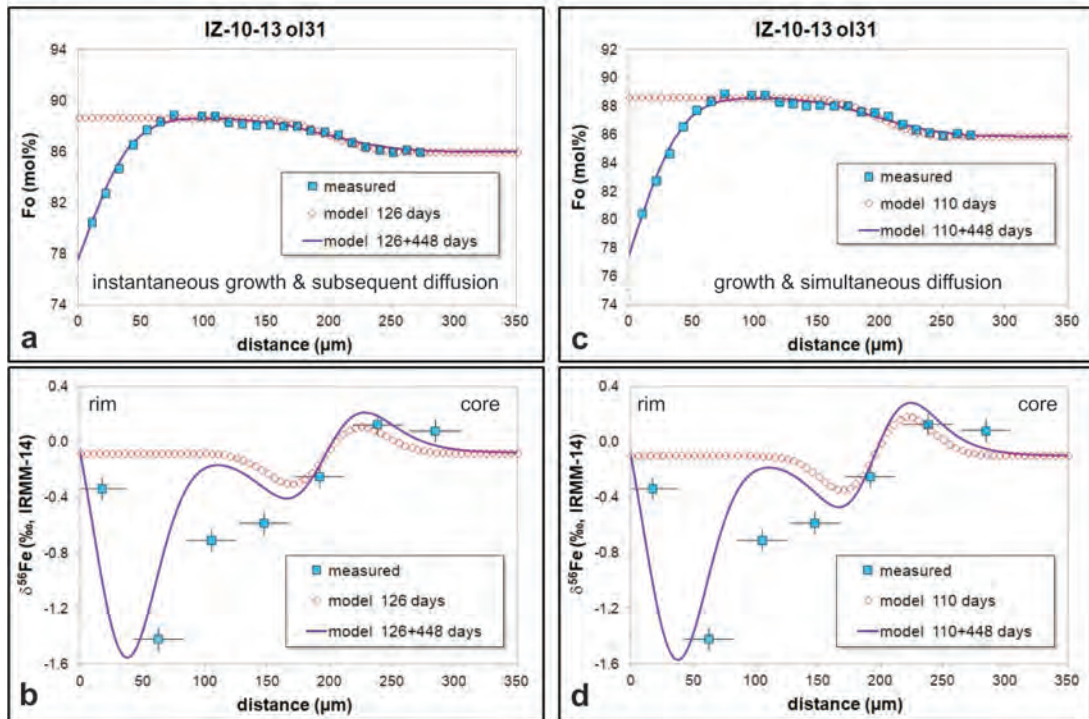


Figure 11

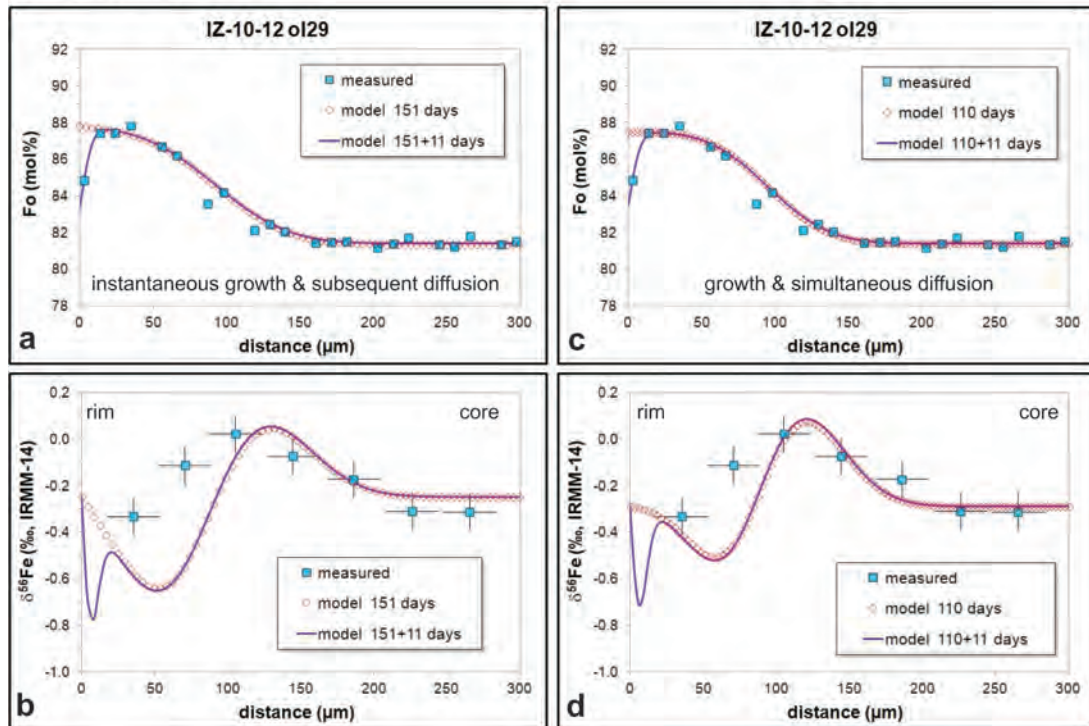




