

Phosphovanadylite: A new vanadium phosphate mineral with a zeolite-type structure

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ABSTRACT

Phosphovanadylite, whose simplified formula is $(\text{Ba,Ca,K,Na})_x[(\text{V,Al})_4\text{P}_2(\text{O,OH})_{16}] \cdot 12\text{H}_2\text{O}$, is a new vanadium phosphate zeolite mineral found in the Phosphoria Formation at Monsanto's Enoch Valley Mine, Soda Springs, Idaho. Its formula in more detail is $(\text{Ba}_{0.38}\text{Ca}_{0.20}\text{K}_{0.06}\text{Na}_{0.02})_{\Sigma 0.66}[\text{P}_2(\text{V}_{3.44}\text{Al}_{0.46})_{\Sigma 3.90}\text{O}_{10.34}(\text{OH})_{5.66}] \cdot 12\text{H}_2\text{O}$. The drusy mineral occurs as pale greenish-blue euhedral cubes (20–50 μm edge) coating phosphatic, organic-rich mudstone. The chemical composition determined by electron microprobe is (in weight percent) V-28.02, P-9.91, Al-1.97, Ca-1.31, Ba-8.28, Cd-0.09, Zn-0.34, Na-0.15, K-0.73, O-46.57, and F-0.03. The index of refraction is $n_D = 1.566$ (4) and specific gravity is 2.16 (3). The X-ray powder pattern shows strong reflections at 3.16 Å (422), 2.58 (600), 2.44 (620), and 7.73 (200), which are indexed on the basis of a cubic body-centered unit cell with $a = 15.470$ (4) Å. From the single-crystal structure analysis, its space group was determined to be $I\bar{4}3m$, $Z = 6$, and its structure consists of V_4O_{16} octahedral clusters linked to each other by P atoms to form a cubic lattice, creating cavities 7.0 and 5.5 Å in diameter where mainly H_2O resides. Final residual indexes are $R = 0.066$, $R_w = 0.061$, goodness-of-fit = 0.75, and 93 observations and 24 parameters.