Figure 12

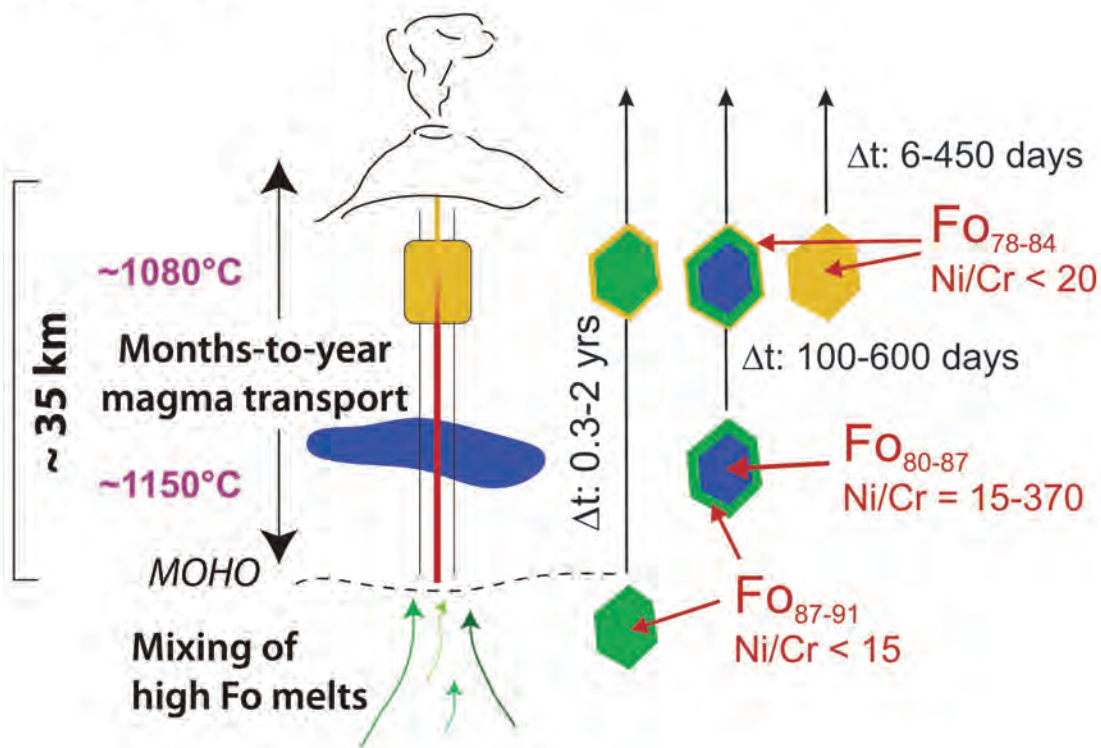
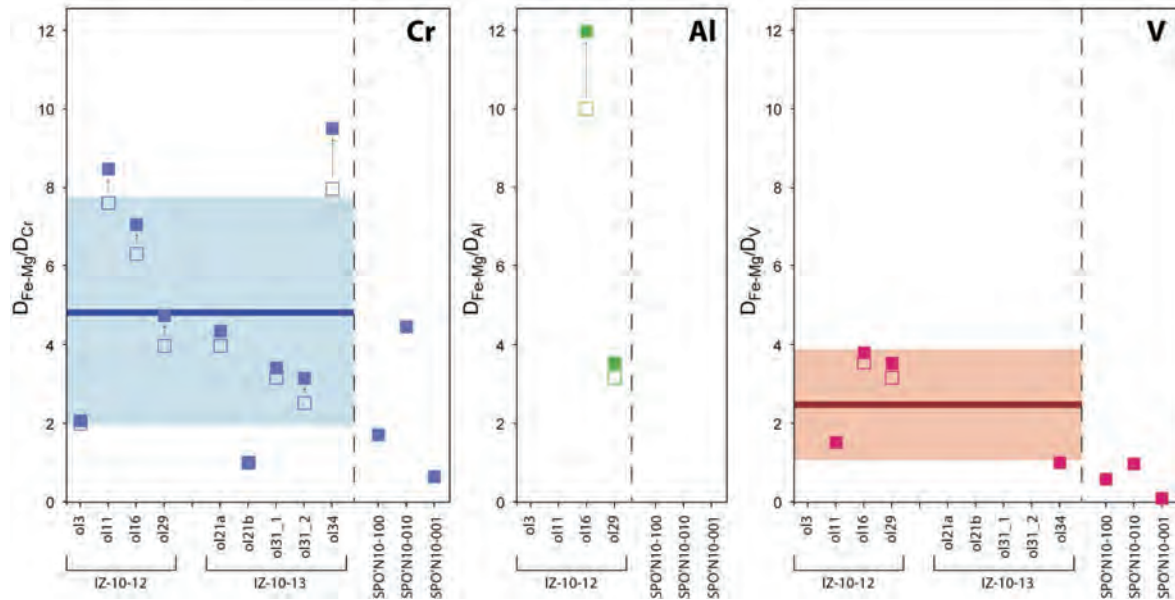
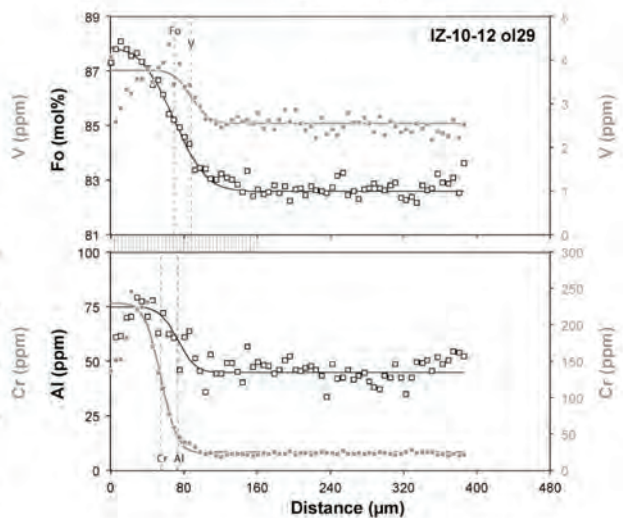
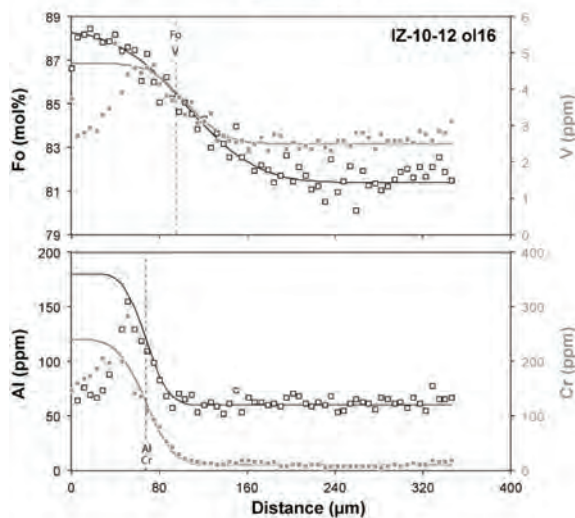
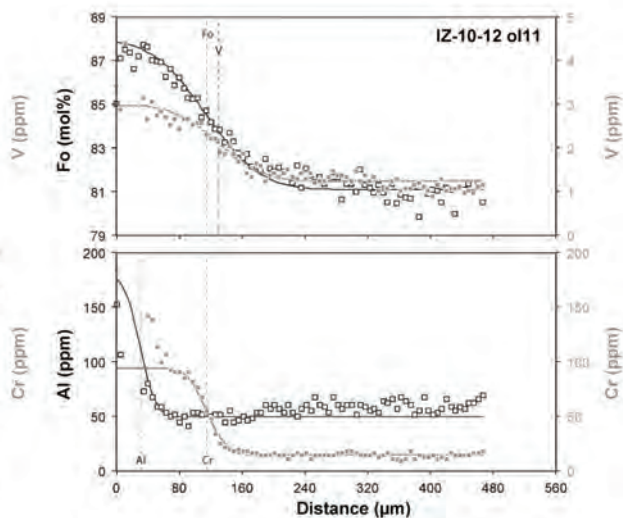
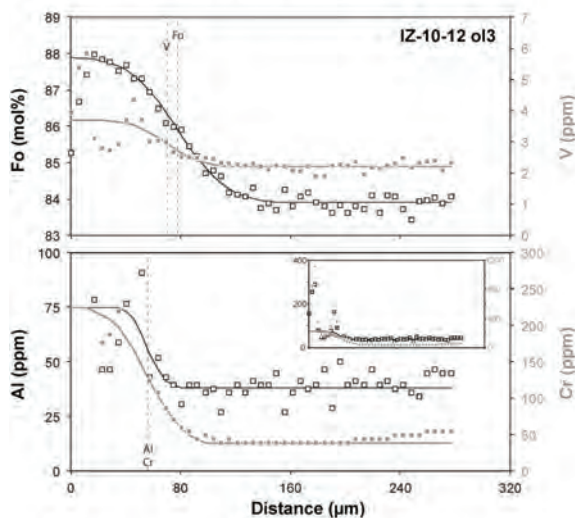


Figure 13





# Figure 14



**Table 1: Timescales of diffusive re-equilibration for complexly and normally zoned olivine**

sample	type of Fe-Mg zoning	t1 (days) <sup>a</sup>	t2 (days) <sup>b</sup>	t <sub>total</sub> (days) <sup>c</sup>
IZ-10-12 ol3	complex	168	48	216
IZ-10-12 ol11	complex	533 <sup>d</sup>	56 <sup>d</sup>	589
IZ-10-12 ol16	complex	92	6	98
IZ-10-12 ol29	complex	151	11	162
IZ-10-13 ol21	complex	294	1456	1750
IZ-10-13 ol31	complex	126	448	574
IZ-10-13 ol34	complex	185	73	258
IZ-10-12 ol4	normal	x <sup>e</sup>	168	168
IZ-10-13 ol8	(multiple) normal	x	112	112
IZ-10-13 ol22	normal	x	224	224

<sup>a</sup> Model parameters for diffusion episode 1:  $T = 1150^{\circ}\text{C}$ ,  $f\text{O}_2$  at NNO+1,  $X_{\text{Fe}} = 0.15$ ,  $P = 10^5$  bar

<sup>b</sup> Model parameters for diffusion episode 2:  $T = 1080^{\circ}\text{C}$ ,  $f\text{O}_2$  at NNO+1,  $X_{\text{Fe}} = 0.15$ ,  $P = 10^5$  bar

<sup>c</sup> Given the uncertainties of the temperature estimation ( $\pm 50^{\circ}\text{C}$ ), of the assumed  $f\text{O}_2$ , and of the assumed  $X_{\text{Fe}}$

<sup>d</sup> Timescales obtained by fitting the width of the isotopic diffusion profile, as the Fe-Mg cation diffusion

<sup>e</sup> x = not applicable.

**y zoned Irazú olivines, using a model of instantaneous growth and subsequent diffu**

$\beta_{\text{Fe}}$	$\cos(\alpha)^2$	$\cos(\beta)^2$	$\cos(\gamma)^2$
0.205	0.952	0.009	0.038
0.220	0.900	0.086	0.014
0.150	0.324	0.140	0.535
0.180	0.045	0.755	0.200
0.250	0.261	0.733	0.006
0.220	0.638	0.313	0.050
0.160	0.576	0.281	0.142
0.120	0.514	0.436	0.050
0.150	0.735	0.215	0.051
0.140	0.016	0.012	0.972

= 500 MPa.

= 100 MPa.

d of the experimental calibration of  $D_{\text{Fe-Mg}}$ , the uncertainty of the modeled timescales is within a factor of 2. If the chemical diffusion profile appears to be distorted due to the polishing of the crystal before the EMPA

tion for diffusion episode 1 (t1) if applicable.

factor of 4.

u.

**Table 2: Timescales of diffusive re-equilibration and integrated growth rate**

sample	type of Fe-Mg zoning	t1 (days) <sup>a</sup>	t2 (days) <sup>b</sup>	t <sub>total</sub> (days)
IZ-10-12 ol3	complex	128	48	176
IZ-10-12 ol11	complex	530 <sup>d</sup>	56 <sup>d</sup>	586
IZ-10-12 ol16	complex	90	6	96
IZ-10-12 ol29	complex	110	11	121
IZ-10-13 ol21	complex	290	1456	1746
IZ-10-13 ol31	complex	110	448	558
IZ-10-13 ol34	complex	180	73	253

<sup>a</sup> Model parameters for diffusion episode 1:  $T = 1150^{\circ}\text{C}$ ,  $f\text{O}_2$  at NNO+1,  $X_{\text{Fe}} = 0.15$ ,  $P$

<sup>b</sup> Model parameters for diffusion episode 2:  $T = 1080^{\circ}\text{C}$ ,  $f\text{O}_2$  at NNO+1,  $X_{\text{Fe}} = 0.15$ ,  $P$

<sup>c</sup> Rate of growth ( $R$ ) of the Mg-rich layer during diffusion episode 1.

<sup>d</sup> Timescales obtained by fitting the width of the isotopic diffusion profile, as the Fe-Mg

<sup>e</sup> Peclet number ( $R \cdot \Delta x / D$ ), where  $R$  is the growth rate,  $\Delta x$  is the zoning lengthscale, and

es for complexly zoned Irazú olivines, using a model of growth and simultaneous

growth rate (m/s) <sup>c</sup>	Pe <sup>e</sup>
1.0 x 10 <sup>-11</sup>	21
3.0 x 10 <sup>-12</sup>	13
1.3 x 10 <sup>-11</sup>	16
9.5 x 10 <sup>-12</sup>	12
1.0 x 10 <sup>-11</sup>	38
2.1 x 10 <sup>-11</sup>	46
9.9 x 10 <sup>-12</sup>	22

= 500 MPa.

= 100 MPa.

chemical diffusion profile appears to be distorted due to the polishing of the crystal before the  
and  $D$  is the diffusion coefficient, for diffusion episode 1.

**ous diffusion for diffusion episode 1 (t1).**

e EMPA.